Paper presentation

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Photoemission Spectroscopy and Atomic Force Microscopy Investigation of Vapor-Phase Codeposited Silver/Poly(3-hexylthiophene) Composites

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Introduction:

- The nanoengineering of hybrid polymer-metal, -metal oxide, -semiconductors polymers and –fullerenes as thin films is a fast developing field of nanotechnology.
- Applications in novel organic optoelectronics such as solid-state electronics, electroluminescent devices and photovoltaics, and photodetectors.
- •The functionality and performance of these devices are largely dependent on the charge-transfer process across the interfacial structure between organic semiconductors (polymer) and nanoparticles.
- At the hetero junction, transfer of electrons from polymer to filler (nanoparticles) will occur.
- The presence of solvent will diminish the effective interaction between filler and polymer and also charge transfer.
- Vapor phase codeposition presents an alternative route to synthesize a metal/polymer blend

In this paper,

- Poly (3-hexyl thiophene) and silver can be simultaneously thermally evaporated to form a nanocomposite, with different metal and polymer content.
- The composite materials were investigated by photoemission spectroscopy (XPS and UPS) and atomic force microscopy (AFM).
- XPS and UPS gives the electronic and chemical structure and AFM gives the morphology of the of the composites.
- Three nominal Ag/P3HT volume ratios were selected including 1:3, 1:1, and 3:1.

Experimental Section:

- The P3HT deposition was kept at a constant low deposition rate (ca. 1 Å/min) and the silver deposition rate was varied to alter the metal loadings.
- The thickness for all samples is estimated to be around 50 nm.





Biswas et al. Appl. Phys. Lett. 88, 013103 2006

Results and Discussion:



FIGURE 1. Tapping mode AFM topological micrographs of Ag/P3HT composites on Si substrates with Ag/P3HT volume ratio of (a) 3:1 and (b) 1:3. (c) High-resolution image displays a small particle size of about 20 nm across, as confirmed by (d) the line profiles.



FIGURE 2. XPS spectra of codeposited Ag/P3HT matrix on Ag foil with different Ag loadings (a) Ag 3d, (b) C 1s, and (c) S 2p. The ratios represent the amount of Ag to P3HT. Spectra of vapor-phase-deposited P3HT films on Ag were also superimposed for comparison



FIGURE 3. XPS spectra of codeposited Ag/P3HT matrix on silicon wafer with different Ag contents (a) Ag 3d, (b) C 1s, and (c) S 2p. The ratios indicate the amount of Ag to P3HT.



FIGURE 4. He I UPS spectra of codeposited Ag/P3HT matrix on Ag foil with three different Ag contents. Vacuum shifts (Δ) and the highest occupied band edge are shown.



FIGURE 5. He I UPS spectra of codeposited Ag/P3HT matrix on Si coupons with two different Ag contents. Vacuum shifts (Δ) and highest occupied band edges are shown.



FIGURE 6. Schematic energy diagram of Ag, P3HT, and Ag/P3HT nanocomposites as a function of Ag content with respect to bulk silver. The energy information for the blend film is from the Ag/P3HT mixture rather than the interface with the underlying substrate.

CONCLUSION

- AFM showed Ag nanoparticles embedded in P3HT matrix
- XPS suggest the two Ag-S species
- UPS: Large secondary cutoff edge shifts (Δ) to higher binding energies were observed with increasing P3HT content.
- The magnitude of HOB edge (barrier height) increases with polymer content which impedes the charge injection from the Fermi energy of Ag into the valence band of the composite materials.
- Three small yet clear peaks were observed in the 5-12 eV region which is related to the σ states of P3HT backbone.
- It is possible to tune the value of the barrier height (ε_v^F) from 0.55 to 1.36 eV by simply varying the composition of the blend film (Ag/P3HT ratios).

