

Supporting information for the paper:

**Hemoprotein bioconjugates of Au and Ag nanoparticles and Au nanorods:
structure-function correlations**

Renjis T. Tom, A. K. Samal, T. S. Sreeprasad, and T. Pradeep*

DST Unit on Nanoscience

Department of Chemistry and Sophisticated Analytical Instrument Facility

Indian Institute of Technology Madras

Chennai - 600 036, India

* For correspondence, Email: pradeep@iitm.ac.in Fax: 044-2257 0545/0509

Supporting Information 1

Calculation of coverage

We illustrate the calculation with the example of Hb.

$$\begin{aligned} \text{(a) Cross sectional area of Hb molecule, circle with diameter 5.5 nm} &= \pi r^2 \\ &= 3.14 \times (2.75 \text{ nm})^2 \\ &= 23.75 \text{ nm}^2 \end{aligned}$$

$$\begin{aligned} \text{(b) Cross sectional area of Hb molecule ellipse with diameters 5.5 nm and 7 nm} &= \pi r_1 r_2 \\ &= 3.14 \times (2.75 \text{ nm} \times 3.5 \text{ nm}) = 30.22 \text{ nm}^2 \end{aligned}$$

$$\text{Average cross sectional area from (a) and (b)} = 27 \text{ nm}^2$$

In order to get the layer thickness of an adsorbed layer, we calculate the average diameter of the molecule. This is necessary as the adsorption can occur the horizontal or perpendicular direction (note that the molecule is an ellipsoid).

$$\text{Average diameter of the molecule} = [5.5+5.5+7.0]/3 \text{ nm} = 6 \text{ nm}$$

$$\begin{aligned} \text{Surface area of Ag nanoparticle with diameter 60 nm} &= 4\pi r^2 \\ &= 4 \times 3.14 \times (30 \text{ nm})^2 \\ &= 11304 \text{ nm}^2 \end{aligned}$$

$$\begin{aligned} \text{Maximum coverage on first layer} &= 11304 \text{ nm}^2 / 27 \text{ nm}^2 \\ &= 419 \end{aligned}$$

$$\begin{aligned} \text{Surface area of second layer on Ag nanoparticle (d= [6 + 60 + 6] nm)} &= 4\pi r^2 \\ &= 4 \times 3.14 \times (36 \text{ nm})^2 \\ &= 16277.76 \text{ nm}^2 \end{aligned}$$

$$\begin{aligned} \text{Maximum coverage on second layer} &= 16277.76 \text{ nm}^2 / 27 \text{ nm}^2 \\ &= 603 \end{aligned}$$

Similarly, the number of molecules in the 3rd, 4th and 5th layers can be calculated as, 821, 1072 and 1357, respectively.

$$\begin{aligned} \text{Therefore, the number of molecules in 4 layers of Hb on Ag (60 nm) particle} \\ &= 419 + 603 + 821 + 1072 \\ &= 2915 \end{aligned}$$

The experimentally observed figure namely, 2858 is close to the calculated one and the difference can be due to reduced packing density.

Supporting Information 2

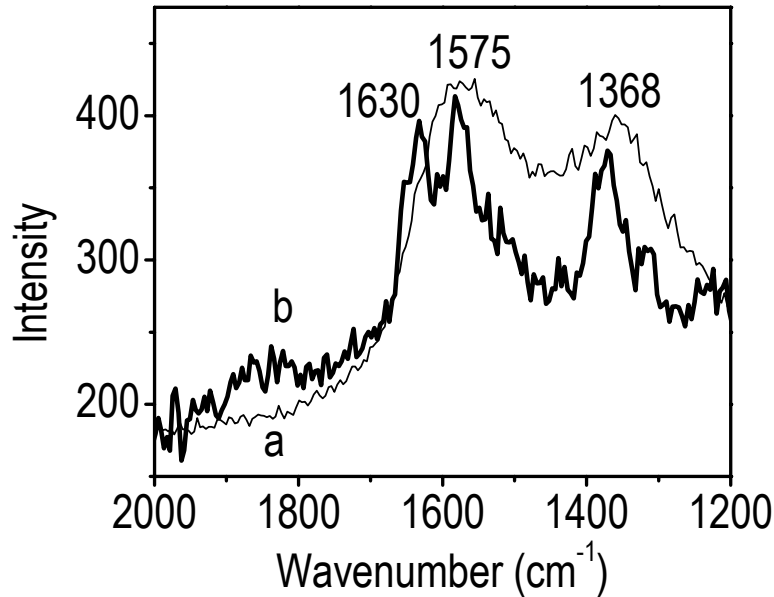


Figure 1. Raman spectra of Mb (a) and AuNR@Mb (b), recorded using 514.5 nm laser excitation through a confocal (60 X) objective.

Supporting Information 3

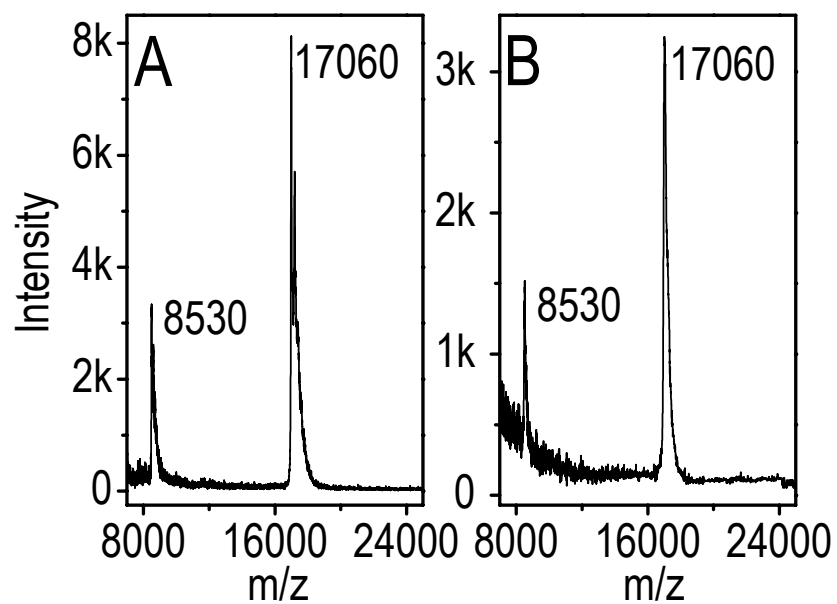


Figure 2. MALDI-TOF mass spectra of the samples measured with 337 nm N_2 laser using sinapinic acid matrix. The figures A and B correspond to Mb and AuNR@Mb, respectively.

Supporting Information 4

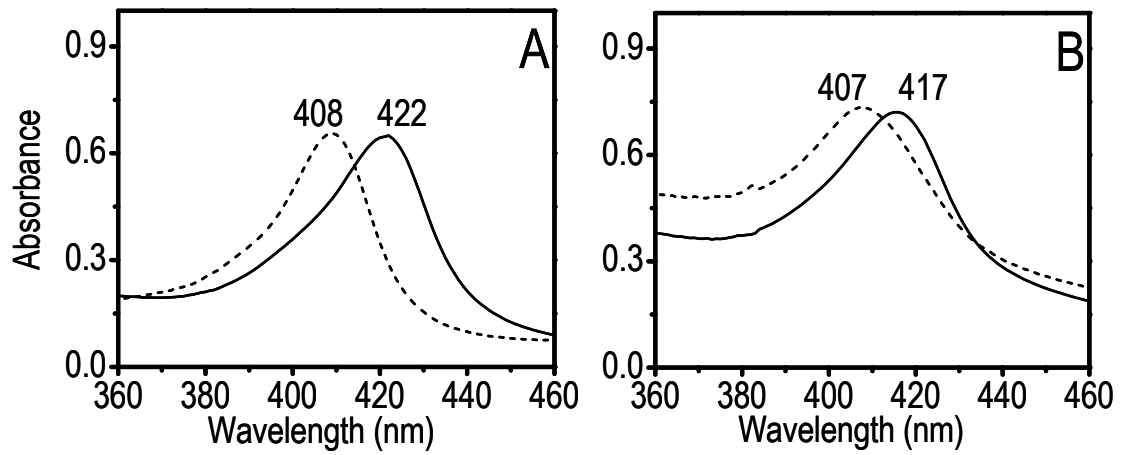


Figure 3. Absorption spectra of Mb (A) and Hb (B) in aqueous phosphate buffered saline. The solid line and dashed lines correspond to samples kept with 10 mM sodium azide for 24 hours and control sample without sodium azide, respectively.

Supporting Information 5

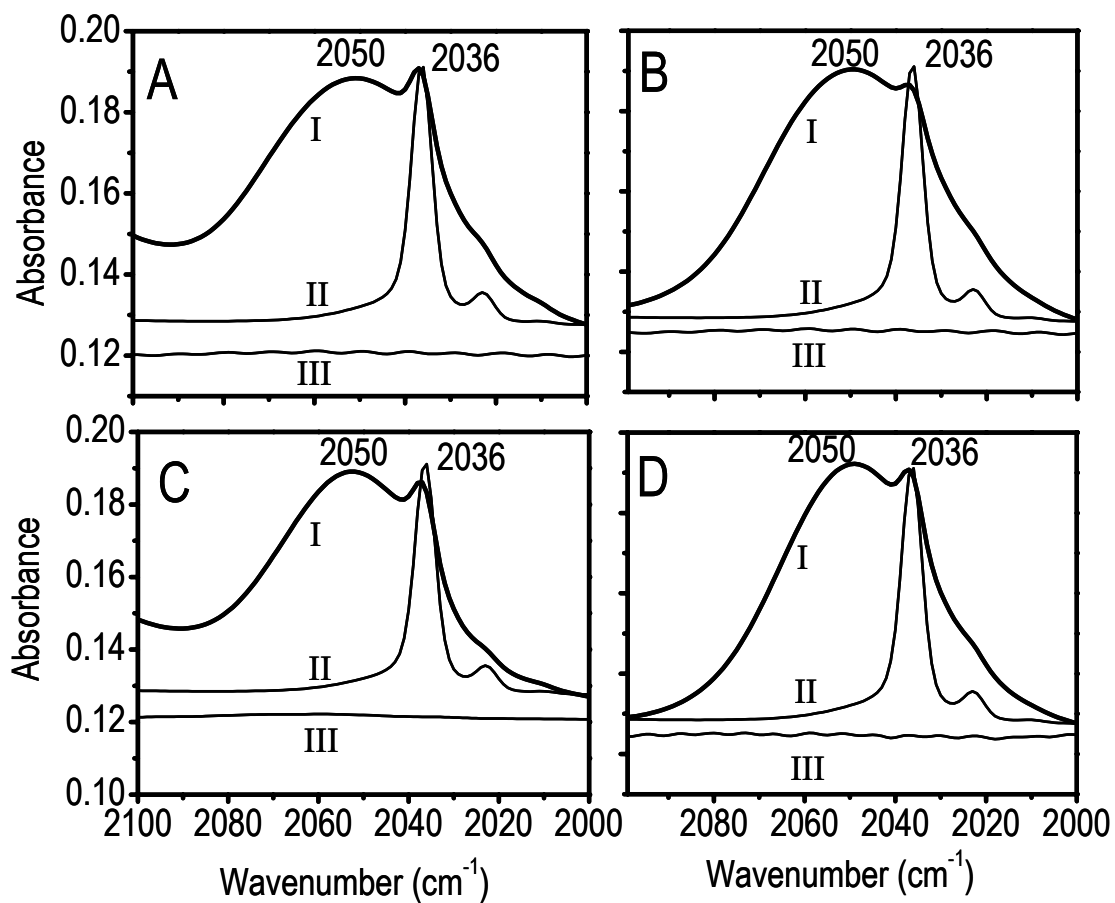


Figure 4. FTIR spectra of Au@Mb-N₃ (A), Ag@Mb-N₃ (B), Au@Hb-N₃ (C) and Ag@Hb-N₃ (D) recorded in KBr matrix. Traces (I), (II) and (III) of A, B, C and D represent M@X-N₃, M@N₃ and M@X, respectively.

Supporting Information 6

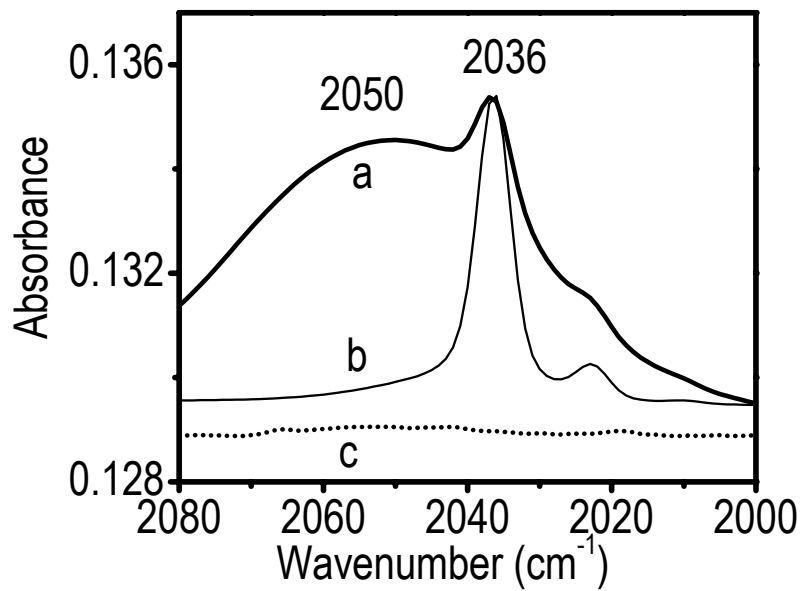


Figure 5. FTIR spectra recorded in KBr matrix. Traces (a), (b) and (c) represent AuNR@Mb-N₃, AuNR@N₃ and AuNR@Mb, respectively.