

# Accurate Thermal Resection of Atomically Precise Copper Clusters to Achieve Near-IR Light-Driven CO<sub>2</sub> Reduction

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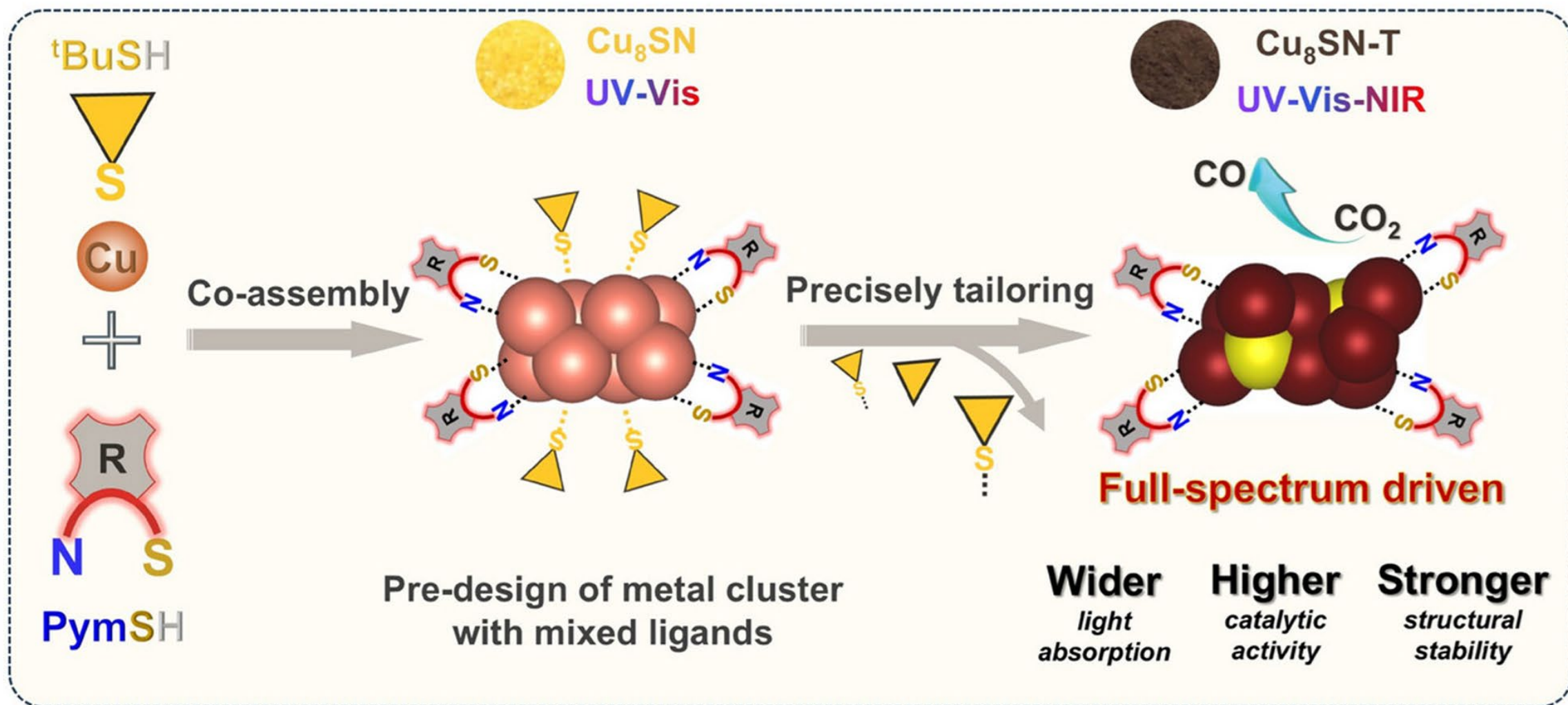
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Paper presentation

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10<sup>th</sup> May, 2025

# Abstract



# Background

## Nanoscale

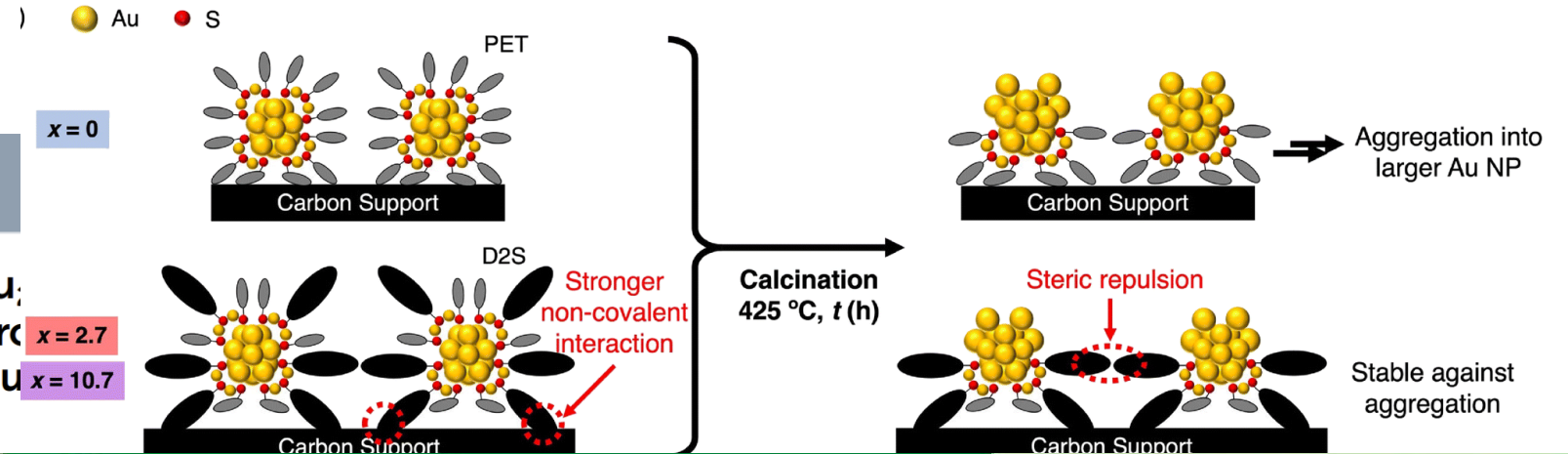
### PAPER



Cite this: *Nanoscale*, 2024, **16**, 20608

Carbon-supported Au decorated with dendronized ligands for CO<sub>2</sub> photoreduction reaction†

Kosuke Sakamoto, Shinya Masuda,



Angewandte  
International Edition  
Chemie

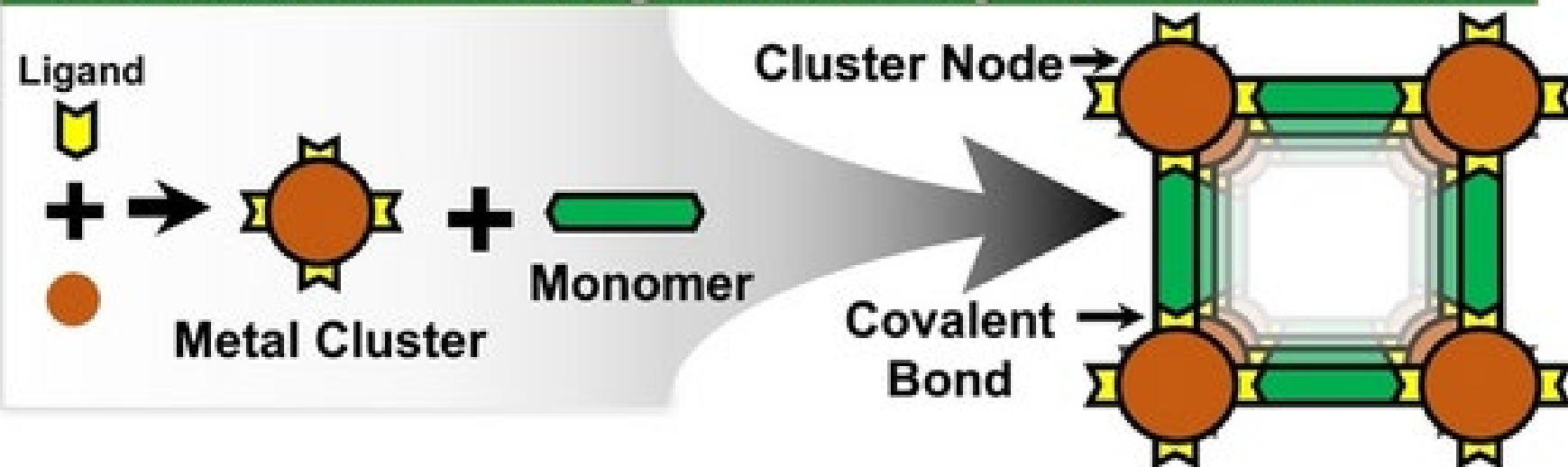
Research Article Angewandte Chemie International Edition

## 3D Cluster-Based Covalent Organic Frameworks for CO<sub>2</sub> Photoreduction

Yue Xu, Jian-Peng Dong, Le Wang, Rong-Li Geng, Rui Wang, Thomas C. W. Mak

First published: 18 February 2025 | <https://doi.org/10.1002/anie.202500000>

## Covalent Linkage Assembly 3D Cluster-COF



# Motivation

- Increasing CO<sub>2</sub> percentage in atmosphere became threat to human civilization, CO<sub>2</sub> reduction is the most reliable approach to solve this problem.
- Using solar energy to get rid off from global warming is one of the dream of scientists.
- Uses of metal nanocluster's potential in such noble work is inspiring to all the researcher working with nanocluster.



# Why this paper?

- This article provides a comprehensive analytical essence of how selective ligand removal change the properties of NC drastically and enhance cluster's catalytic properties.
- This article widen the window of catalysis using nanocluster which enhance the relevance of nanocluster.

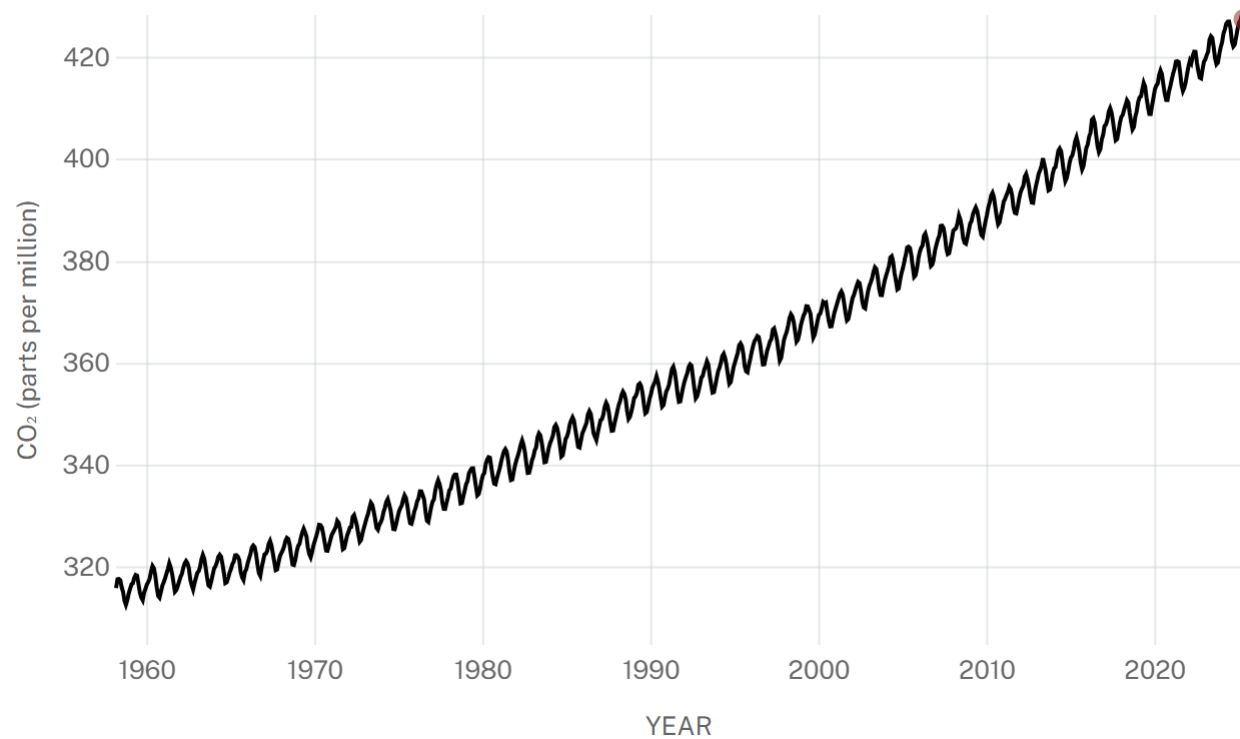




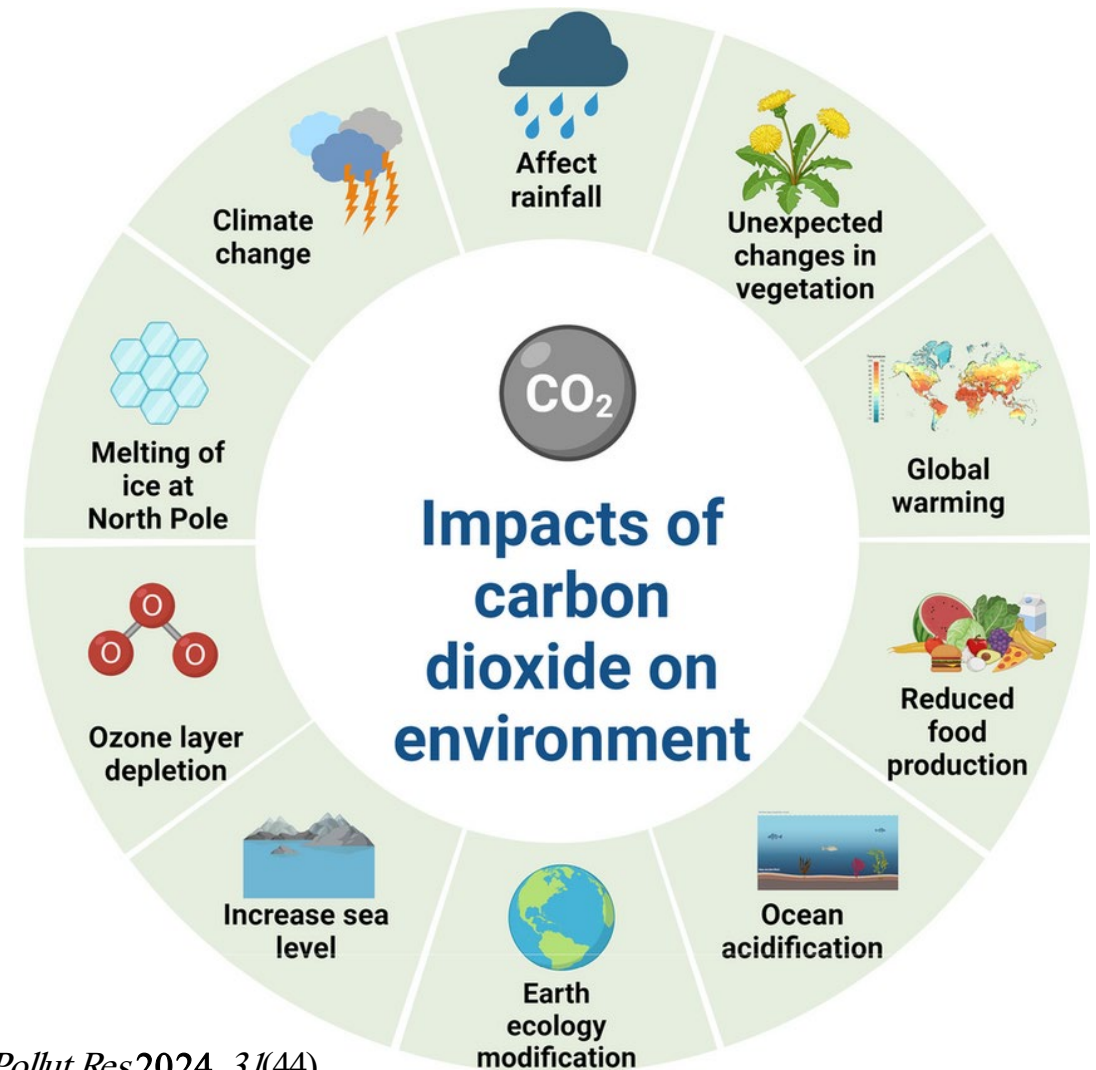
# Introduction

## DIRECT MEASUREMENTS: 1958-PRESENT

Data source: NOAA, measured at the Mauna Loa Observatory



Ravichandran, M; et. al. *Environ Sci Pollut Res* 2024, 31(44), 55895–55916.



# Introduction

What is Nanocluster?

**Nanoclusters** are atomically precise, crystalline materials most often existing on the 0-2 nanometer scale

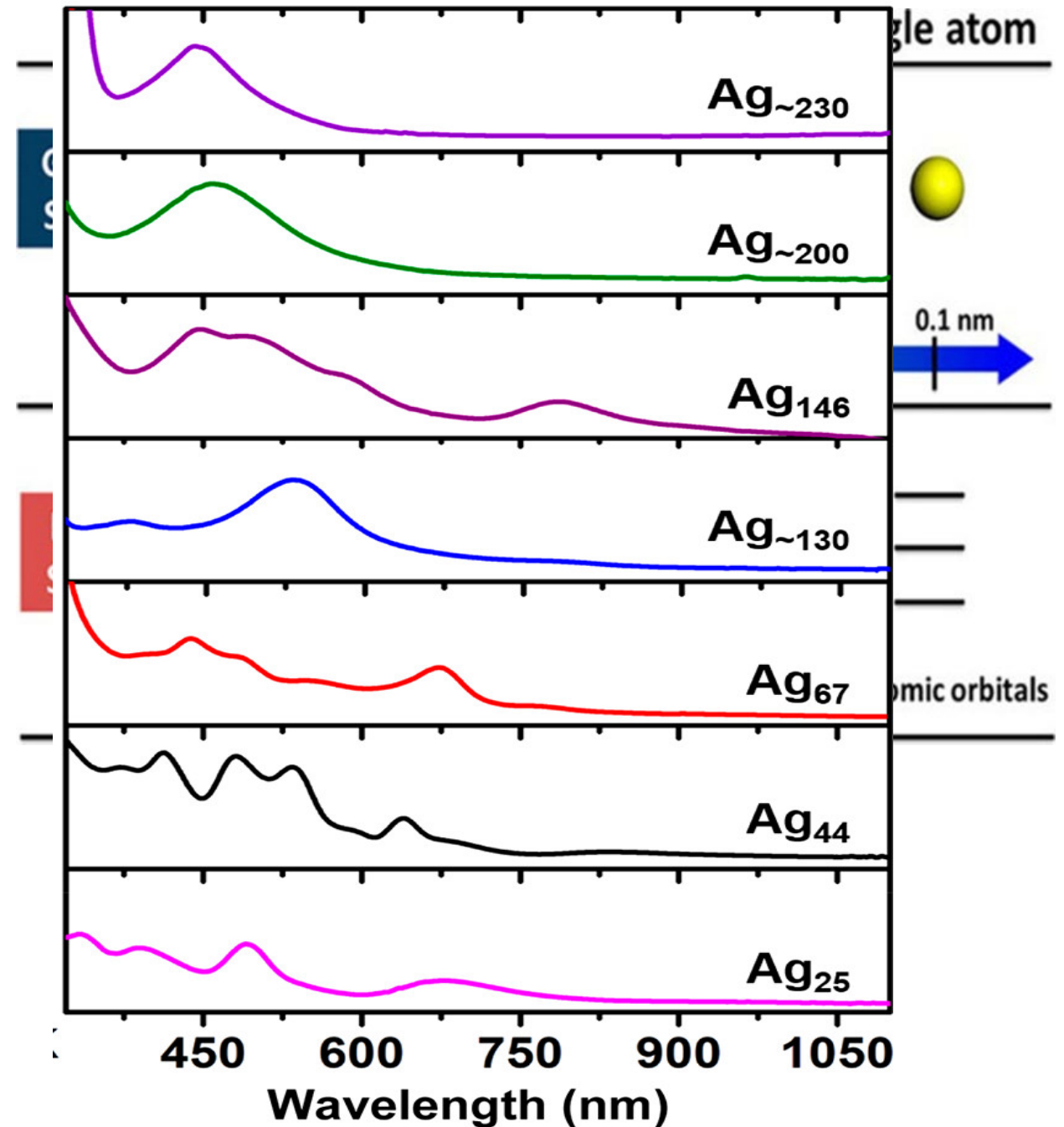
Unique optical properties

Atomically precise

Nanocluster

Discrete band gap

Chirality



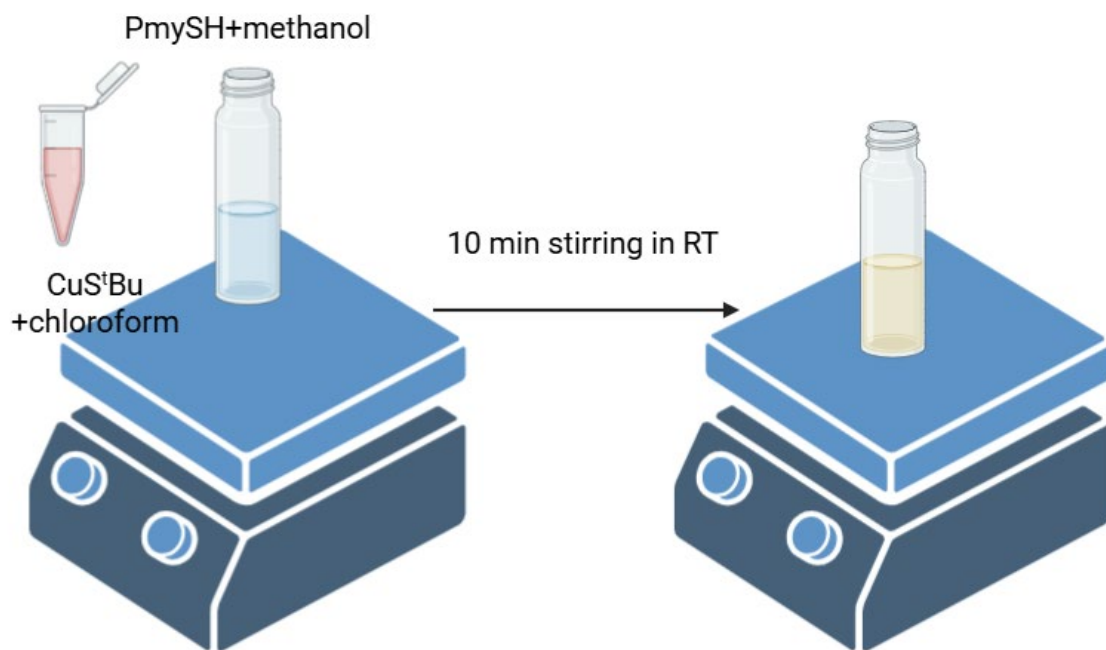
# Introduction

- Various types of materials namely metal organic framework (MOF), g-CN, metal nitrides, metal sulfides etc. have been used to reduce CO<sub>2</sub> but the efficiency is lower than the threshold over which practically it can be implemented.
- The limiting reason for low efficiency is their adsorption is limited to the certain range of wavelength of solar spectrum.
- Nanoclusters are one of the materials among those which has drawn maximum attention.
- Ligand makes clusters stable but less active due to non-accessibility of metal atom.
- Selective removal of ligands can prohibit the aggregation of cluster and enhance the catalytic efficiency by making metal atom accessible.

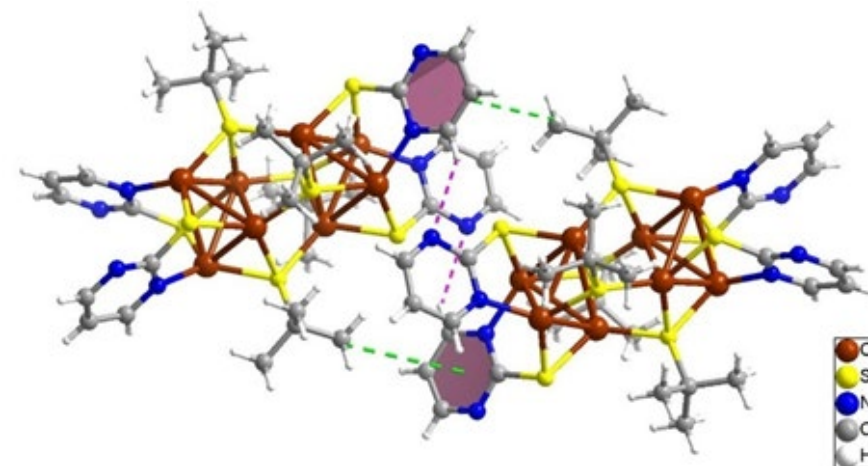


# Results and discussion

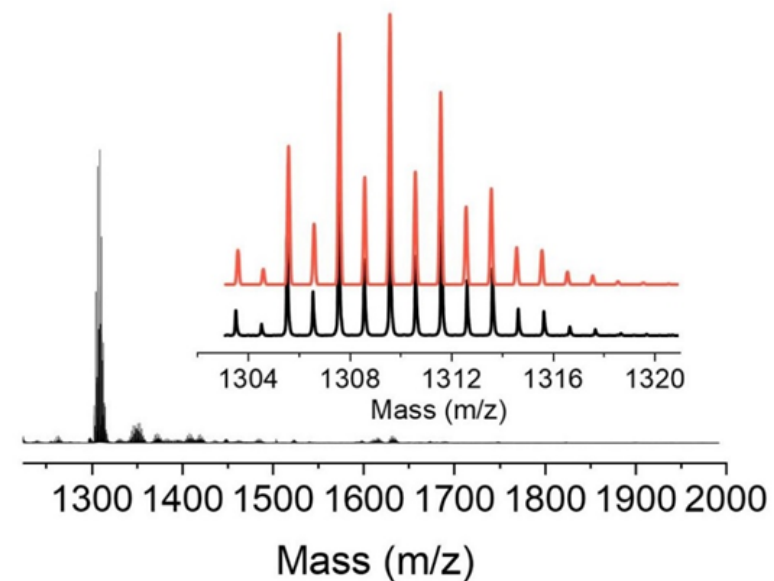
## Synthesis and characterization of $\text{Cu}_8(\text{S}^t\text{Bu})_4(\text{PymS})_4[\text{Cu}_8\text{SN}]$



SC-XRD driven structure

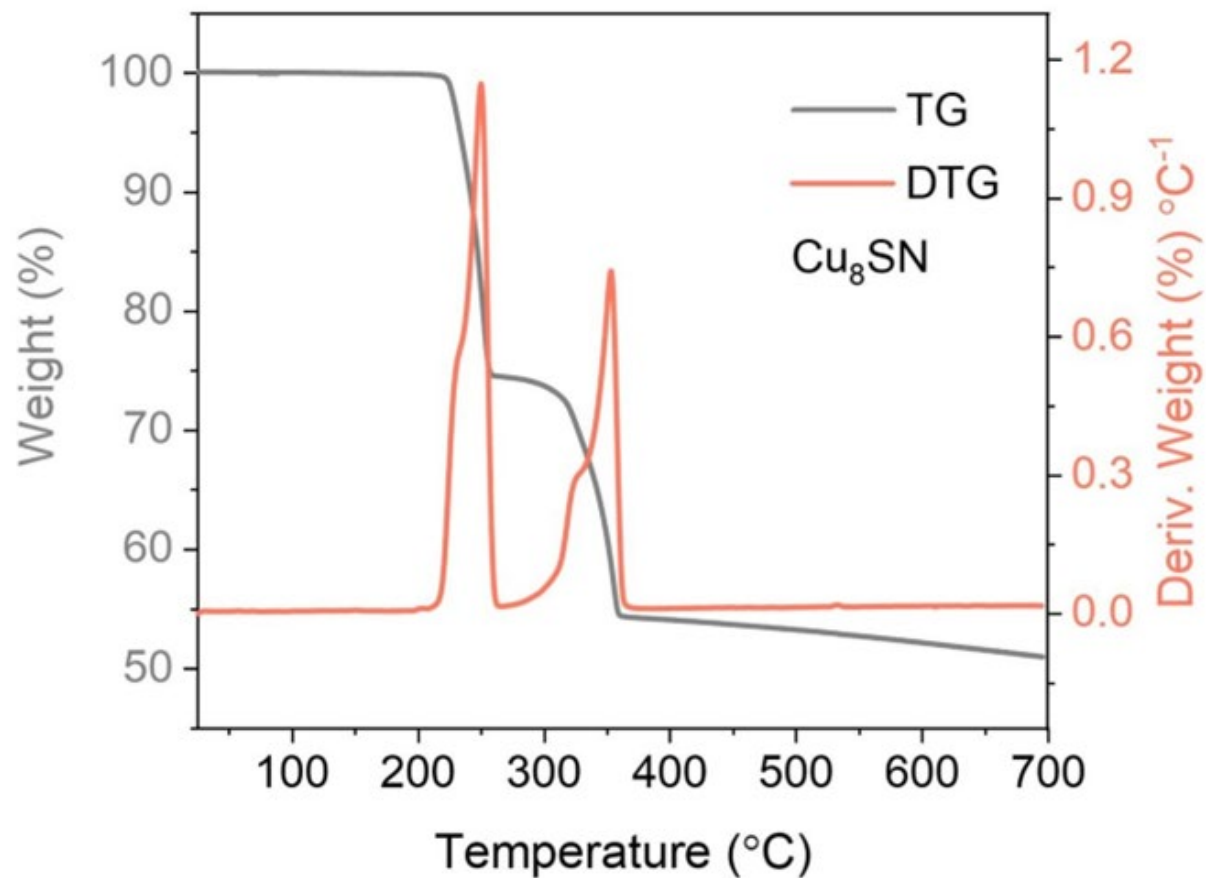


ESI-MS data



# Results and discussion

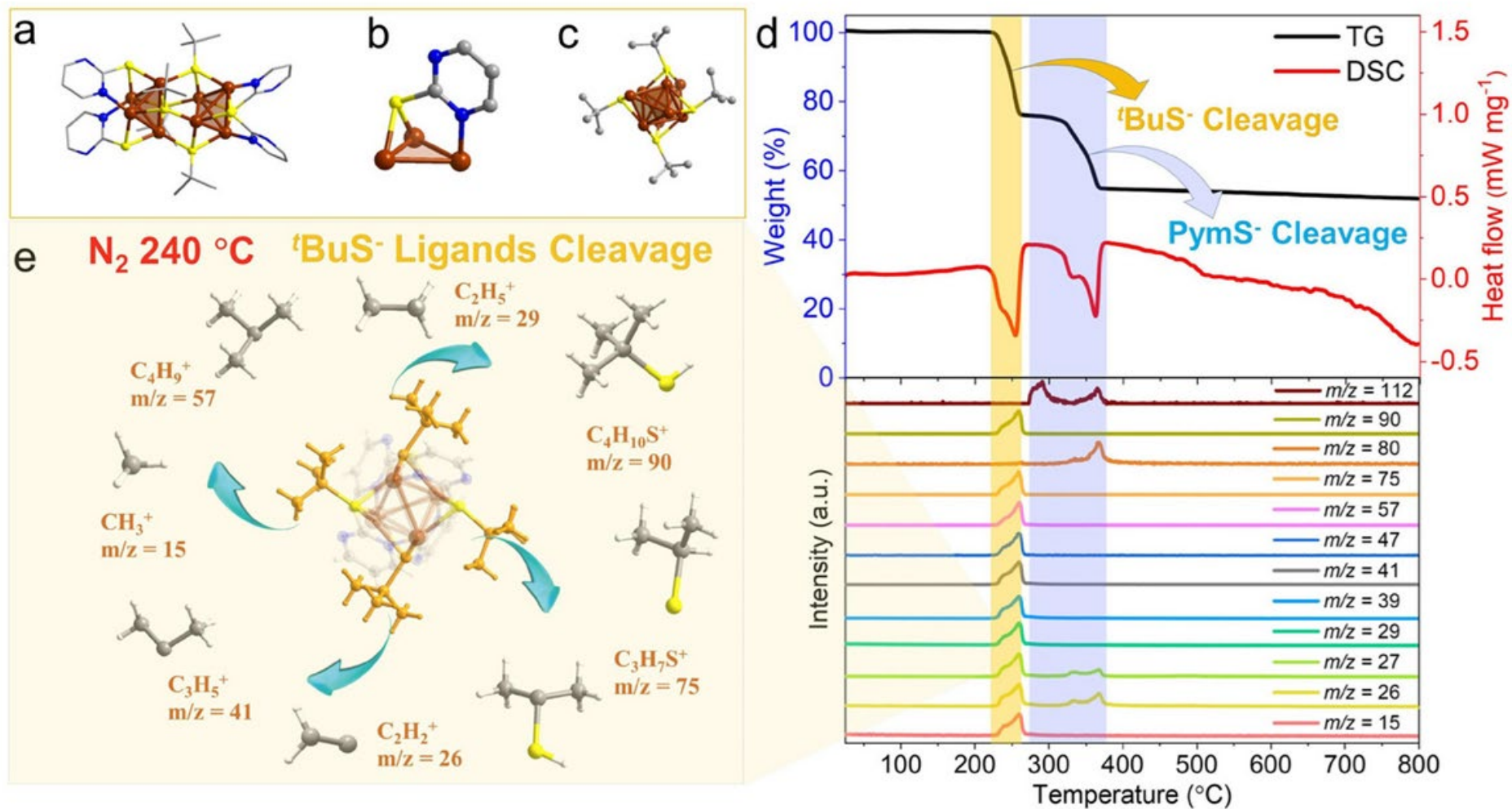
TGA analysis



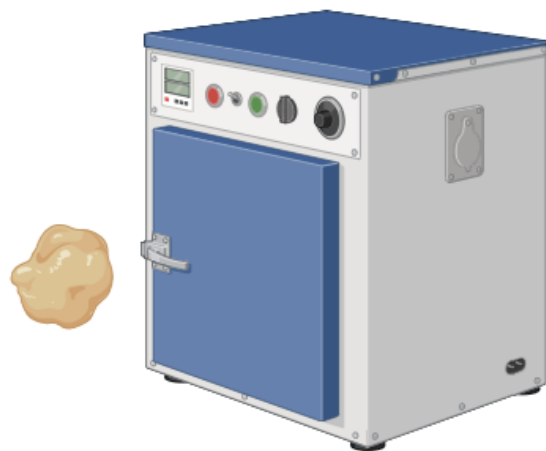
Component	Formula Count	Mass (g/ mol)	Theoretical Mass %
Cu	8	508.40	38.69%
S <sup>t</sup> Bu	4	356.68	27.15%
PymS	4	448.60	34.16%
Total		1313.68	100%

# Results and discussion

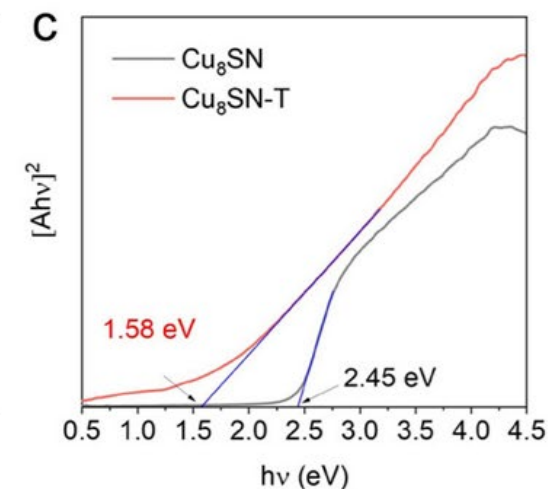
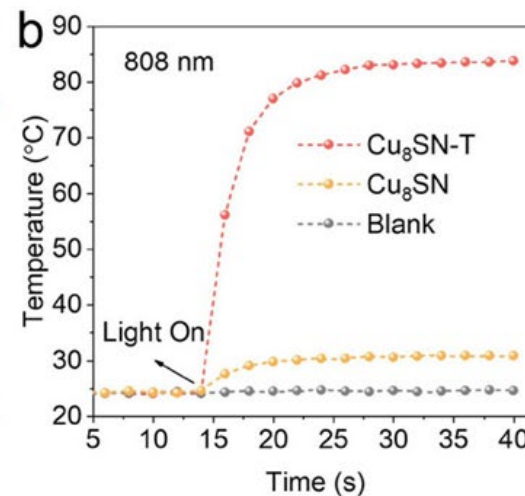
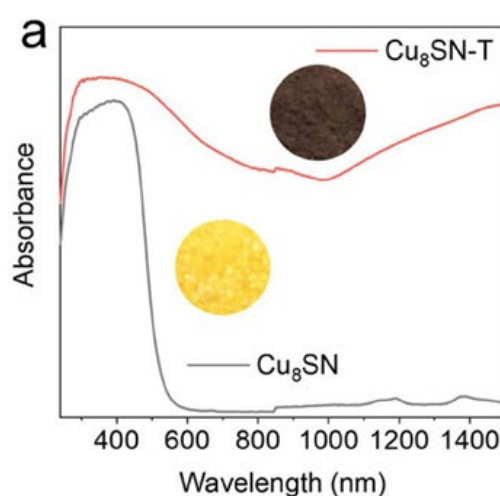
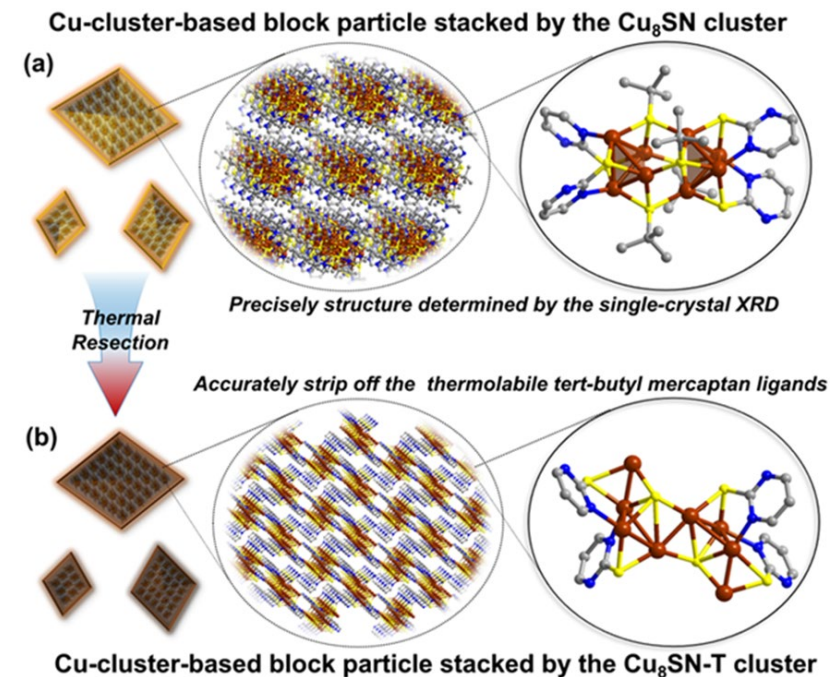
TG-MS data



# Results and discussion



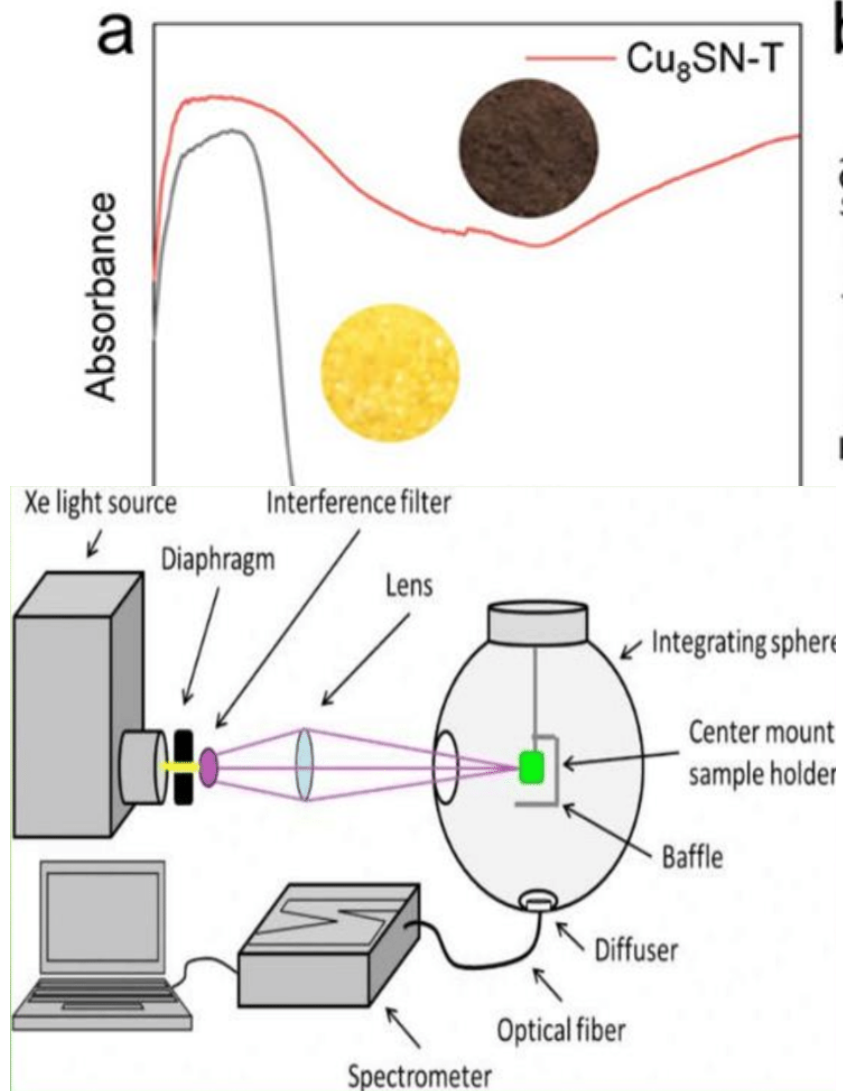
240° C  
N<sub>2</sub> environment for 1 h



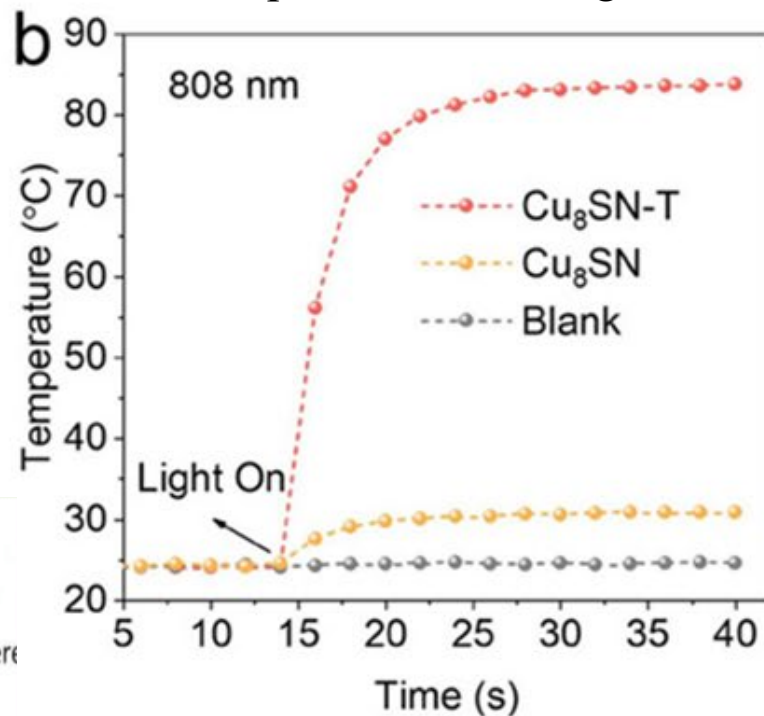


# Results and discussion

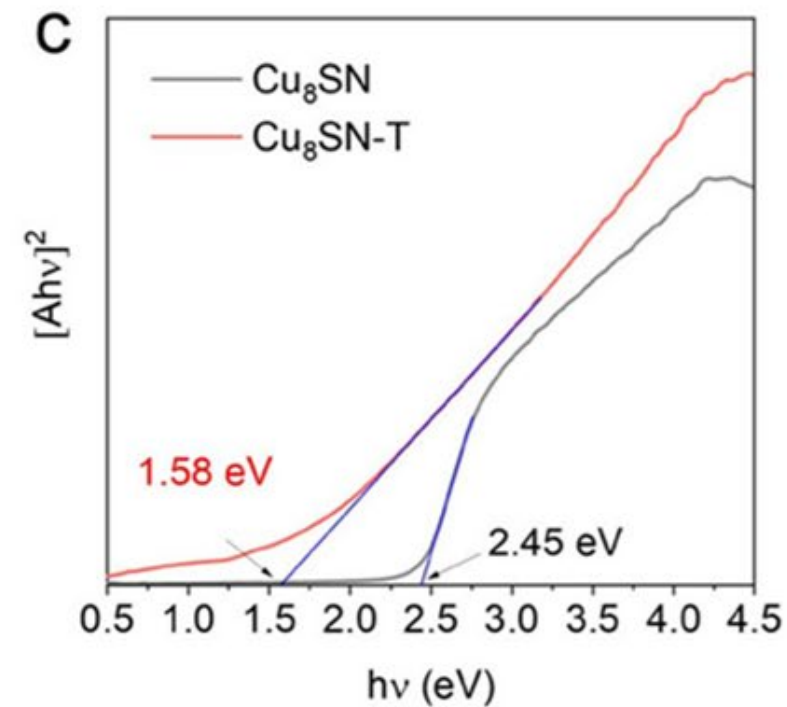
UV-Vis-NIR DRS data



IR photothermal image



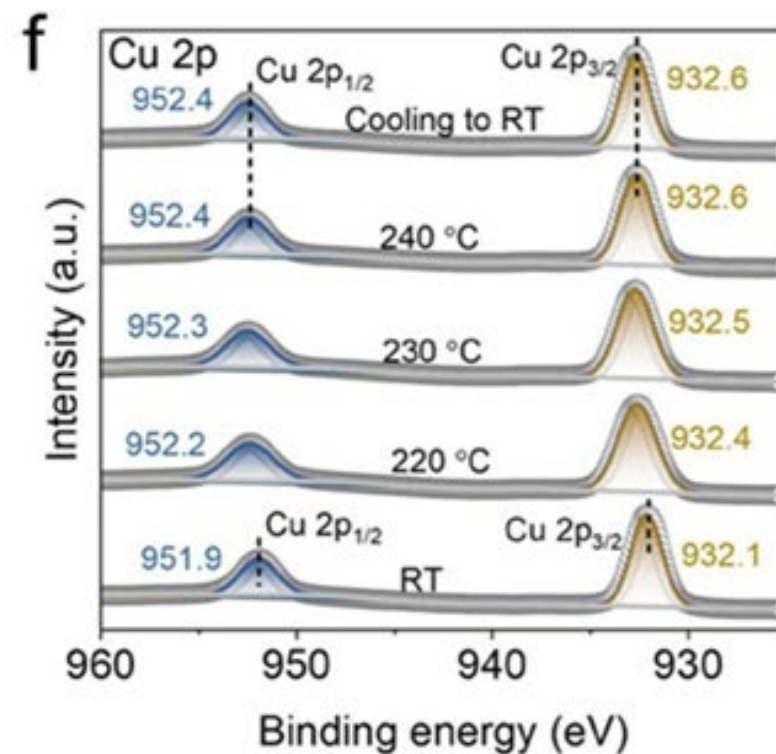
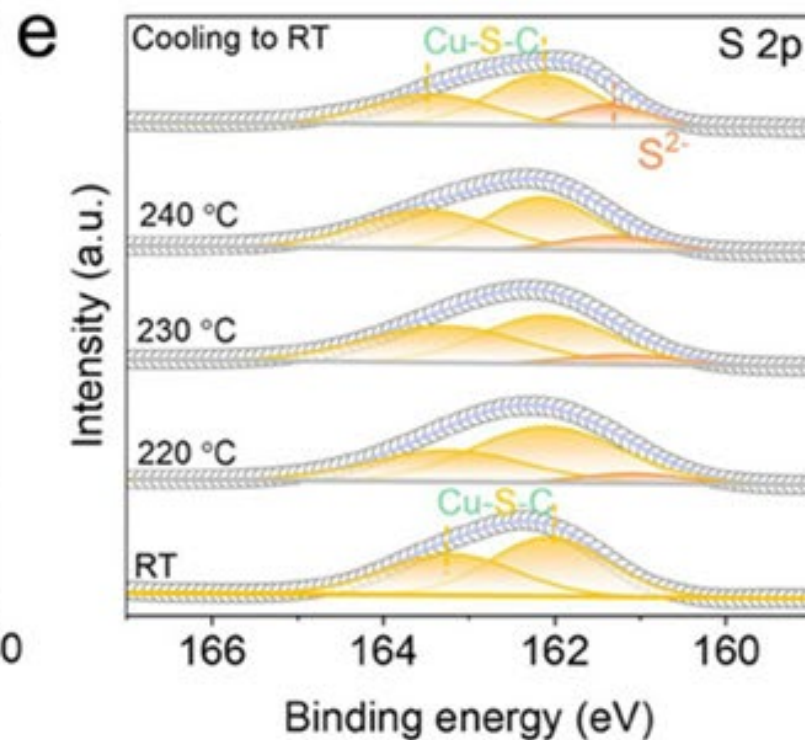
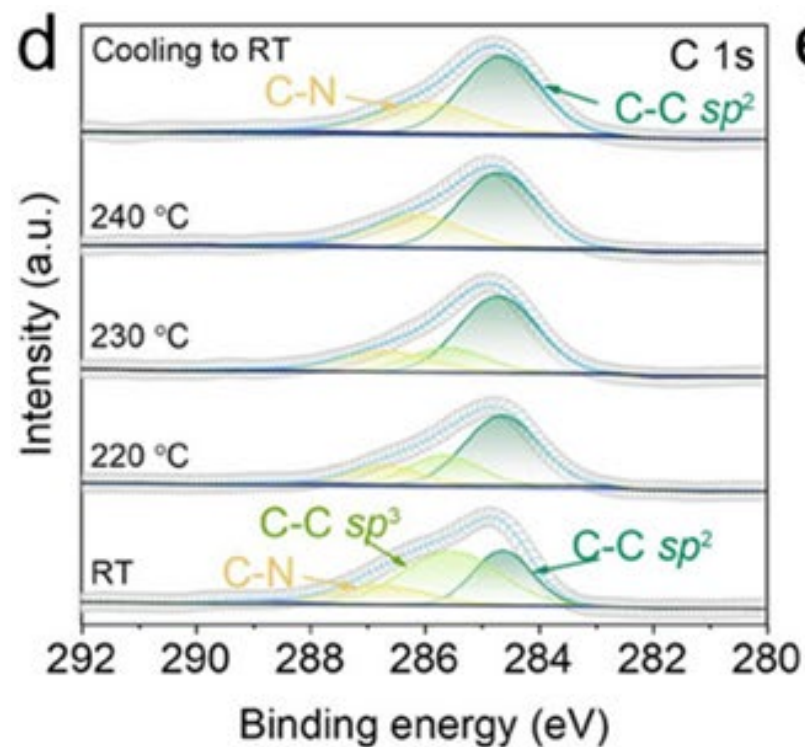
Tauc plot



$$(\alpha h\nu)^{1/n} = A(h\nu - E_g)$$

# Results and discussion

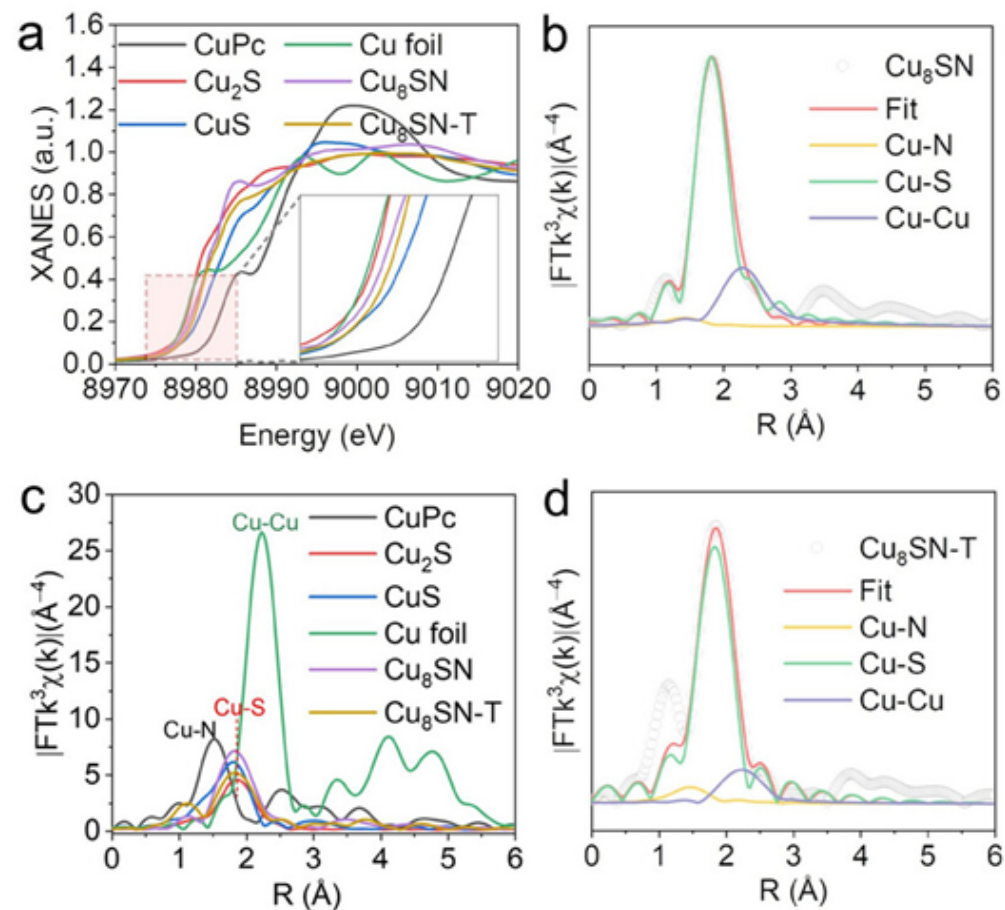
In situ XPS data during thermal resection





# Results and discussion

## XAFS data

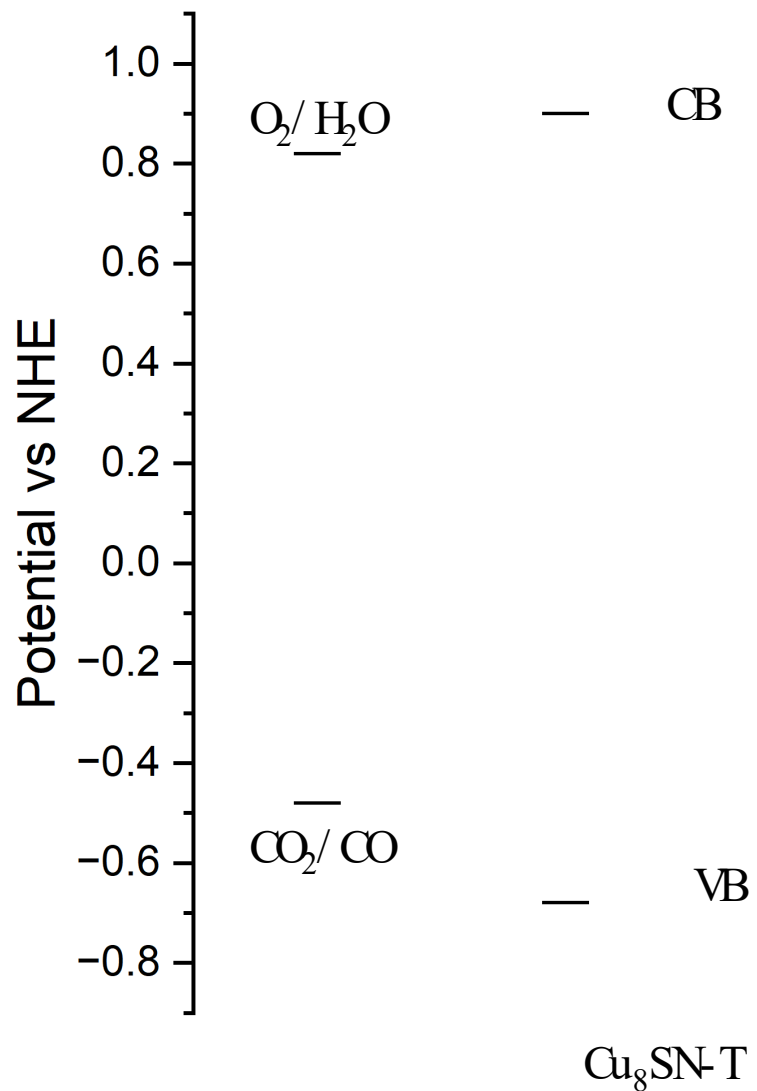


## Findings

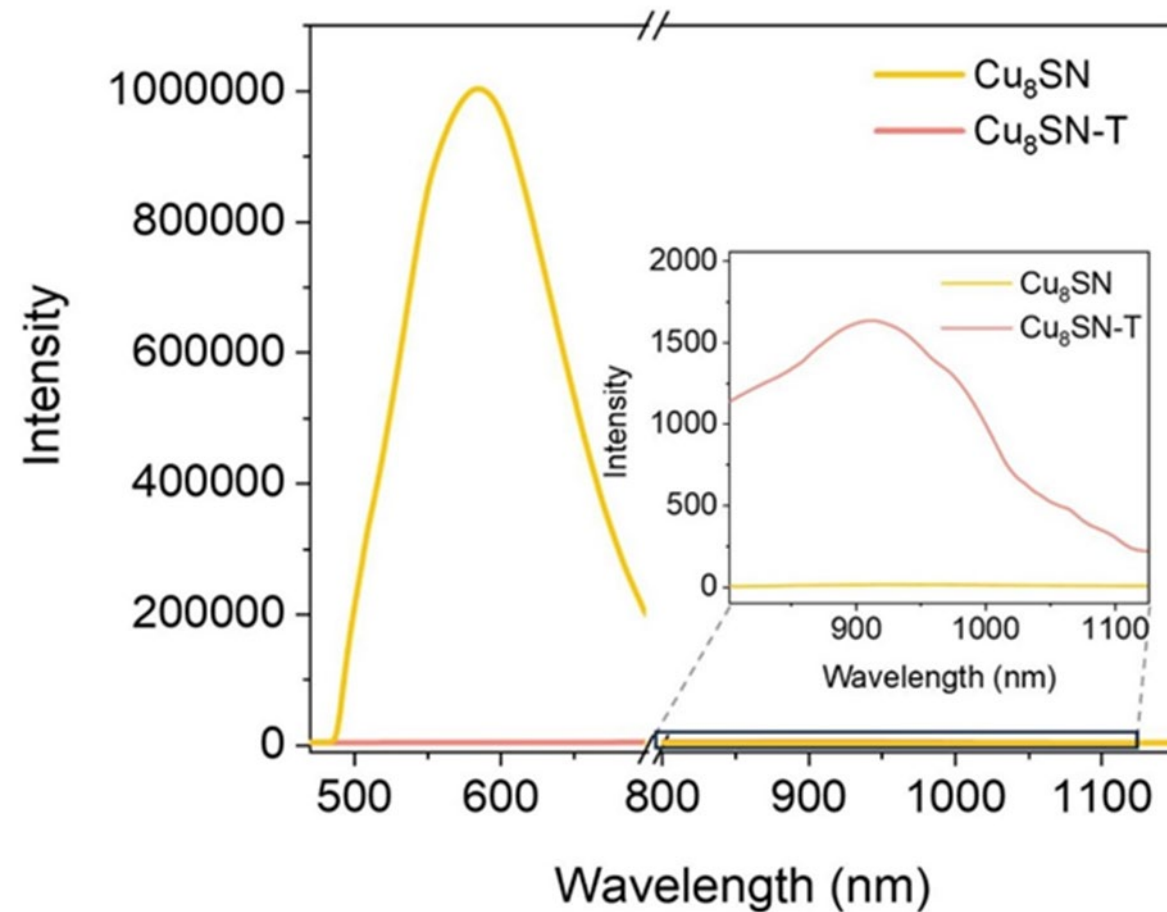
- For  $\text{Cu}_8\text{SN}$  the absorption edge is in between Cu foil and  $\text{Cu}_2\text{S}$  and closer to the  $\text{Cu}_2\text{S}$  revealing monovalent nature of Cu.
- EXAFS study reveals the ratio of Cu-N, Cu-S and Cu-Cu is 1:5:1 which support the predicted formula of  $\text{Cu}_8\text{SN}$ .
- And for  $\text{Cu}_8\text{SN-T}$  the ratio of Cu-N, Cu-S and Cu-Cu environment is 4:18:3 which supports the predicted formula for  $\text{Cu}_8\text{SN-T}$  too.

# Results and discussion

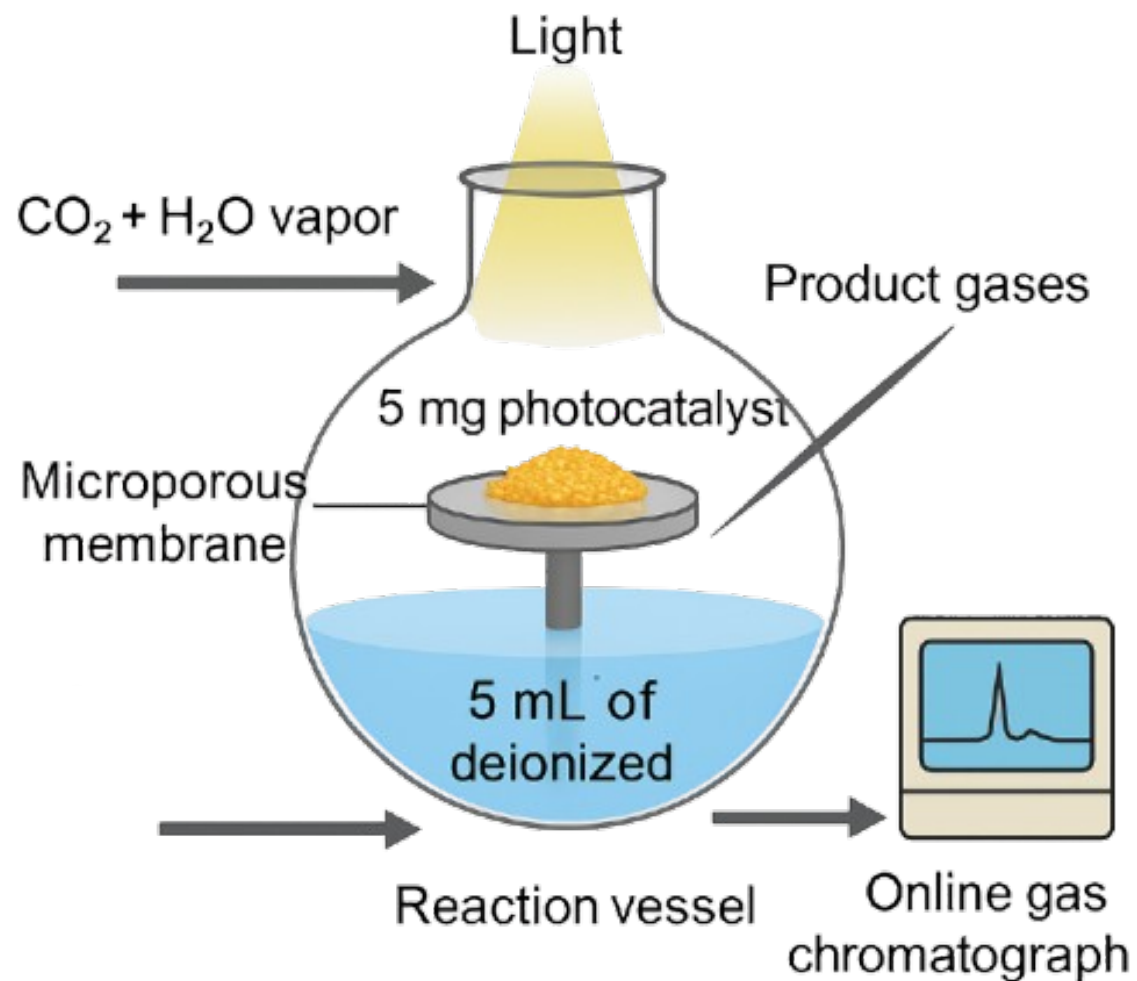
Band structure comparison



PL study



# Results and discussion

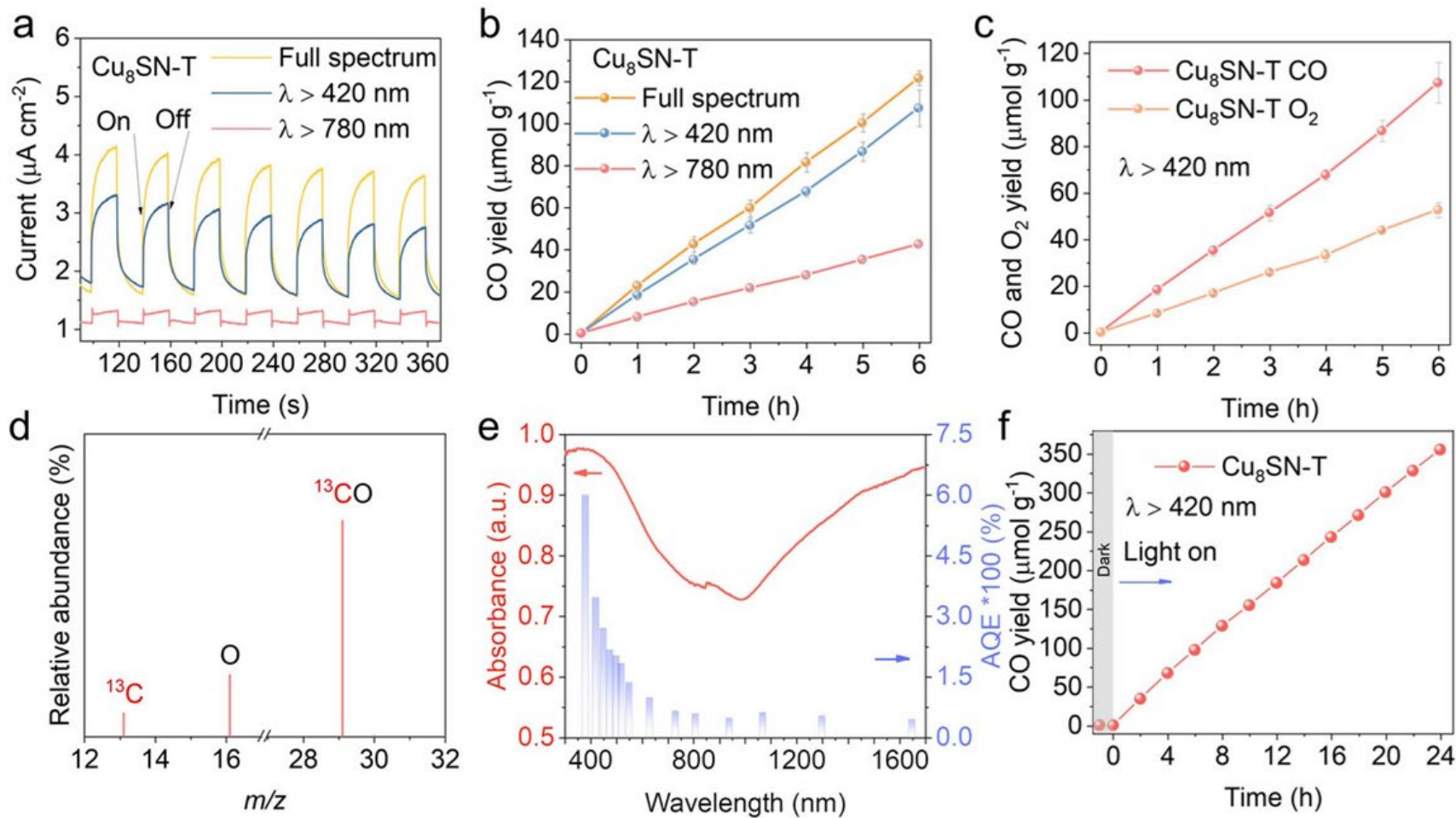


Labsolar-6A



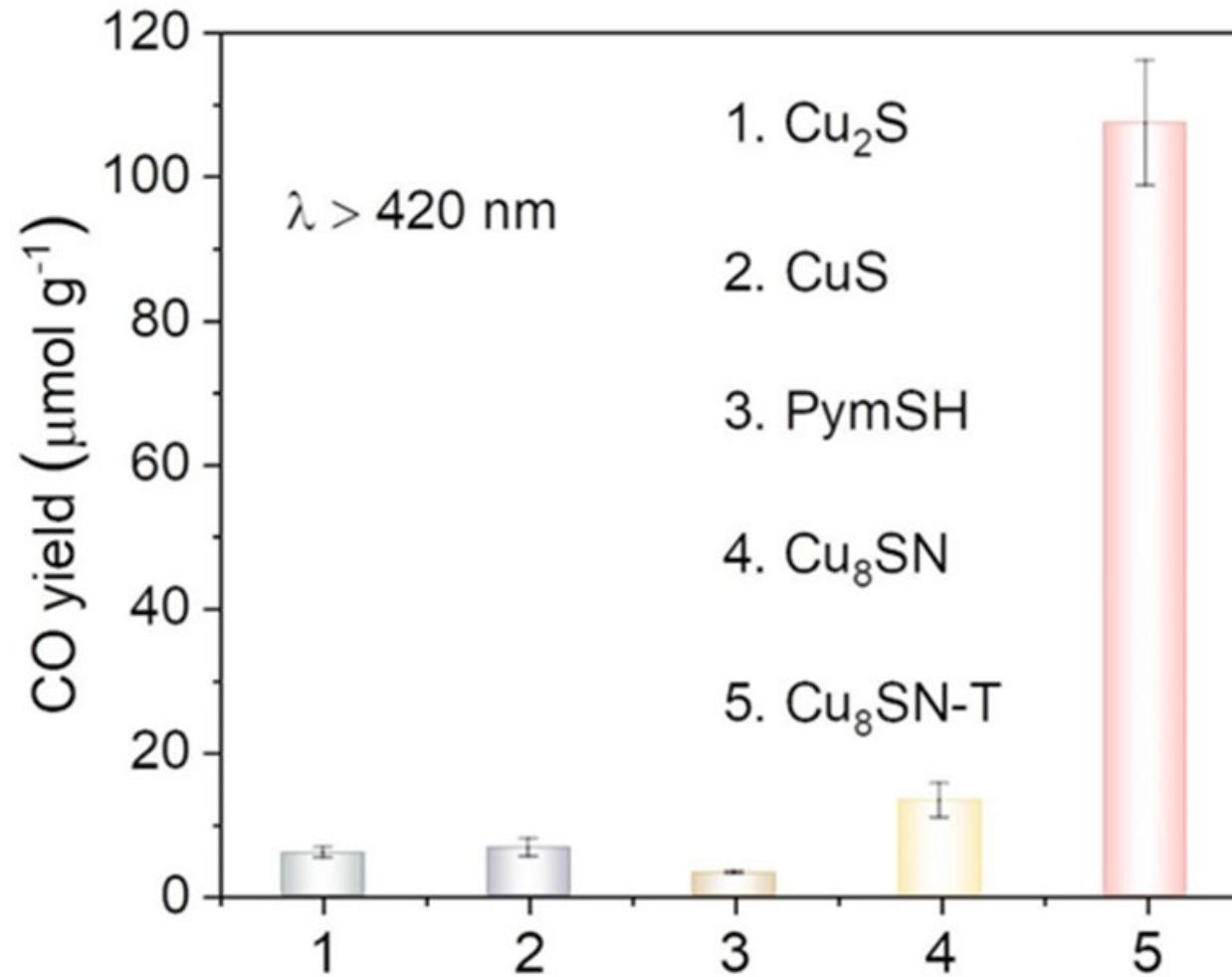
# Results and discussion

## Photoreduction data



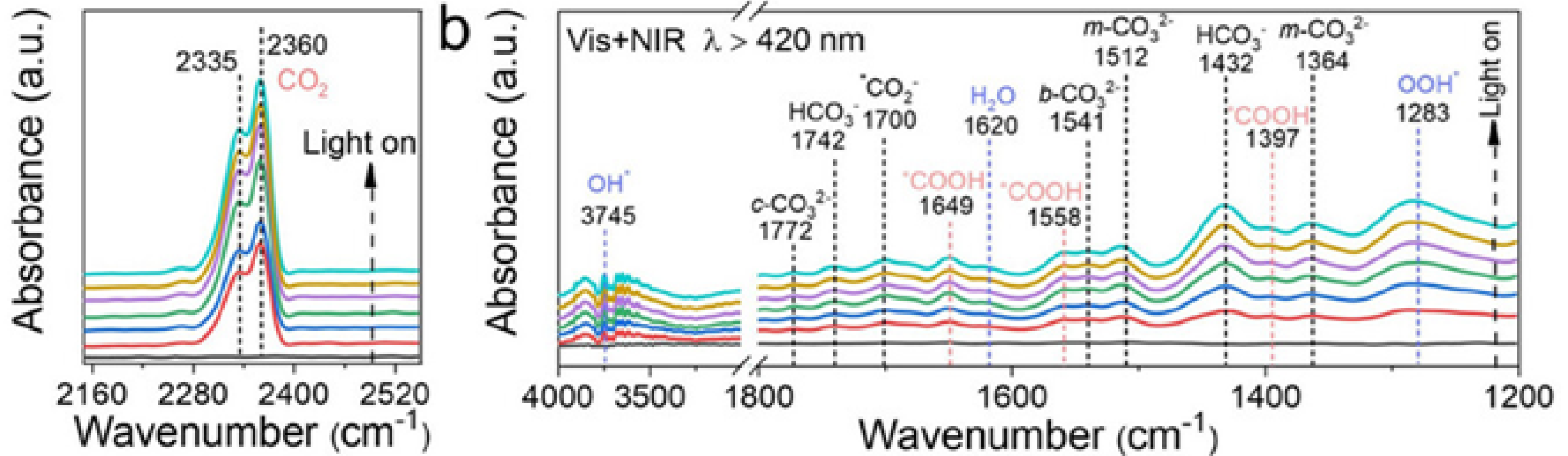
# Results and discussion

Efficiency comparison



# Results and discussion

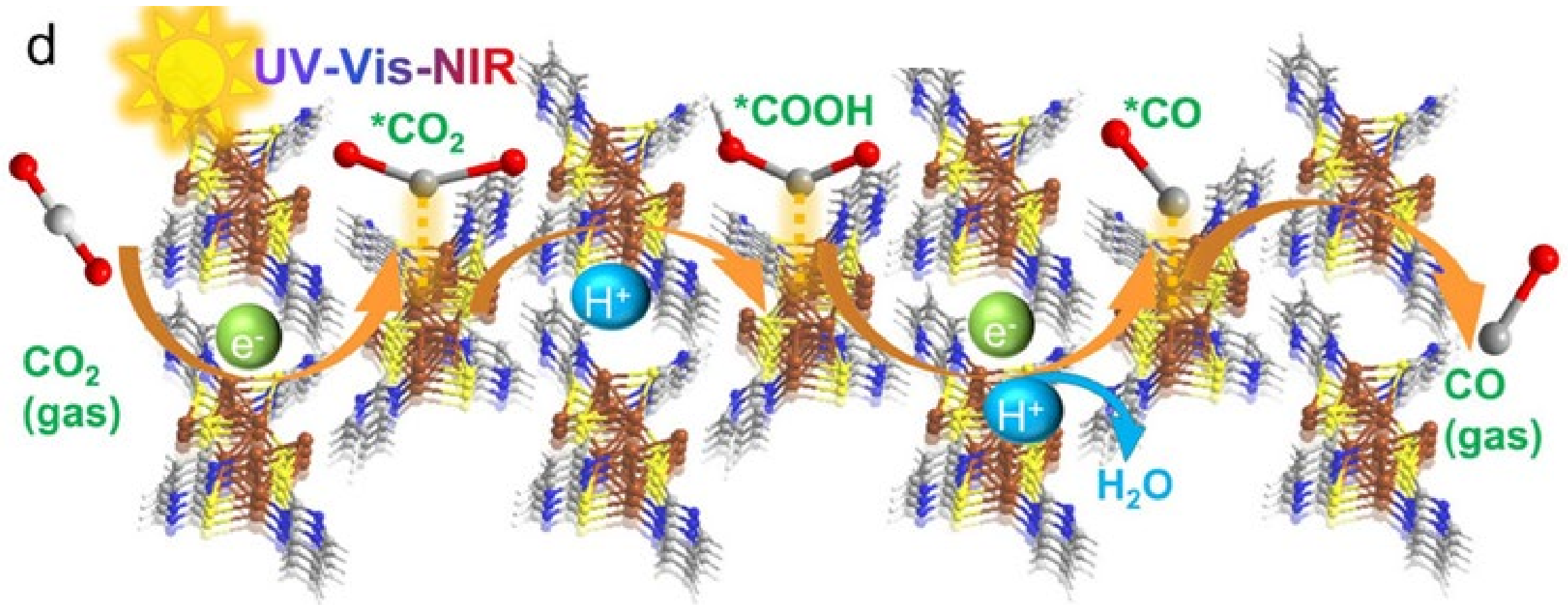
In situ DRIFT spectra during photo reduction





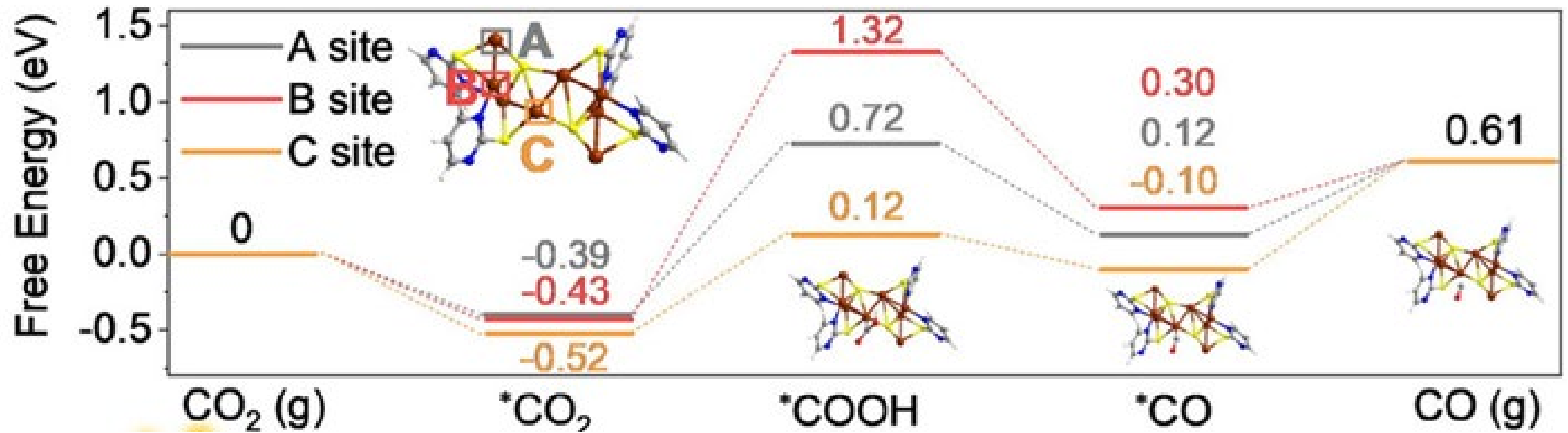
# Results and discussion

Theoretical proposed mechanism



# Results and discussion

Free energy profile diagram



# Conclusion

- This article provides an idea how selective ligand resection can enhance the catalytic activity.
- Ligand removal broadens the range of absorption wavelength starting from UV to NIR.
- Complete study of ligand removal has been done using various experimental techniques.
- Due to specific band position this material shows almost 100% selectivity for CO production.