

The Pivotal Radical Intermediate $[\text{Au}_{21}(\text{SR})_{15}]^+$ in the Ligand-Exchange-Induced Size-Reduction of $[\text{Au}_{23}(\text{SR})_{16}]^-$ to $\text{Au}_{16}(\text{SR})_{12}$

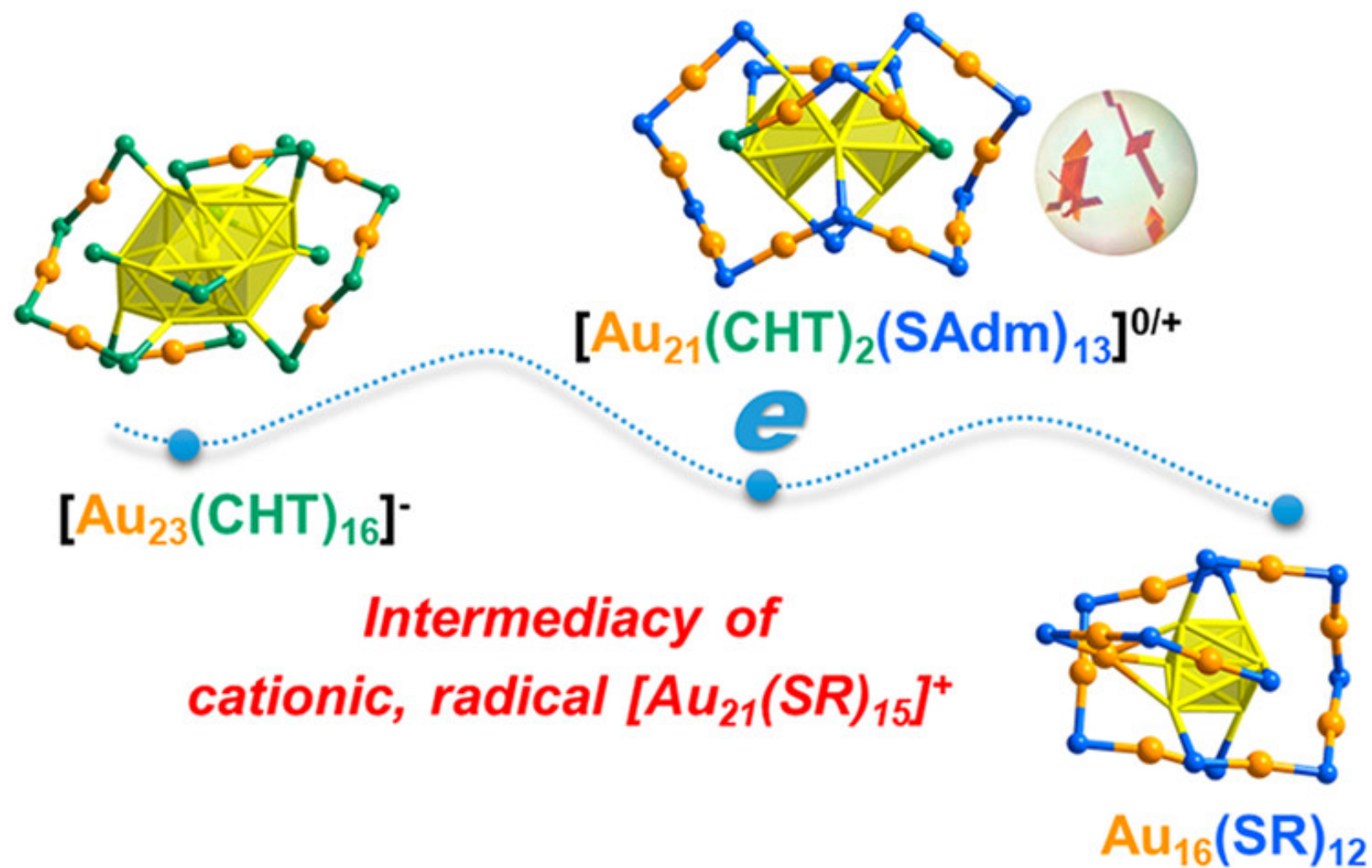
Mengting Cui, Yanan Shi, Xiangyu Ma, Qingliang Li, Ling Chen, Lichao Zhang, Junfei Wu, Haizhu Yu,* and Manzhou Zhu*

Department of Chemistry and Centre for Atomic Engineering of Advanced Materials, Key Laboratory of Structure and Functional Regulation of Hybrid Materials of Ministry of Education, Institutes of Physical Science and Information Technology and Anhui Province Key Laboratory of Chemistry for Inorganic/Organic Hybrid Functionalized Materials, Anhui University, Hefei, Anhui 230601, China.

Published: February 2, 2024

Swetashree Acharya
30.03.2024

Abstract



Total Structure Determination of $\text{Au}_{21}(\text{S-Adm})_{15}$ and Geometrical/Electronic Structure Evolution of Thiolated Gold Nanoclusters

Shuang Chen,^{†,||} Lin Xiong,^{‡,||} Shuxin Wang,^{*,†} Zhongyun Ma,[‡] Shan Jin,[†] Hongting Sheng,[†] Yong Pei,^{*,‡} and Manzhou Zhu^{*,†}

[†]Department of Chemistry and Center for Atomic Engineering of Advanced Materials, Anhui University, Hefei, Anhui 230601, People's Republic of China

[‡]Department of Chemistry, Key Laboratory of Environmentally Friendly Chemistry and Applications of Ministry of Education, Xiangtan University, Xiangtan, Hunan 411105, People's Republic of China

Nanoscale



PAPER

View Article Online

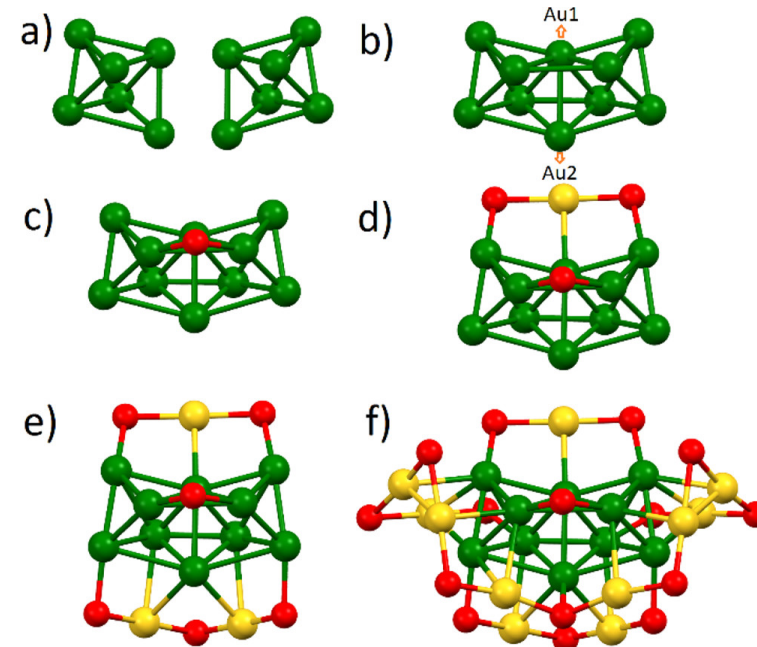
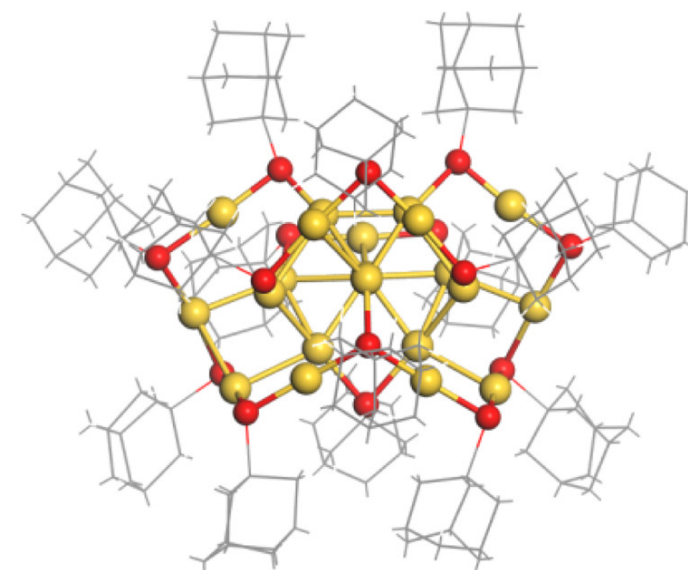
View Journal | View Issue



Cite this: *Nanoscale*, 2020, 12, 23694

Structure determination of a metastable $\text{Au}_{22}(\text{SAdm})_{16}$ nanocluster and its spontaneous transformation into $\text{Au}_{21}(\text{SAdm})_{15}^{\dagger}$

Qinzhen Li,^a Sha Yang,^b Tao Chen,^a Shan Jin,^{id} Jinsong Chai,^{*,b,c} Hui Zhang^{id} ^{*,a} and Manzhou Zhu^{id} ^{*,b,c}



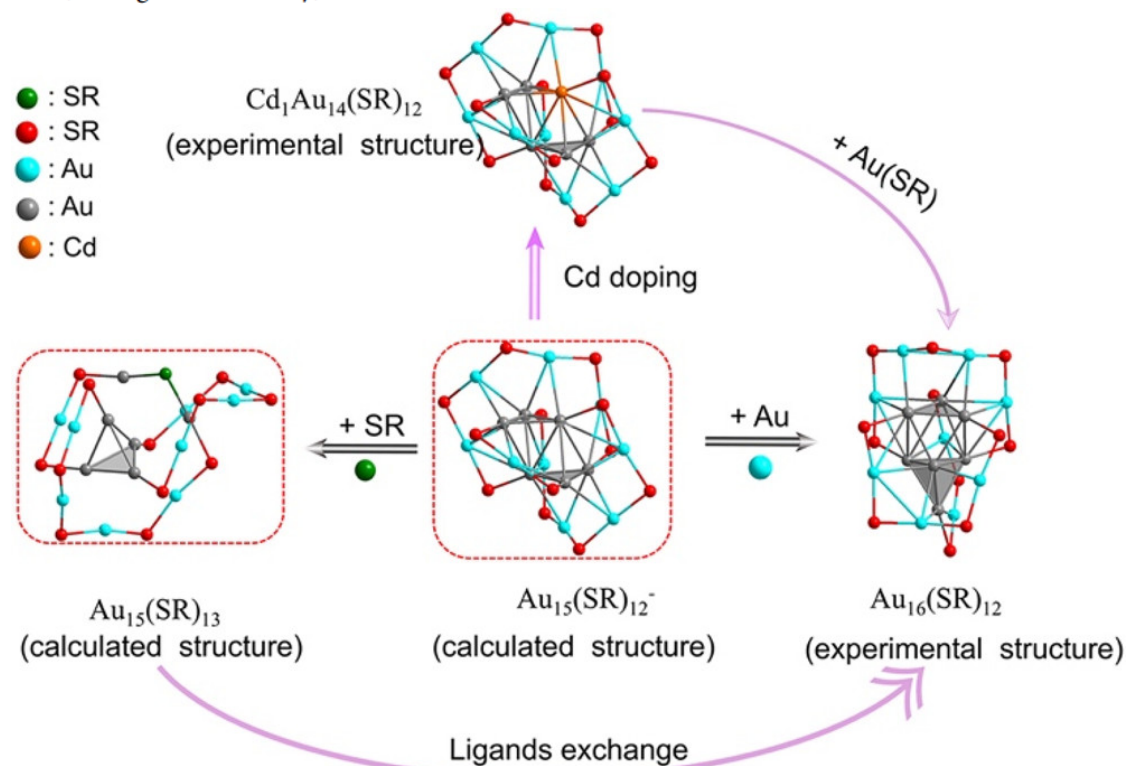
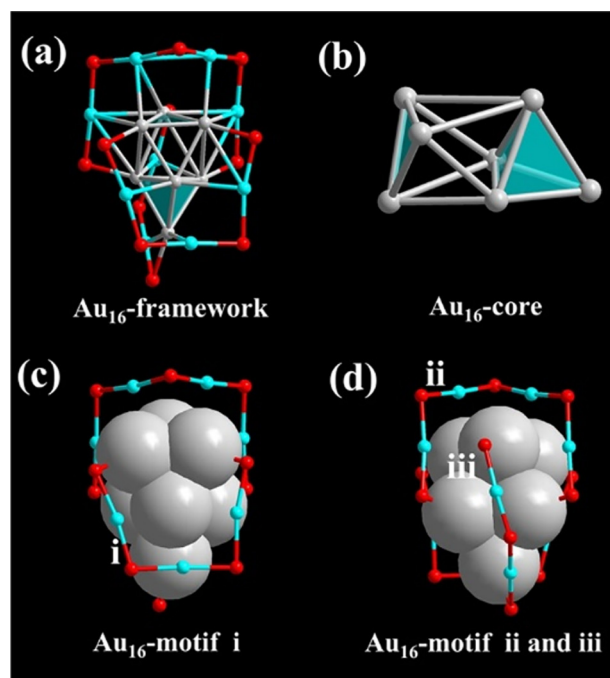
Total Structure Determination of $\text{Au}_{16}(\text{S-Adm})_{12}$ and $\text{Cd}_1\text{Au}_{14}(\text{StBu})_{12}$ and Implications for the Structure of $\text{Au}_{15}(\text{SR})_{13}$

Sha Yang,^{†,||} Shuang Chen,^{†,||} Lin Xiong,^{‡,||} Chong Liu,^{§,ID} Haizhu Yu,^{†,ID} Shuxin Wang,^{*,†,ID}
Nathaniel L. Rosi,^{§,ID} Yong Pei,^{*,‡,ID} and Manzhou Zhu^{*,†,ID}

[†]Department of Chemistry and Centre for Atomic Engineering of Advanced Materials, Anhui Province Key Laboratory of Chemistry for Inorganic/Organic Hybrid Functionalized Materials, Anhui University, Hefei, Anhui 230601, China

[‡]Department of Chemistry, Key Laboratory of Environmentally Friendly Chemistry and Applications of MOE, Xiangtan University, Xiangtan, Hunan 411105, China

[§]Department of Chemistry, University of Pittsburgh, Pittsburgh, Pennsylvania 15213, United States



Motivation

- In recent years, ligand-exchange-induced enhancement in photoemission, chirality, or solubility has greatly expanded the practical applications of these ultrasmall nanoclusters in bioclinics, catalysis, and interfacial assembly.
- Structural characterization on the key intermediate, and ideally with single crystal strategies, becomes a highly demanding and rewarding issue.

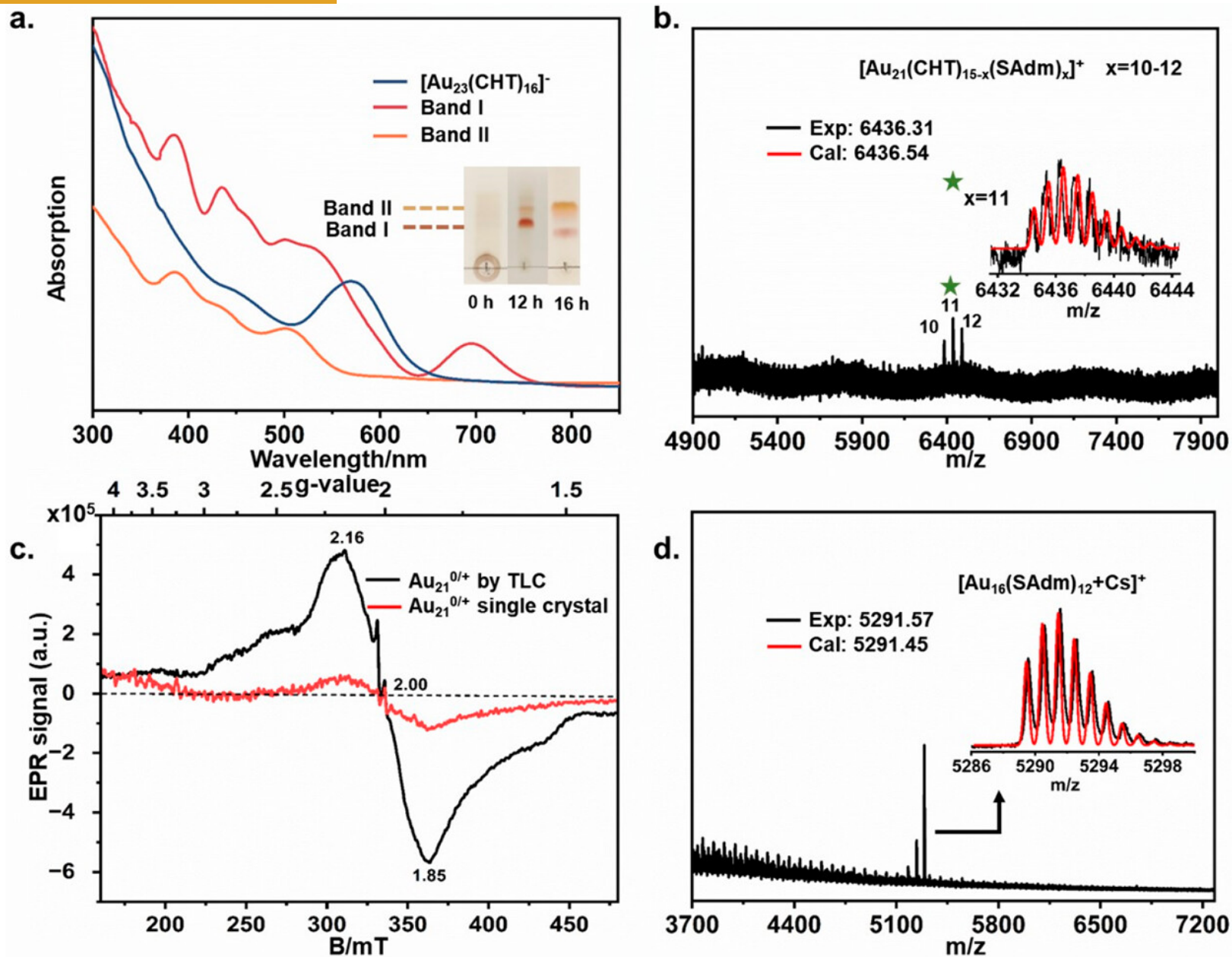
Importance

- Explored the reaction and especially the kinetic profiles of the reaction of $[\text{Au}_{23}(\text{CHT})_{16}]^-$ with HSAdm.
- Showed that, the neutral, 6e $\text{Au}_{21}(\text{SR})_{15}$ is a resting state, and its 1e-reduction state, i.e., the 5e, cationic radical, $[\text{Au}_{21}(\text{SR})_{15}]^+$, is an active intermediate to mainly account to the formation of thermodynamic stable Au_{16} products. The oxidation of $8\text{e} \rightarrow 6\text{e} \rightarrow 4\text{e}$ occurs associated with the core size-reduction (via preferentially removing the corner Au atoms on the core surface and the longer $\text{Au}_n\text{S}_{n+1}$ staples)

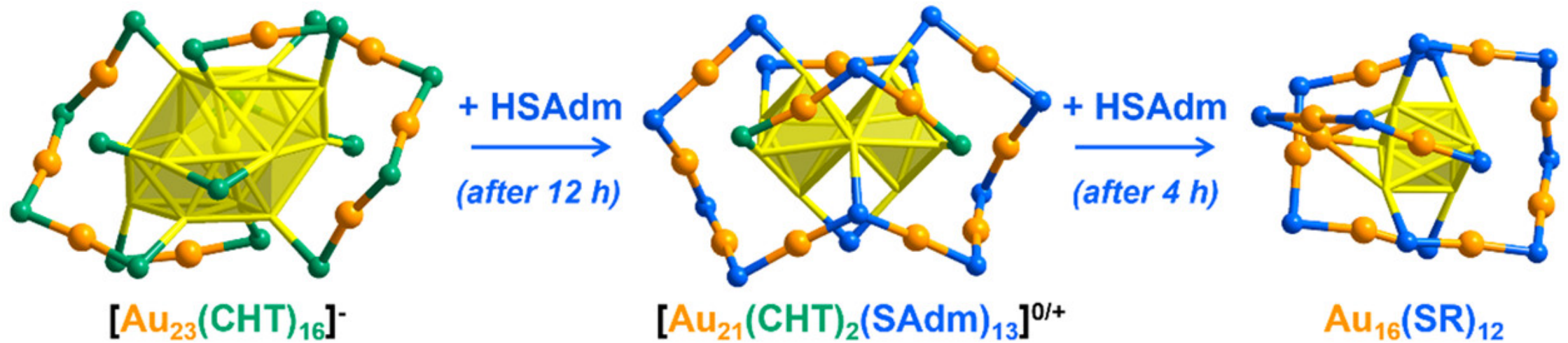
Introduction

- The atomic precision of sub-nanometer-sized metal nanoclusters makes it possible to elucidate the kinetics of metal nanomaterials from the molecular level.
- Herein, the size reduction of an atomically precise $[\text{Au}_{23}(\text{CHT})_{16}]^-$ (HCHT = cyclohexanethiol) cluster upon ligand exchange with HSAdm (1-adamantanethiol) has been reported.
- During the 16 h conversion of $[\text{Au}_{23}(\text{CHT})_{16}]^-$ to $\text{Au}_{16}(\text{SR})_{12}$, the neutral 6e $\text{Au}_{21}(\text{SR})_{15}$, and its 1e-reduction state, i.e. the 5e, cationic radical, $[\text{Au}_{21}(\text{SR})_{15}]^+$, are active intermediates to account for the formation of thermodynamically stable Au_{16} products.
- The combination of spectroscopic monitoring by UV–vis with ESI-MS and DFT calculations indicates the preferential size-reduction on the corner Au atoms on the core surface and the terminal Au atoms on longer $\text{Au}_n\text{S}_{n+1}$ staples.
- This study provides a reassessment on the electronic state of the Au_{21} structure and highlights the single electron transfer processes in cluster systems and thus the importance of the EPR analysis on the mechanistic issues.

Results and discussion

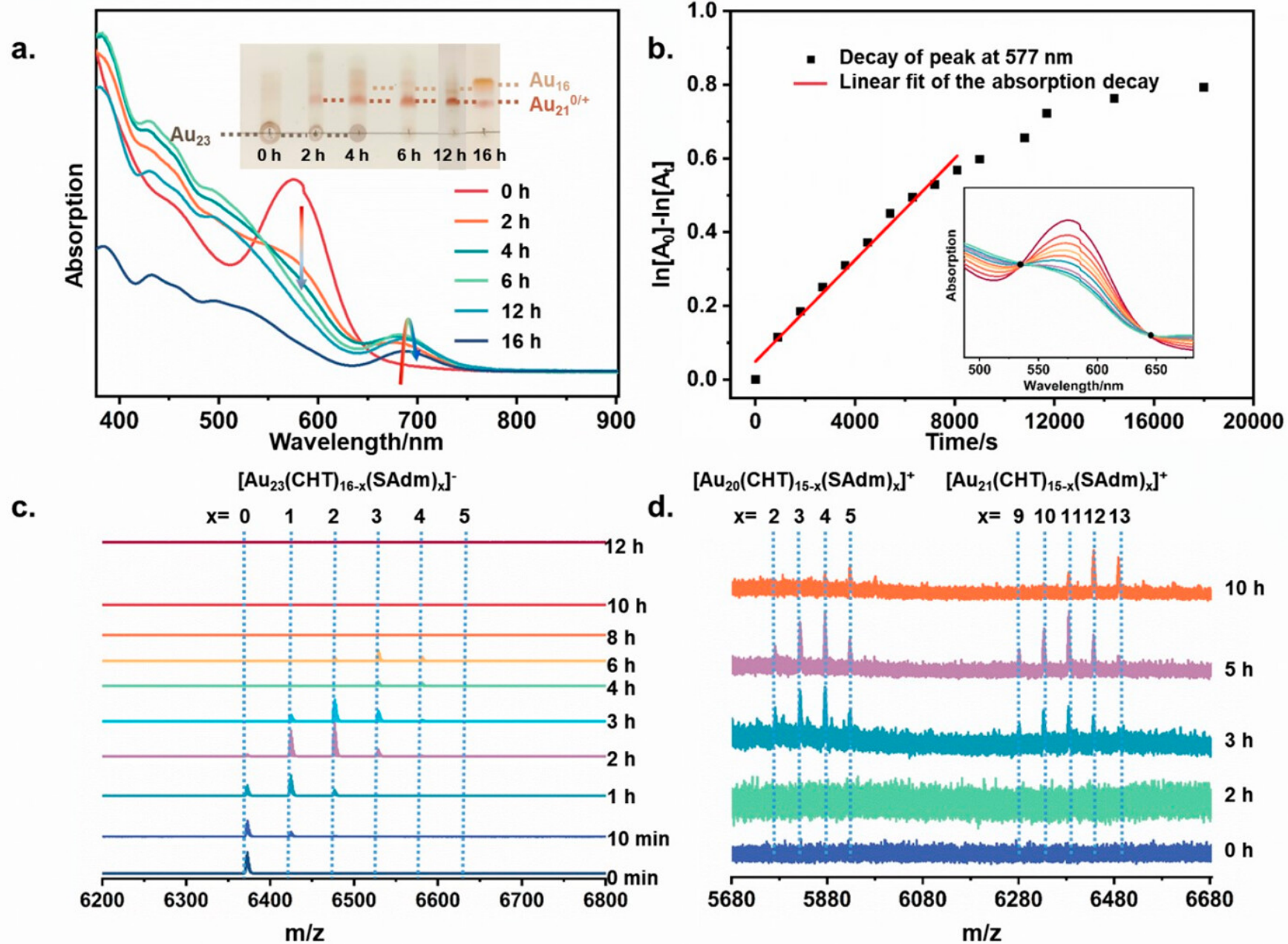


Scheme 1

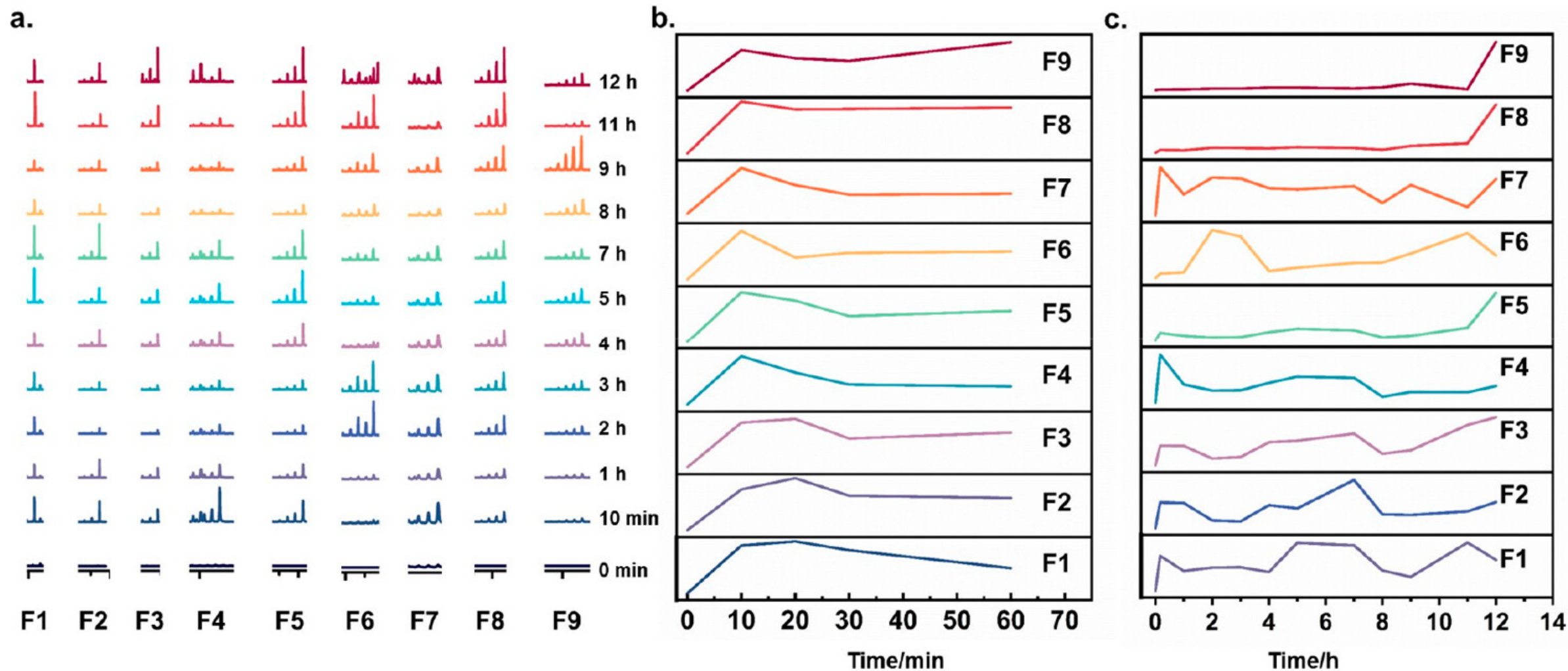


In DCM at RT

Reaction progress

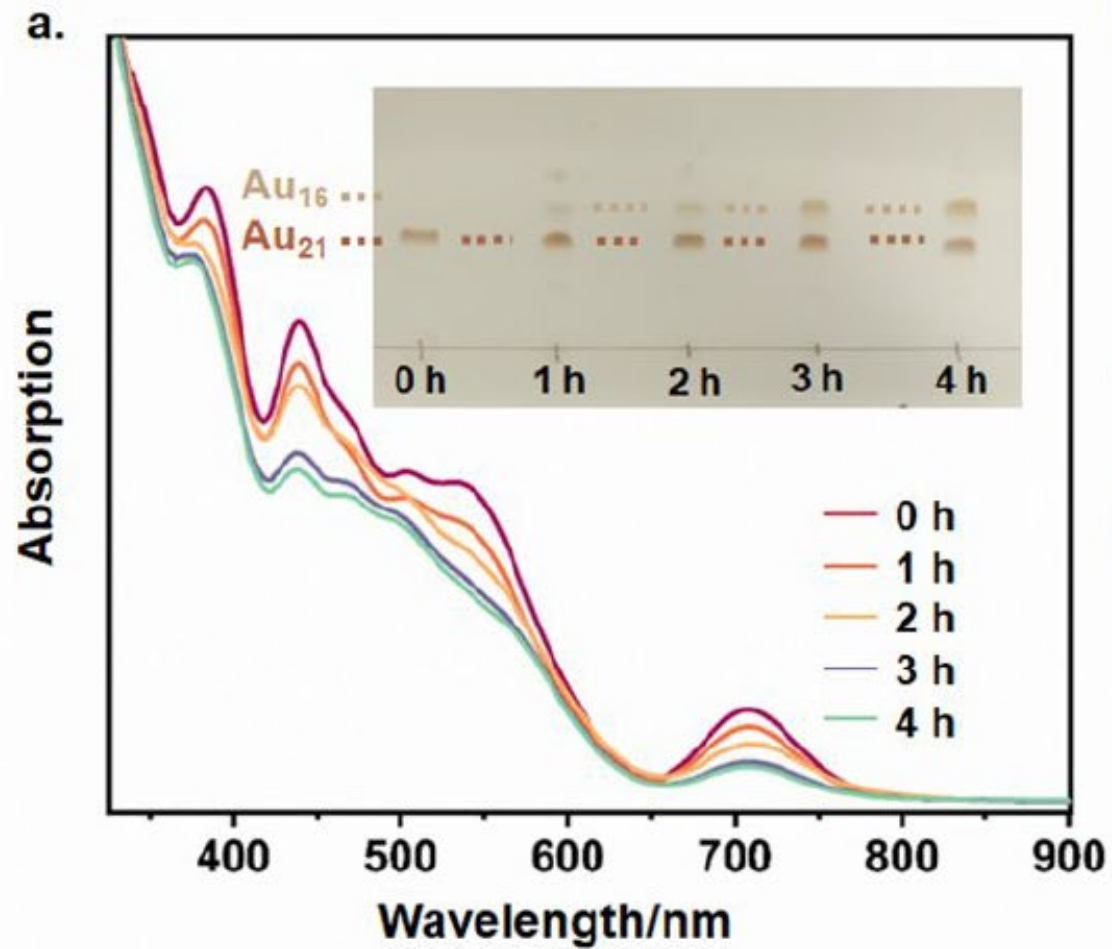


Au-SR complexes

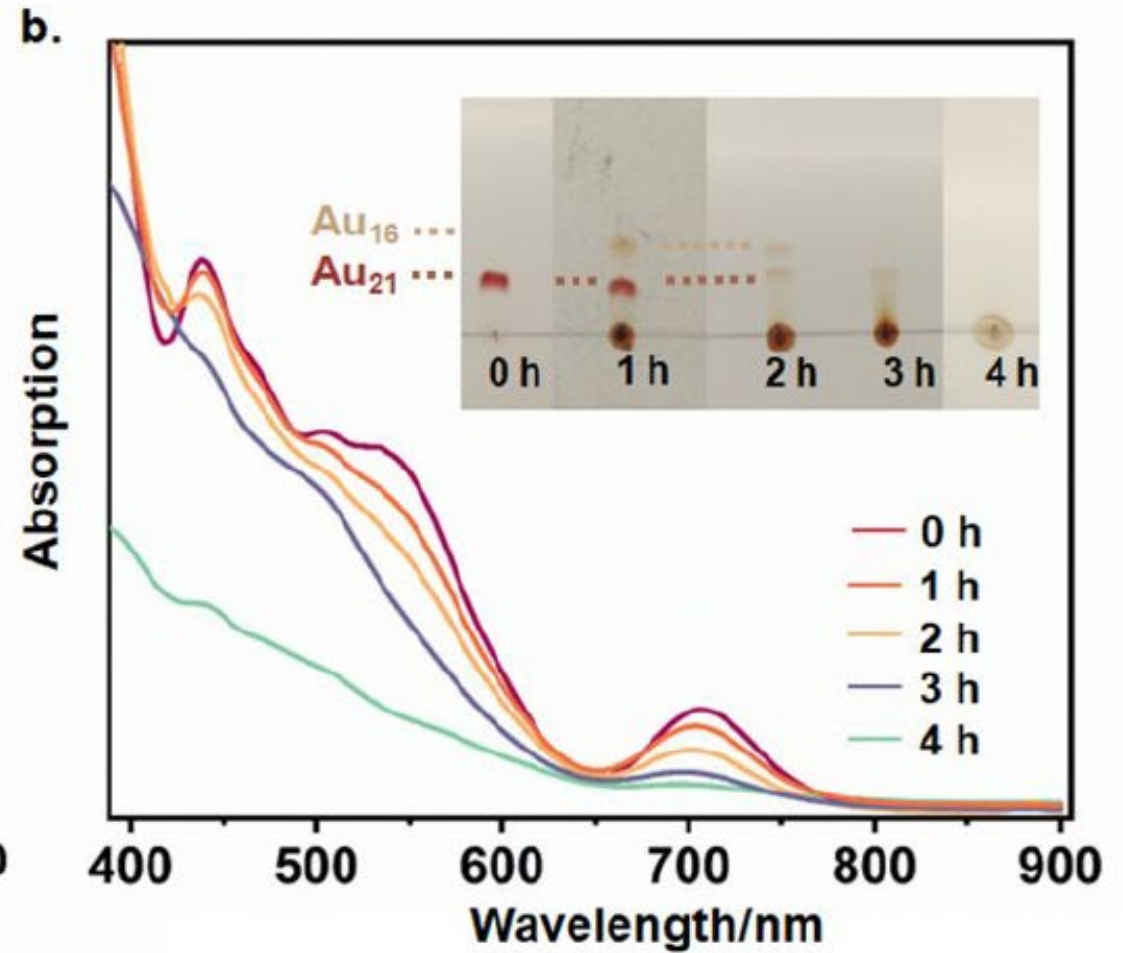


F1: $[\text{Au}_1\text{SAdm}_1\text{Cl}_1]^-$; F2: $[\text{Au}_1\text{SAdm}_2]^-$; F3: $[\text{Au}_2\text{SAdm}_3]^-$; F4: $[\text{Au}_3\text{SAdm}_3\text{CHT}_1]^-$; F5: $[\text{Au}_3\text{SAdm}_4]^-$;
 F6: $[\text{Au}_4\text{SAdm}_4\text{Cl}_1]^-$; F7: $[\text{Au}_4\text{SAdm}_4\text{CHT}_1]^-$; F8: $[\text{Au}_4\text{SAdm}_5]^-$; F9: $[\text{Au}_5\text{SAdm}_6]^-$.

Control experiment

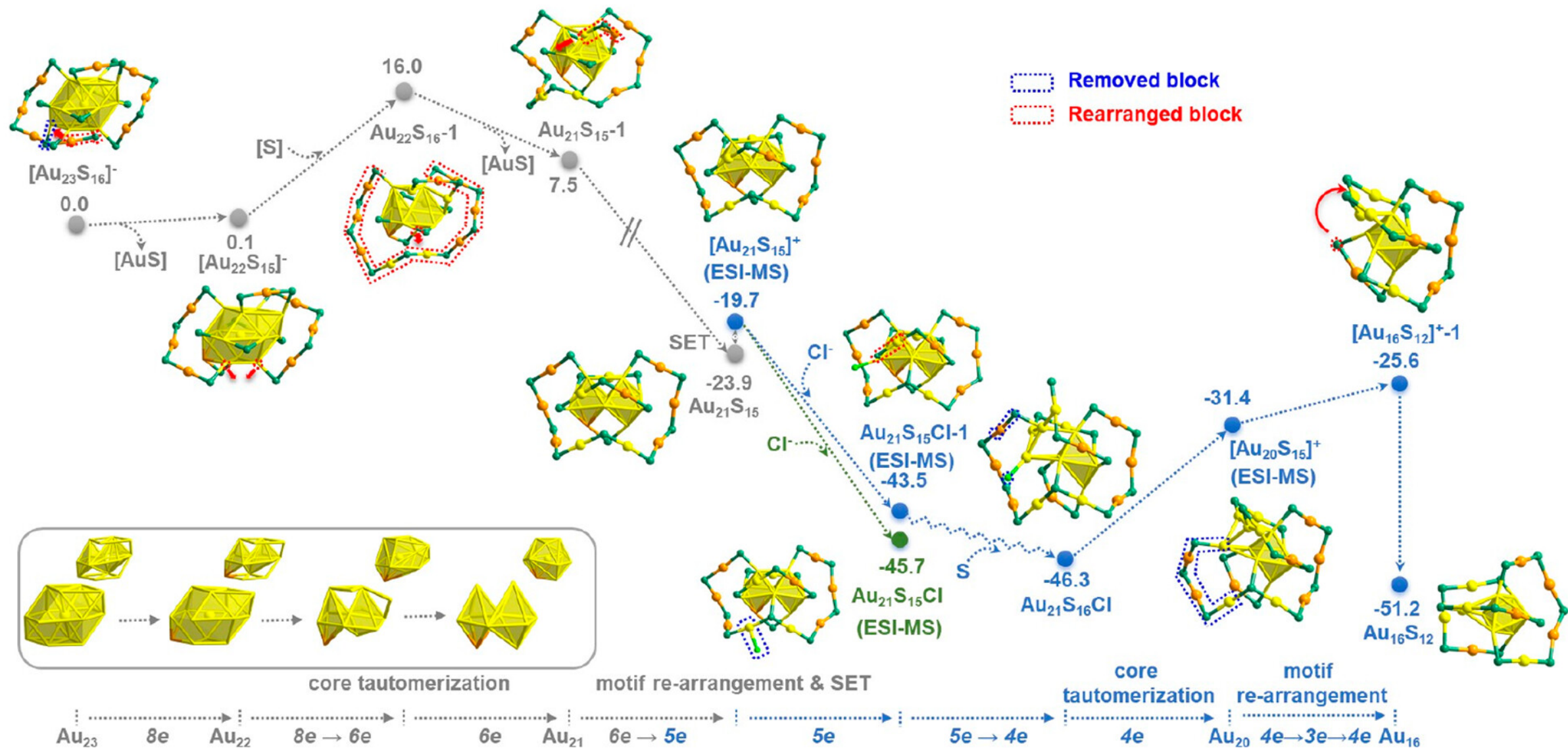


Presence of excess HSAdm



Absence of excess HSAdm

Energy profile



Conclusions

- ❖ This study reported a ligand-exchange-induced size-reduction of $[\text{Au}_{23}(\text{SR})_{16}]^-$ to $\text{Au}_{16}(\text{SR})_{12}$ and elucidates the intermediacy of one redox pair of $[\text{Au}_{21}(\text{SR})_{15}]^{0/+}$ by means of X-ray single crystal diffraction, UV-vis, ESI-MS, and EPR, etc.
- ❖ The radical type $[\text{Au}_{21}(\text{SR})_{15}]^+$ intermediate was identified, and its coexistence with the neutral charge states indicate the flexible electronic state of the $\text{Au}_{21}(\text{SR})_{15}$ framework.
- ❖ Herein, the single electron transfer processes in cluster systems and thus the importance of the EPR analysis on the mechanistic issues was emphasized.
- ❖ With the combination of in situ UV-vis, ESI-MS, and DFT calculations, the conversion mechanism was proposed, the Au atoms on the corner of the metallic core structure were found to be liable to be removed, and the $\text{Au}_{16}(\text{SR})_{12}$ was finally formed as the thermodynamic product.
- ❖ The size-reduction occurs preferentially on the corner Au atoms on the core surface or the terminal Au atoms on longer $\text{Au}_n\text{S}_{n+1}$ staples.