

# In Situ Encapsulation of Atomically Precise Nanoclusters in Reticular Frameworks via Mechanochemical Synthesis

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**VIP** **Hydrogen Production** **Very Important Paper**

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## Heteroatom-Doped Ag<sub>25</sub> Nanoclusters Encapsulated in Metal–Organic Frameworks for Photocatalytic Hydrogen Production

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Communication

## Design and Remarkable Efficiency of the Robust Sandwich Cluster Composite Nanocatalysts ZIF-8@Au<sub>25</sub>@ZIF-67

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# Motivation

The mechanochemical preparation of frameworks has been applied in various fields, such as nanoparticle immobilization, biomedical delivery and gas adsorption, notably for encapsulating enzymes and enhancing their catalytic activity.

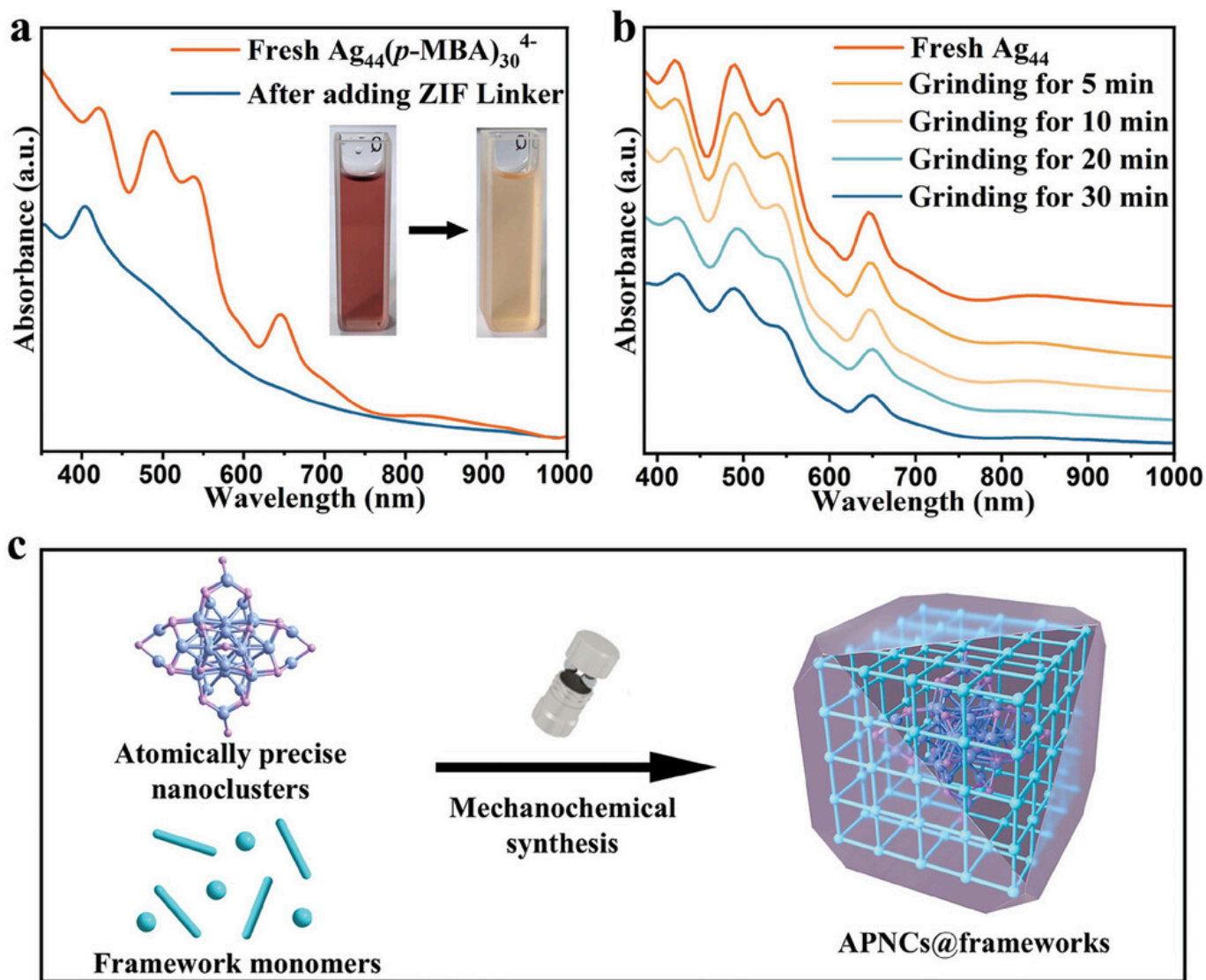
## Why this paper?

Talks about a universal method for fabricating diverse APNC@framework nanocomposites under mild conditions and in a short time- which is highly desirable but challenging.

# Introduction

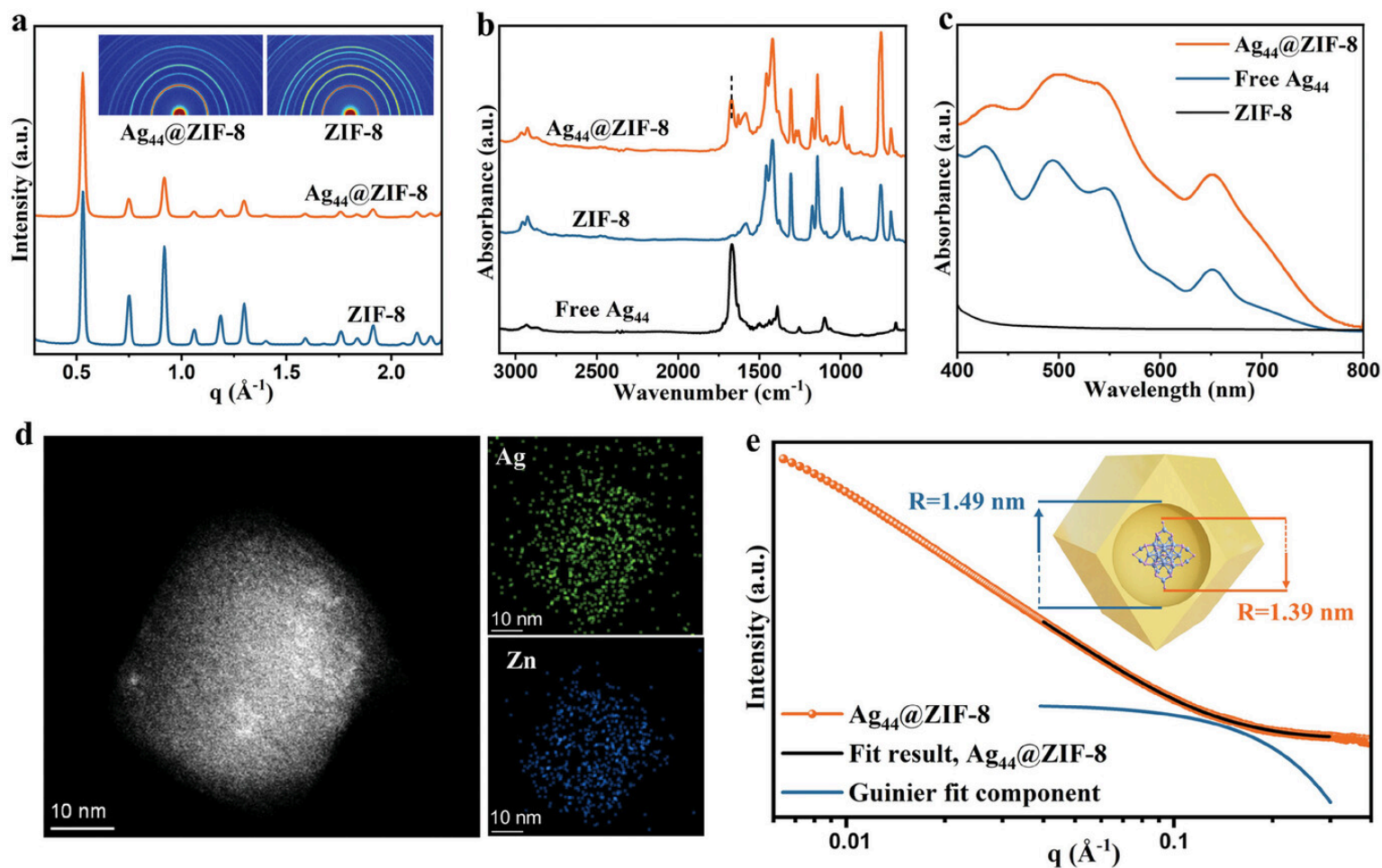
- The combination of atomically precise nanoclusters (APNCs) and reticular frameworks is promising for generating component-specific nanocomposites with emergent properties.
- Traditional liquid-phase synthesis often hampers this potential by damaging APNCs and limiting combination diversity.
- Here, mechanochemical synthesis to explore the encapsulation of diverse oil and water-soluble APNCs within various reticular frameworks is employed. establishing a database of 21 unique APNC–framework combinations, including metal–organic frameworks (MOFs), covalent–organic frameworks (COFs), hydrogen–bonded organic frameworks (HOFs), and multivariate MOFs.
- These framework coatings not only spatially immobilize APNCs but also secure their structures, preventing aggregation and degradation while enhancing stability and activity.
- The mechanochemical synthesis strategy facilitates tailored support screening, catering to specific needs, and shows promise for developing multifunctional systems, including enzyme–APNC@frameworks material for cascade reactions.

# Results and discussion



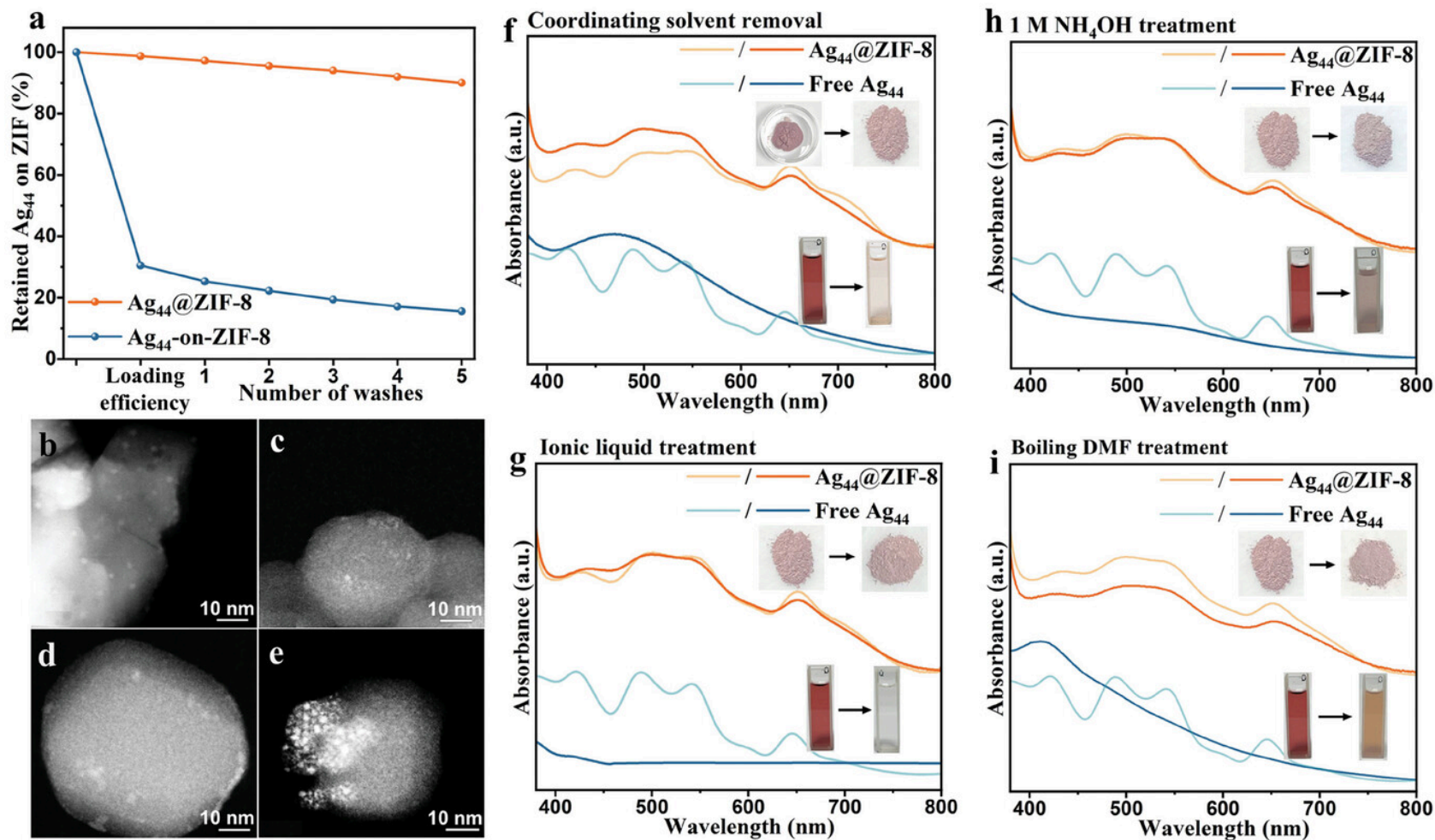
**Figure 1.**  $\text{Ag}_{44}$  compatibility studies and APNCs@frameworks synthesis. a) UV-vis spectra of fresh  $\text{Ag}_{44}$  and a mixture of  $\text{Ag}_{44}$  and the MOF linker. b) UV-vis spectrum of  $\text{Ag}_{44}$  after grinding at 480 rpm. c) Synthesis of APNCs@frameworks.

# Characterization of the $\text{Ag}_{44}@\text{ZIF-8}$ composite

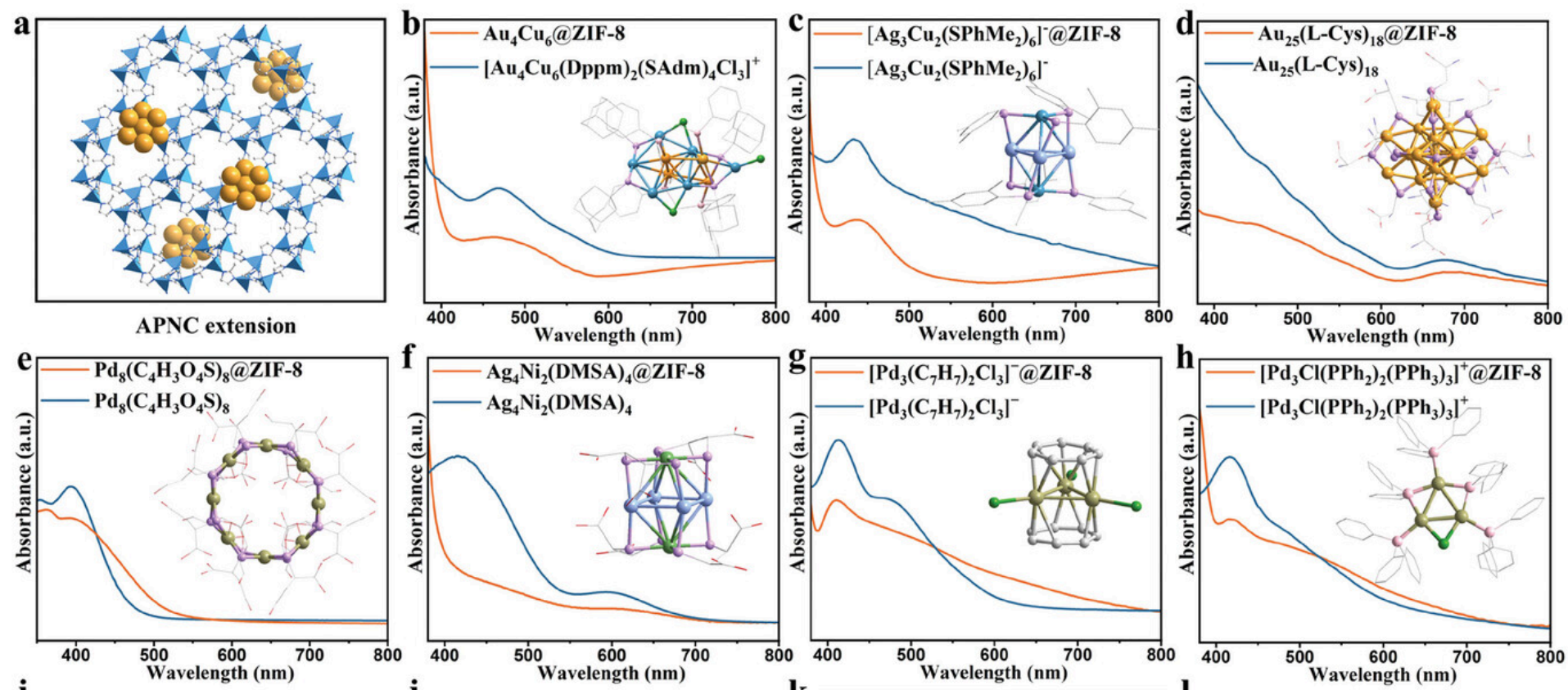


**Figure 2.** Characterization of the  $\text{Ag}_{44}@\text{ZIF-8}$  composite. a) WAXS patterns of  $\text{ZIF-8}$  and  $\text{Ag}_{44}@\text{ZIF-8}$ . b) FTIR data of  $\text{ZIF-8}$ ,  $\text{Ag}_{44}$ , and  $\text{Ag}_{44}@\text{ZIF-8}$ . c) UV-vis spectra of  $\text{ZIF-8}$ ,  $\text{Ag}_{44}$ , and  $\text{Ag}_{44}@\text{ZIF-8}$ . d) HAADF-STEM image of  $\text{Ag}_{44}@\text{ZIF-8}$  and the corresponding elemental mappings of Ag and Zn. e) Fitted SAXS patterns and single-component fits (power law and Guinier fits) of  $\text{Ag}_{44}@\text{ZIF-8}$ . The inset shows the radius (arrows) of  $\text{Ag}_{44}$  and the observed mesopores in  $\text{Ag}_{44}@\text{ZIF-8}$ .

# Enhanced stability of $\text{Ag}_{44}@\text{ZIF-8}$

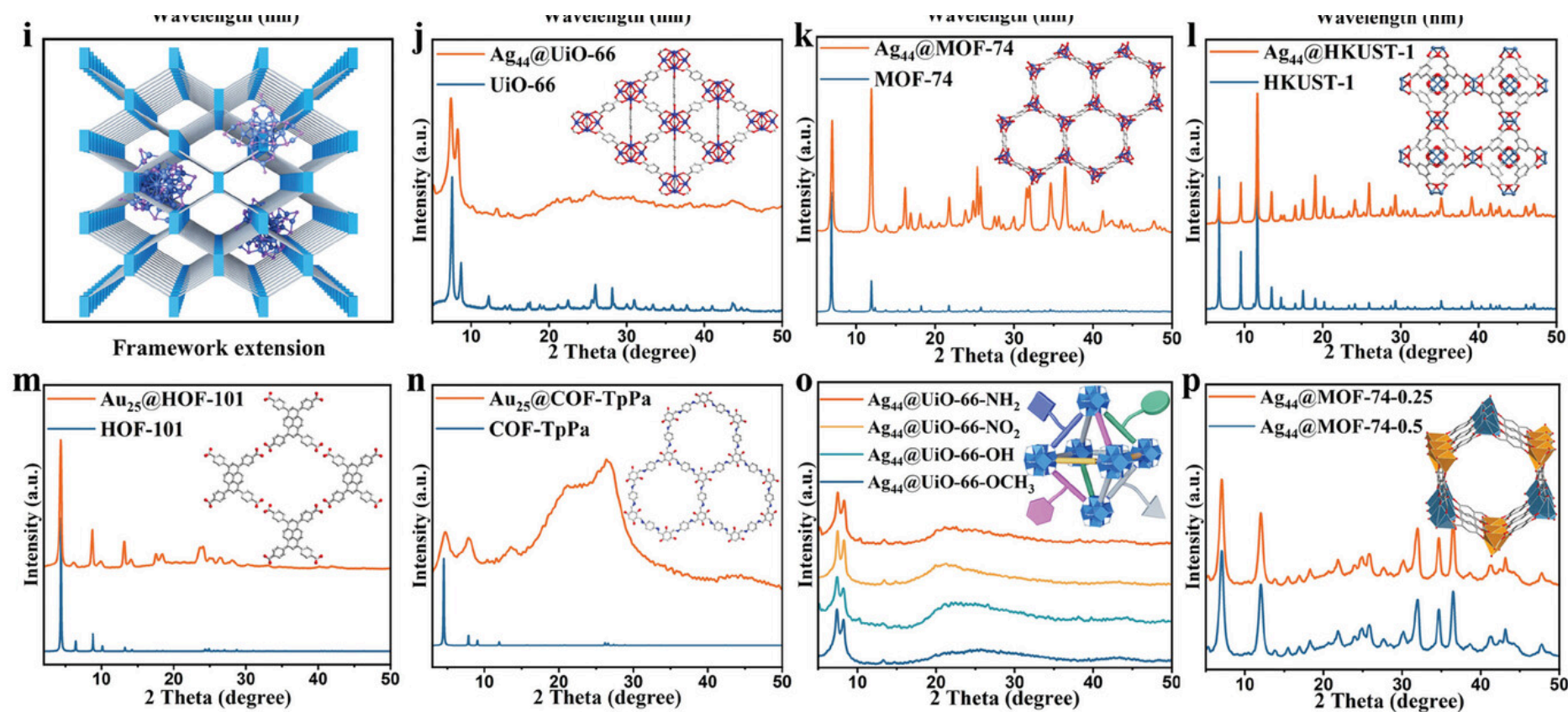


**Figure 3.** Enhanced stability of  $\text{Ag}_{44}@\text{ZIF-8}$ . a) Calculated loading efficiency and the wash experiments of  $\text{Ag}_{44}@\text{ZIF-8}$  and  $\text{Ag}_{44}\text{-on-ZIF-8}$ . HAADF-STEM images of  $\text{Ag}_{44}@\text{ZIF-8}$  b) before and c) after thermal treatment, and  $\text{Ag}_{44}\text{-on-ZIF-8}$  d) before and e) after thermal treatment. UV-vis spectra and digital images of free  $\text{Ag}_{44}$  and  $\text{Ag}_{44}@\text{ZIF-8}$  before (light blue and pale yellow) and after (dark blue and orange) f) coordinating solvent removal, g) ionic liquid treatment, h) 1 m  $\text{NH}_4\text{OH}$  treatment, and i) boiling DMF treatment.



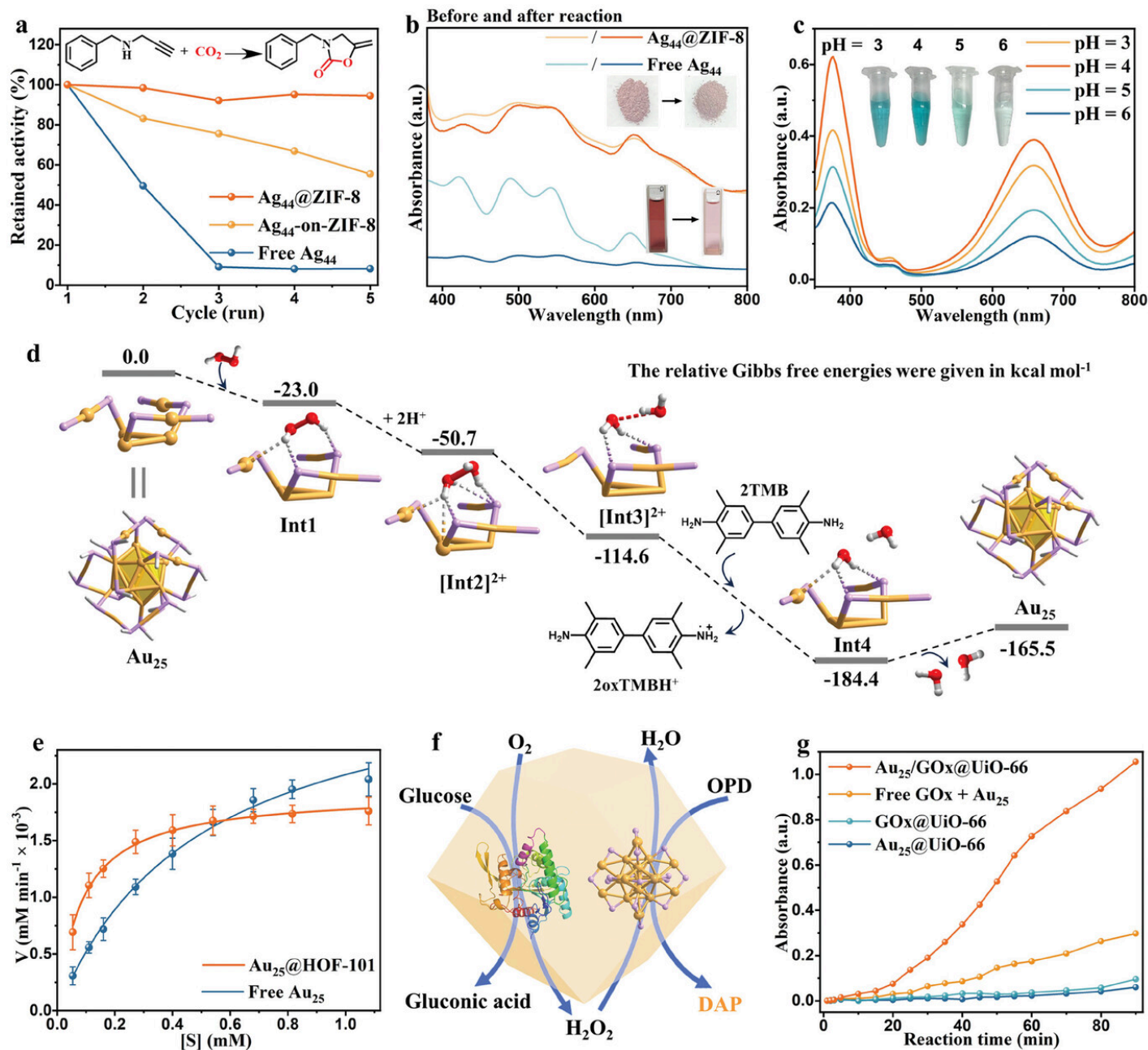
**Figure 4.** APNC extensions. a-h) UV-vis spectra of the composites of ZIF-8 with  $[\text{Au}_4\text{Cu}_6(\text{Dppm})_2(\text{SAdm})_4\text{Cl}_3]^+$  (b),  $[\text{Ag}_3\text{Cu}_2(\text{SPhMe}_2)_6]^-$  (c),  $\text{Au}_{25}(\text{L-Cys})_{18}$  (d),  $\text{Pd}_8(\text{C}_4\text{H}_3\text{O}_4\text{S})_8$  (e),  $\text{Ag}_4\text{Ni}_2(\text{DMSA})_4$  (f),  $[\text{Pd}_3(\text{C}_7\text{H}_7)_2\text{Cl}_3]^-$  (g), and  $[\text{Pd}_3\text{Cl}(\text{PPh}_2)_2(\text{PPh}_3)_3]^+$  (h).

# Reticular framework extensions



**Figure 4.** Reticular framework extensions. i-p) PXRD patterns of the composites of APNCs with UiO-66 (j), MOF-74 (k), HKUST-1 (l), HOF-101 (m), COF-TpPa (n), UiO-66-R (R =  $\text{NH}_2$ ,  $\text{NO}_2$ ,  $\text{OH}$ ,  $\text{OCH}_3$ , o), and MOF-74-X (X = 0.25 and 0.5, p).

# Catalytic performance of selected materials



# Conclusion

- 21 APNCs@frameworks were successfully synthesized by using mechanochemical synthesis.
- This solid-state and room-temperature approach, overcomes compatibility issues associated with framework synthesis conditions, thereby preserving the structural integrity of APNCs during encapsulation.
- The resulting nanocomposites exhibit significantly improved stability and demonstrate a 315-fold increase in reactivity compared to free APNCs.
- This method expands the diversity of both APNCs and reticular frameworks, enabling the design of customized nanocomposites and multifunctional systems for specific applications.
- This diversity allows for the creation of tailormade nanocomposites capable of targeting precise functions such as HRP-mimicking catalysis and facilitates the construction of biocatalytic cascades with multiple catalytic components.