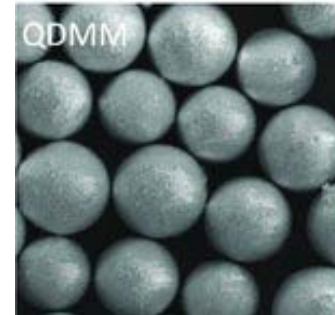
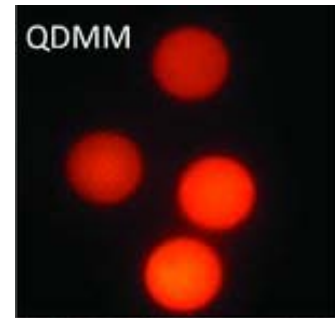
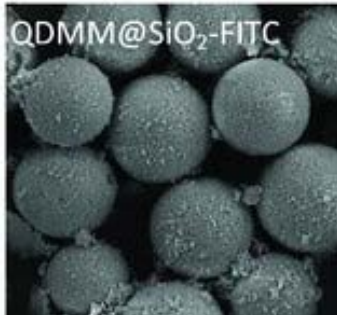
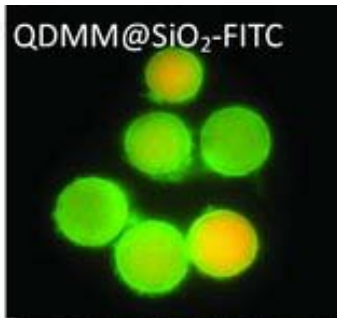


Stable Encapsulation of Quantum Dot Barcodes with Silica Shells

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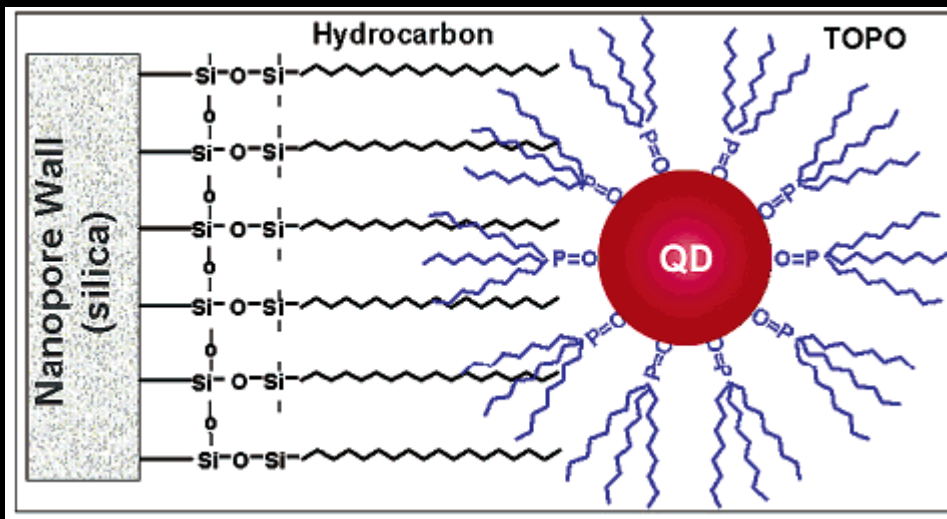


ADVANCED FUNCTIONAL MATERIALS

Adv. Funct. Mater. 2010. ASAP

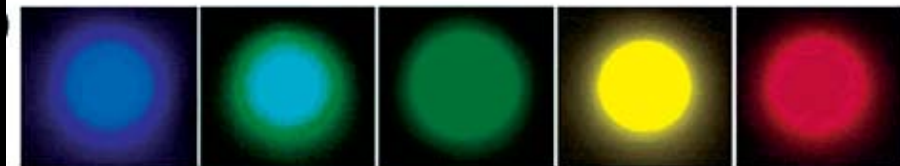
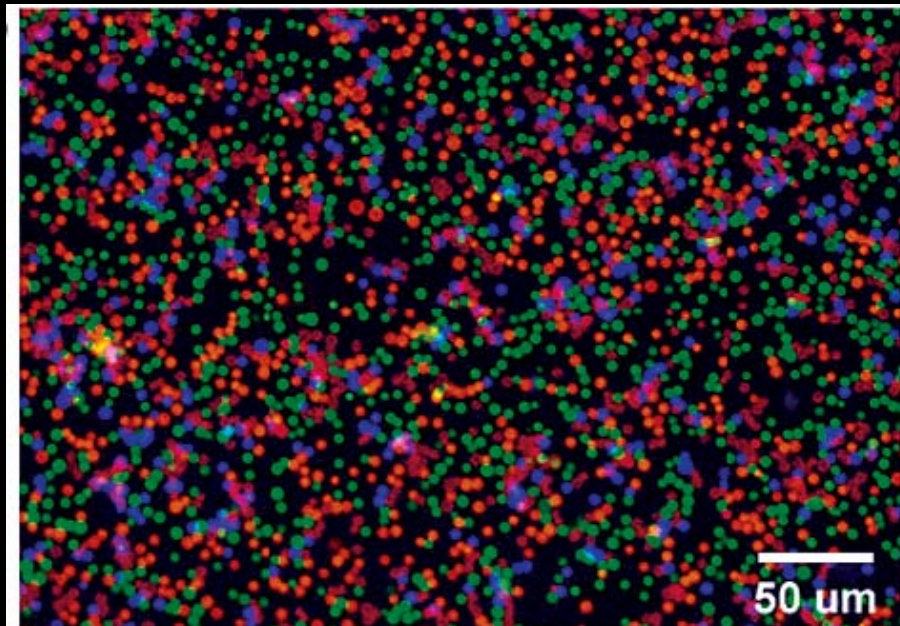
Introduction

- **Optical barcoding technology based on fluorophore-encoded microbeads has become an important alternative to planar chip-based assays for highly multiplexed gene and protein screening.**
- **Quantum dot (QD)-doped microbeads with unique optical properties offer key advantages over microspheres labelled with traditional organic fluorophores, such as a dramatically increased number of possible barcodes, and single light source excitation of multiple colors.**
- **Microbead barcodes are produced and resolved based on spectral information.**
- **An improved brightness of about 50-100 times is observed in case of mesoporous materials due to their large surface area (150 m² per gram of beads) when compared to nonporous beads of the same size.**
- **Mesoporous silica have an ordered structure of nanopores in the size range of 2-100 nm diameter and it provides a stable and accessible surface for immobilizing a variety of functional molecules and small particles.**



← Schematic representation of multivalent hydrophobic interactions between the surfactant (TOPO) molecules on the QD surface and the C-18 hydrocarbon molecules on the pore walls.

True-color fluorescence images of mesoporous silica beads (5 μm diameter, 32 nm pore size) doped with single-color quantum dots emitting light at 488 (blue), 520 (green), 550 (yellow), 580 (orange), or 610 nm (red). (a) Wide-field view of a large population of doped beads that were prepared in batches and then mixed, and (b) enlarged views of single monochromatic beads. Each bead contains up to 2 million dots of the same color.



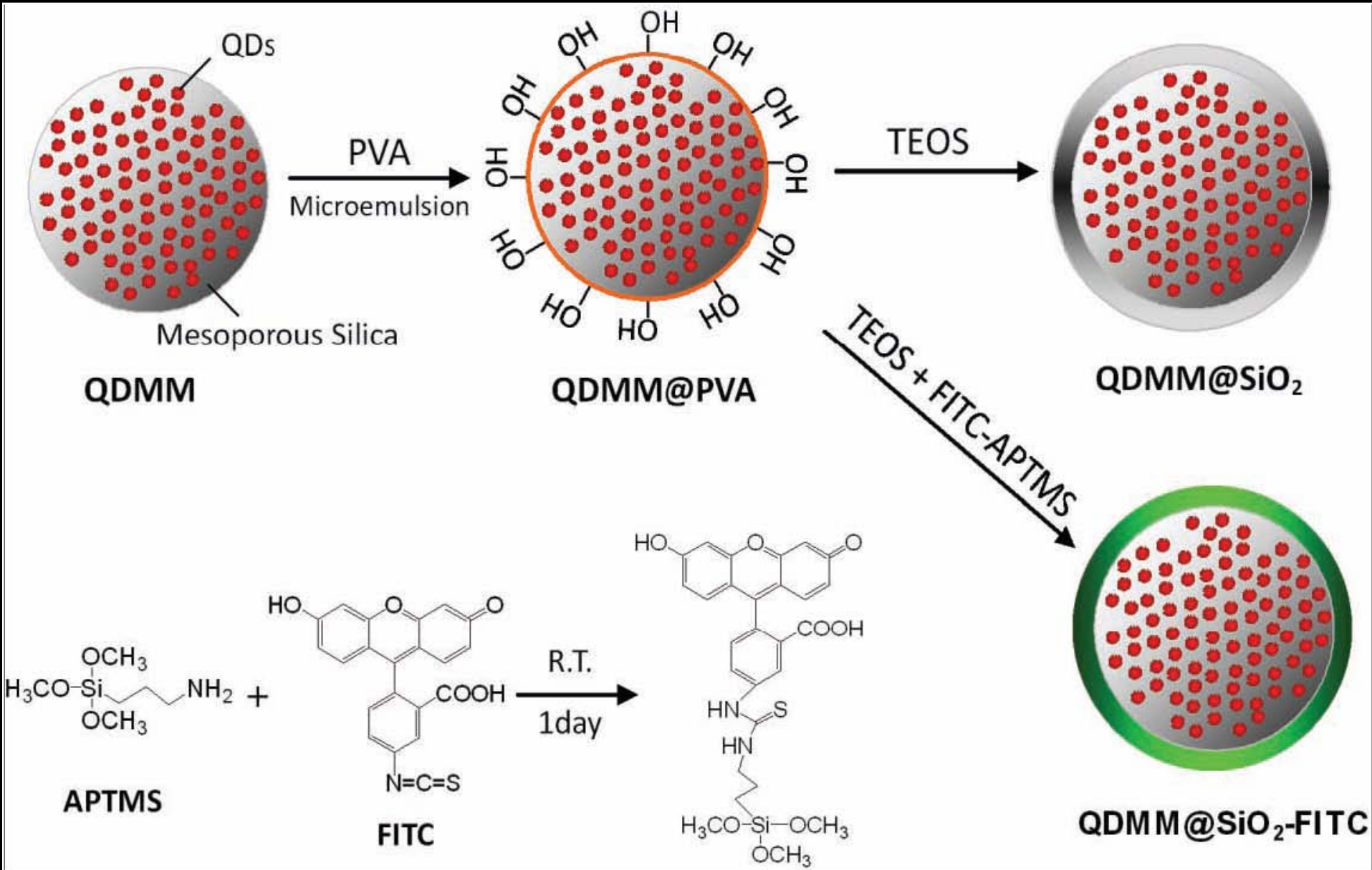
Limitations of the quantum dot-doped mesoporous microbeads (QDMMs) in optical barcoding technology

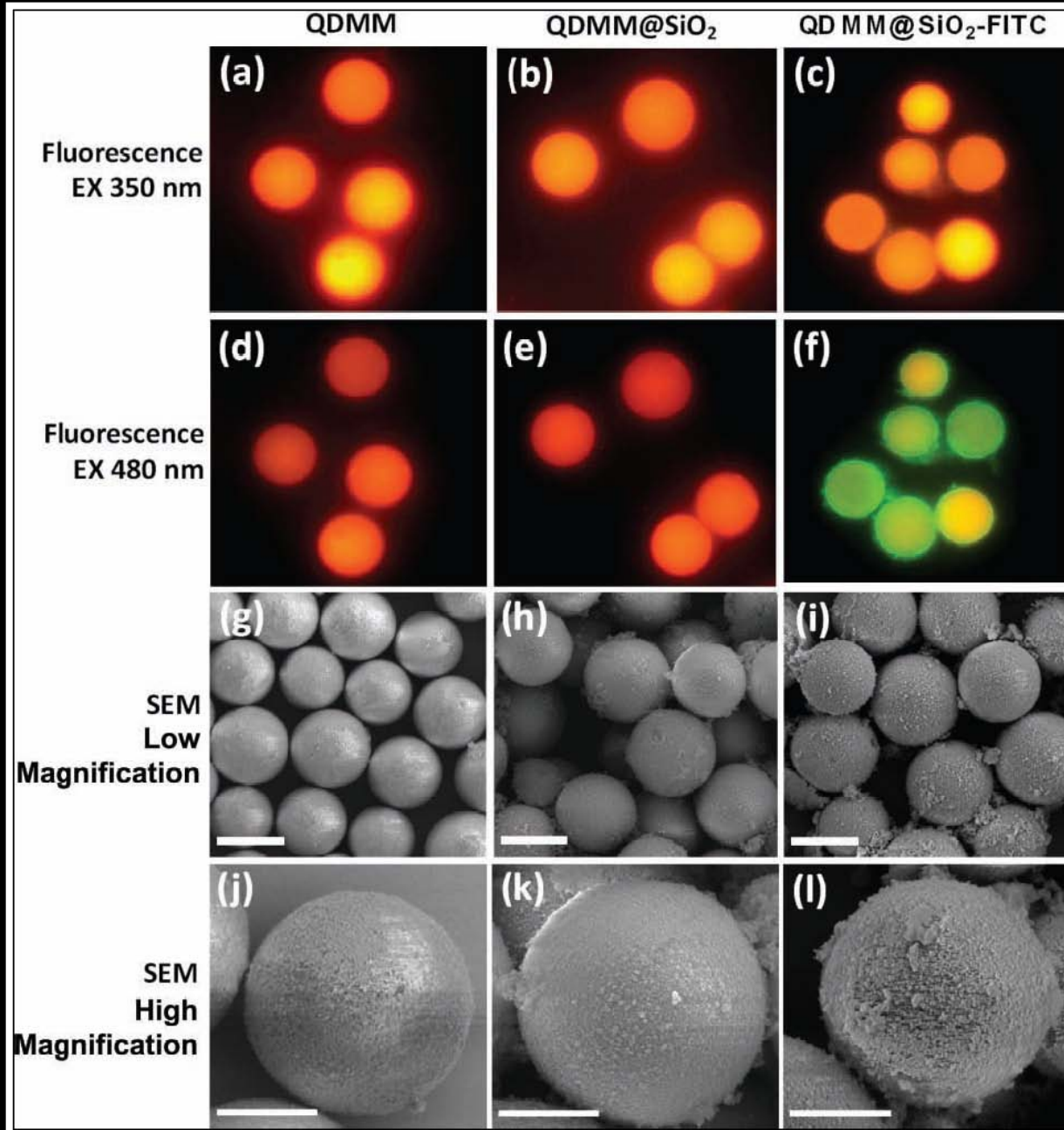
- ❖ Since the microbeads are mesoporous, potential leaching and chemical-induced degradation (e.g., spectral shift and intensity variation) of the embedded QDs are of major concern. i.e. QDs slowly diffuse out of the pores in organic solvents, resulting in a loss of barcode fluorescence intensity.
- ❖ Fluorescence fluctuation is often observed in different biological buffers or in reactions with bioconjugation crosslinking reagents.
- ❖ High hydrophobicity of the QDMMs leads to poor water solubility and lack of functional groups for bioconjugation.

In this paper

- ❖ A simple method for encapsulation of quantum dot-doped mesoporous microbeads with silica shells to stabilize the embedded QDs.
- ❖ Encapsulation using fluorescent silane precursors help in microscopic examination of the silane shells on the QDMMs.
- ❖ Studies were conducted to compare the physical and chemical stability of the QDMMs before and after silica coating.

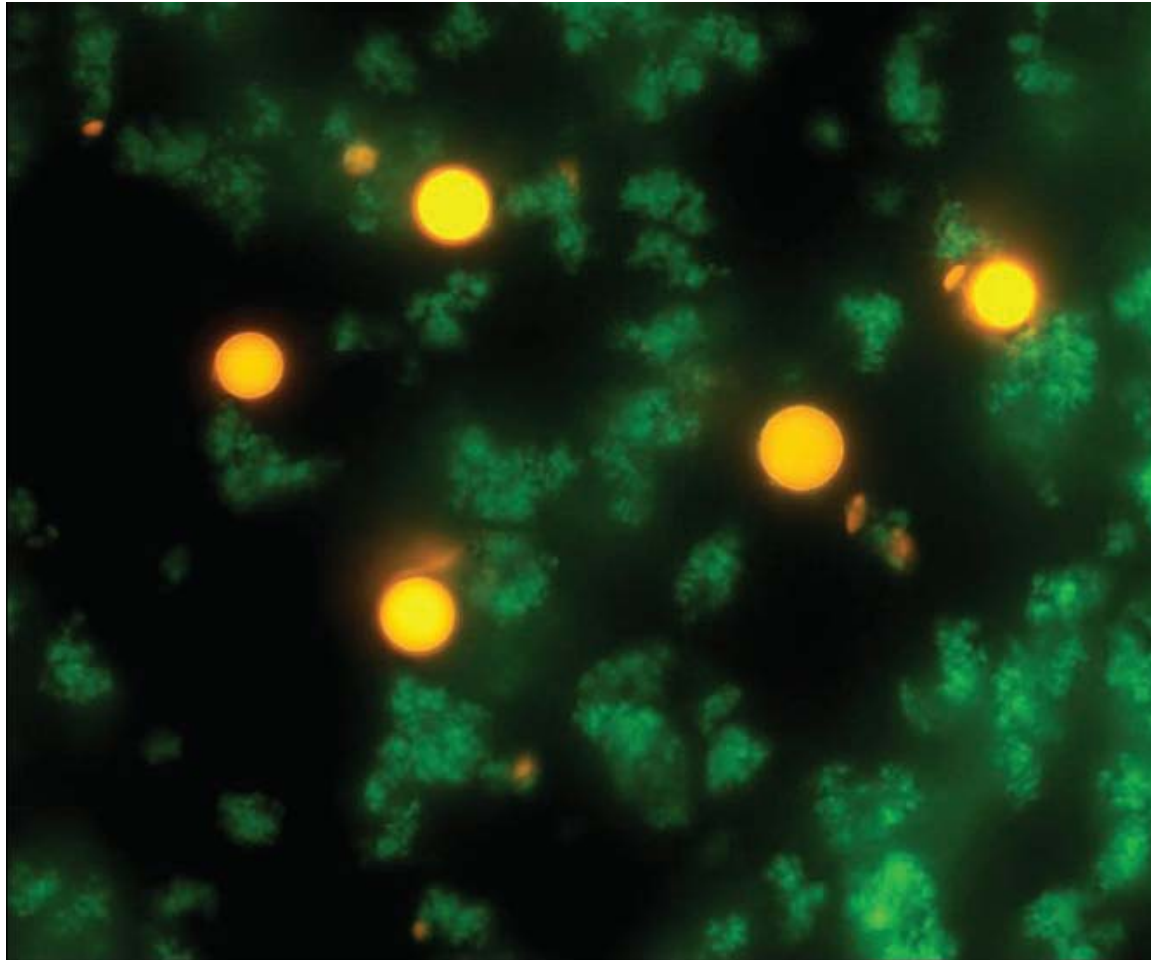
Schematic illustration of key steps involved in silica encapsulation of QDMMs.





Fluorescence and SEM images of QDMMs before and after silica encapsulation. Scale bars for (g-i) 5 μ m, (j-l) 2 μ m.

Role of PVA in silica shell formation



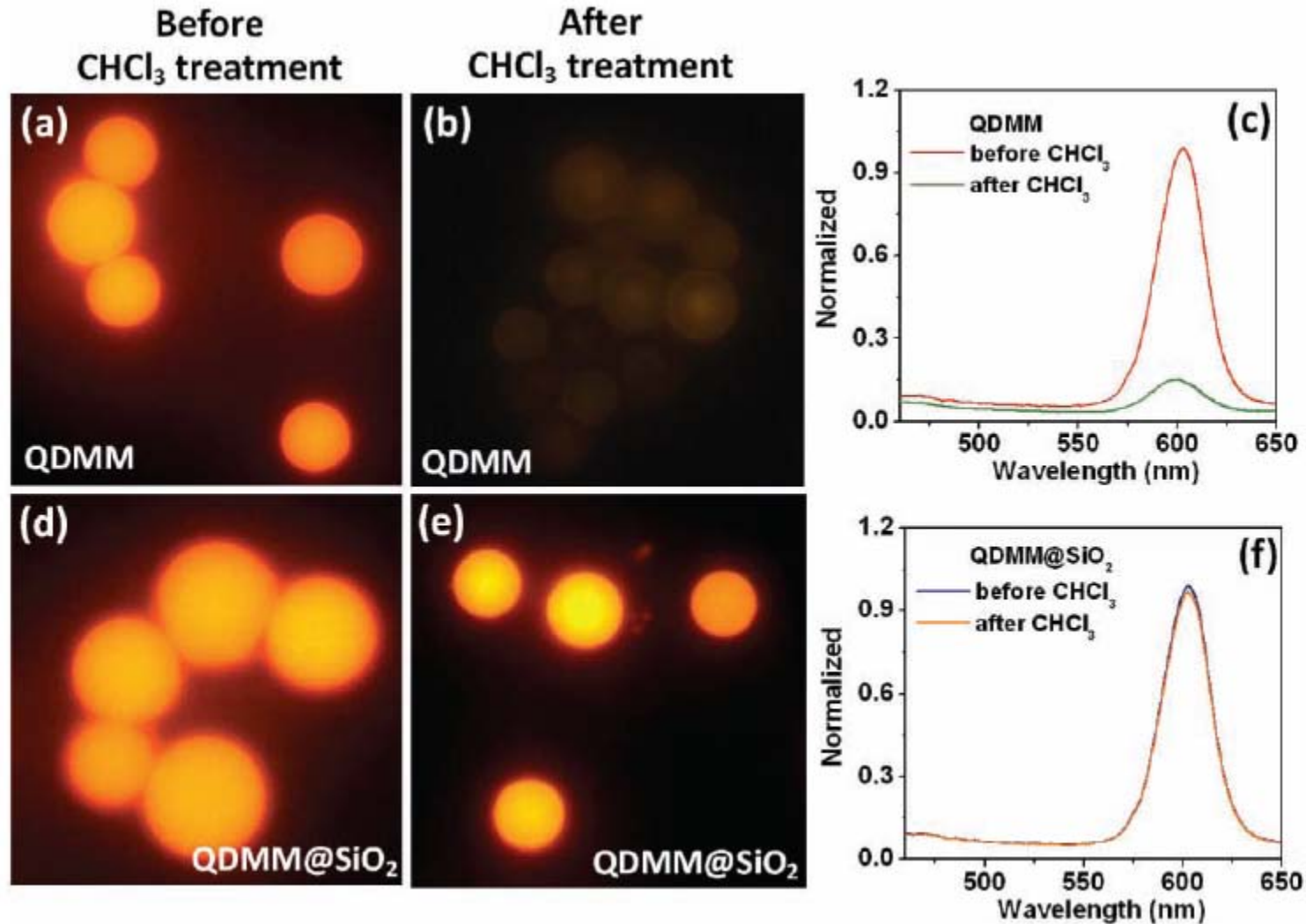
Fluorescence imaging of QDMM@SiO₂ synthesized without the microemulsion process.

Without PVA, the incompatibility between the hydrophobic hydrocarbons on microbead surface and hydrophilic silane compounds hinders nucleation of silane onto microbead surface.

They self-nucleate and condense into irregularly shaped aggregates (green aggregates in the background).

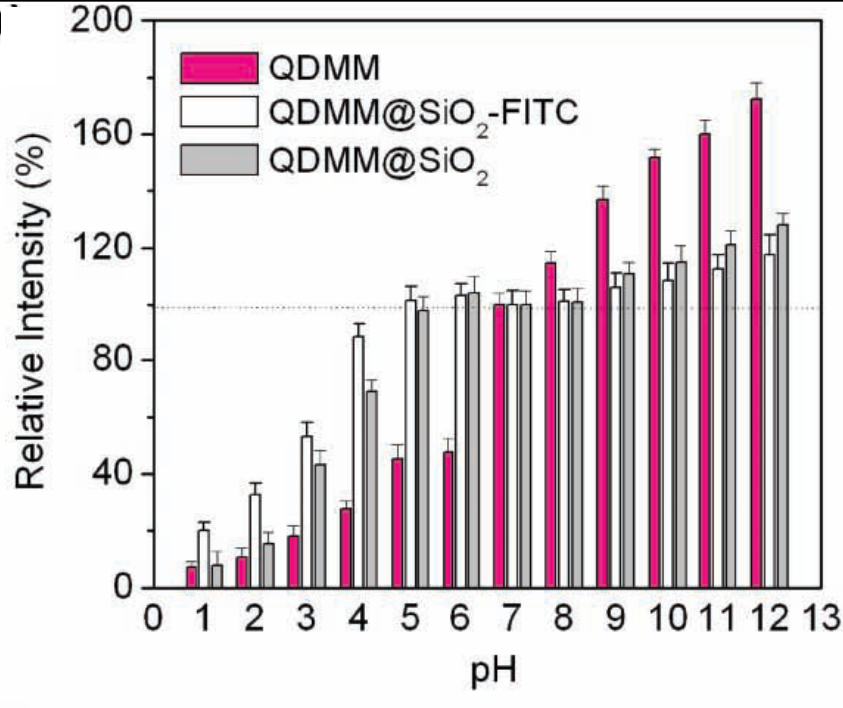
Physical and chemical stability of the embedded QDs

Solvent induced nanoparticle leaching - Study using Fluorescence imaging



Fluorescence microscopic images of (a,b) QDMMs and (d,e) QDMM@SiO₂ before and after sonication for 2 min in chloroform, respectively.

Effect of pH on QDMMs and QDMM@SiO₂



Microbeads were suspended in aqueous solutions with pH values ranging from 1 to 12 (fluorescence at neutral pH was set to 100%).

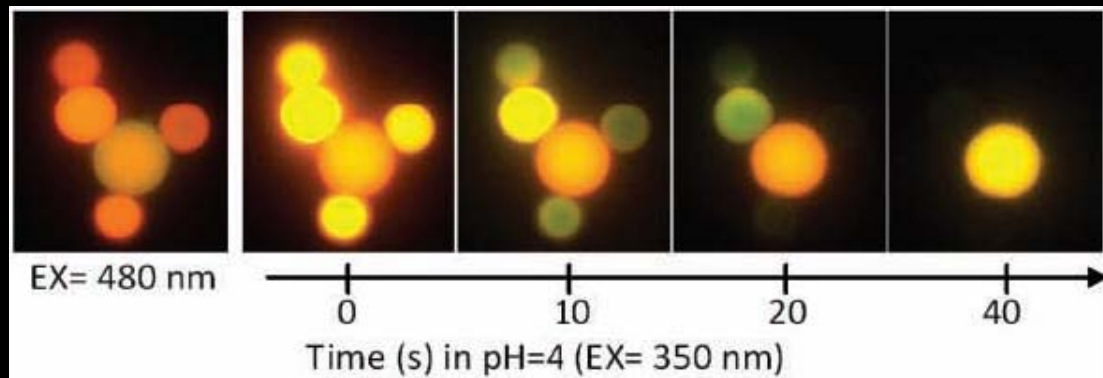
In absence of silica shell,

- Small changes in pH result in significant fluorescence fluctuation.

- Reducing pH from 7 to 5–6, fluorescence intensity of the embedded QDs is reduced by more than 50%.

- When the pH value is increased by 1 (to pH 8), the fluorescence signal is enhanced by approximately 20%.

In situ microscopic imaging of QDMM and QDMM@SiO₂-FITC at pH 4. The QDMMs are quenched significantly faster than the silica encapsulated microbeads.

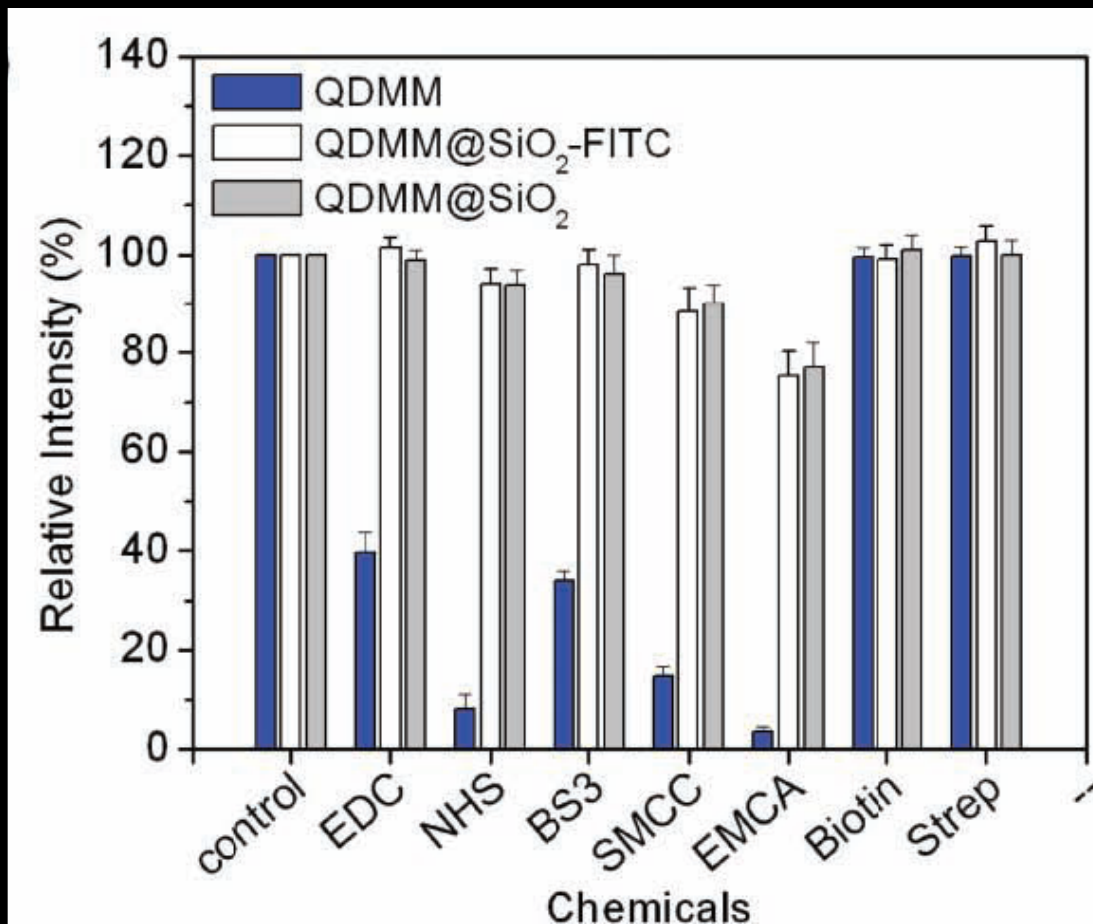


Stability of the QDMMs and QDMM@SiO₂ against chemical treatment

Chemical stability of QDMM, QDMM@SiO₂ and QDMM@SiO₂-FITC when treated with common bioconjugation reagents.

- Cross linking reagents such as EDC, NHS and SMCC decreases the fluorescence of QDMM up to 90% after 24-hour incubation. In contrast that of QDMM@SiO₂ is nearly unaffected under the same condition.

- EMCA quenches the fluorescence of QDs in QDMM@SiO₂ by 20%. Thicker silica shell or decorating additional protection layers on top of the silica shell could possibly solve the problem.



Summary

- ❖ A method for encapsulating QD barcodes with silica shells.
- ❖ To stabilize hydrophobic mesoporous microbeads in polar solvents, a microemulsion process was found highly effective in uniformly coating amphiphilic polymers onto the microbead surface. Subsequent condensation of silane compounds results in QDMM@SiO₂ with significantly improved chemical stability.
- ❖ This new method solved key problems in QD barcoding technology such as nanoparticle leaching. When QDMM@SiO₂ is suspended in organic solvents such as chloroform, in which hydrophobic QDs have excellent solubility, negligible nanoparticle leaching is detected.
- ❖ The barcode fluorescence remains stable when treated for 24 h with pH 5–8 aqueous solutions and a variety of common bioconjugation crosslinkers.
- ❖ Further optimization of this method such as fine tuning the silica shell thickness and linking silica surface with non-fouling polymers will allow clinical translation of the QD barcoding technology.

Thanks
Thanks