



Charged Droplets Hot Paper

International Edition: DOI: 10.1002/anie.201604378

German Edition: DOI: 10.1002/ange.201604378

Solid-to-Liquid Charge Transfer for Generating Droplets with Tunable Charge

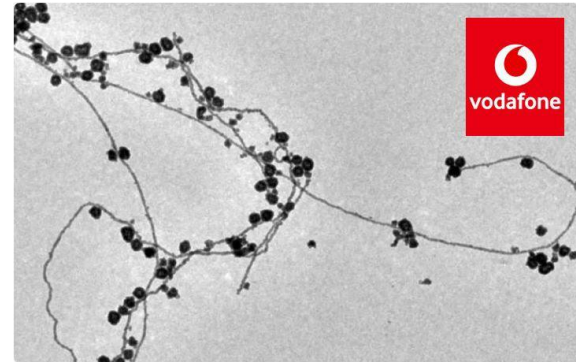
*Yajuan Sun, Xu Huang, and Siowling Soh**

Department of Chemical and Biomolecular Engineering
National University of Singapore
4 Engineering Drive 4, Singapore 117585 (Singapore)

Vishal
05/05/2018

Relevance to lab

1. Electrospray Ionization
2. Drinking water bottles
3. Biology
4. Microfluidic reactors
5. **Scrubbers for removing particulates in an air stream**



Scientists accidentally create gold nanoparticles in water

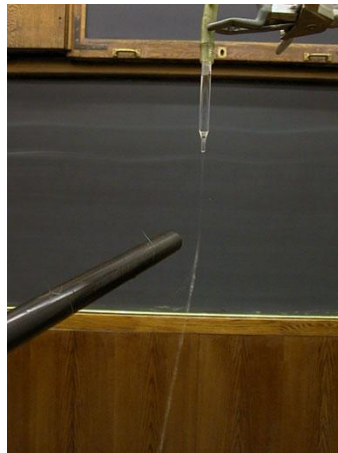
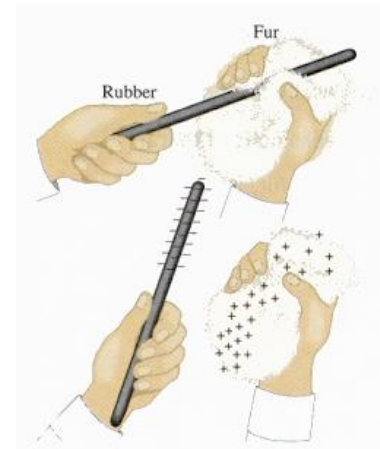
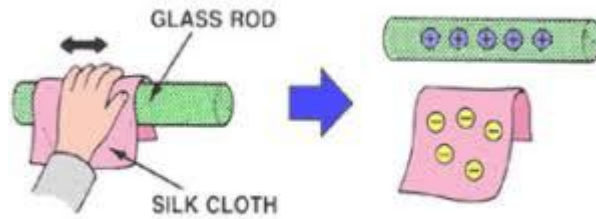
Stanford scientists have accidentally discovered a new way of creating gold nanoparticles and nanowires using water droplets. Gold particles are usually created using gold-containing chloroauric acid and sodium borohydride, a reducing agent. However, when the scientists replaced the reducing agent with water microdroplets, they were surprised to see the reaction still yielded gold nanoparticles.

short by Gaurav Shroff / 20 Apr, 2018



Introduction

1. Contact electrification involves bringing two materials in contact and separating to generate static charge.



A jet of water with charged plastic rod

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1. Solid surfaces are triboelectrically charged.
2. charge transferred from a solid surface charged by contact electrification to water droplet by rolling it across the statically solid surface.
3. Tuning the static charge on solid surface to vary droplet charge.
4. Finally, using the charged droplets for application like manipulating, coalescing, and sorting the water droplets by solid surfaces charged by contact electrification.

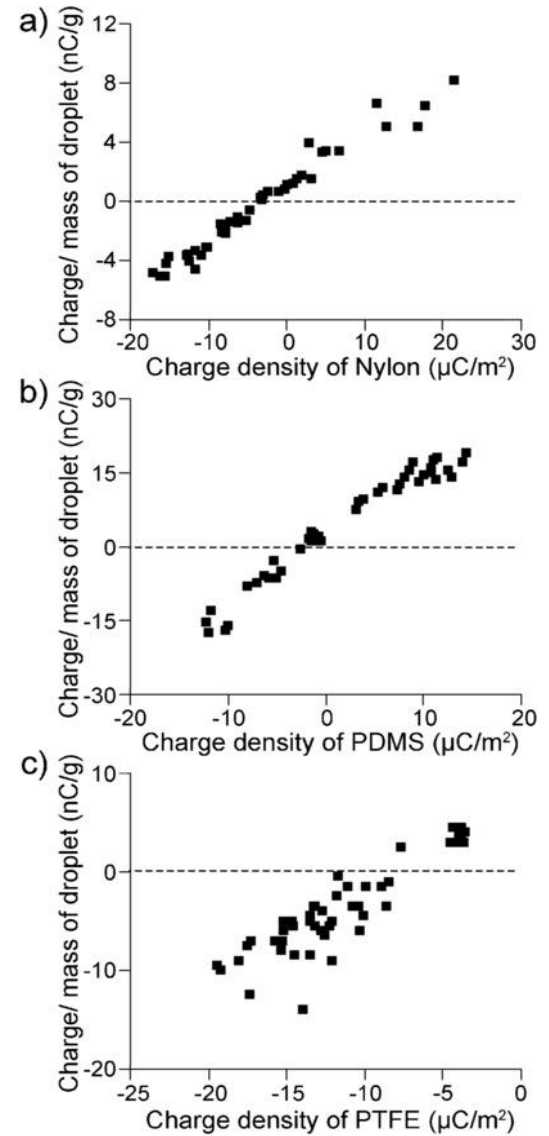
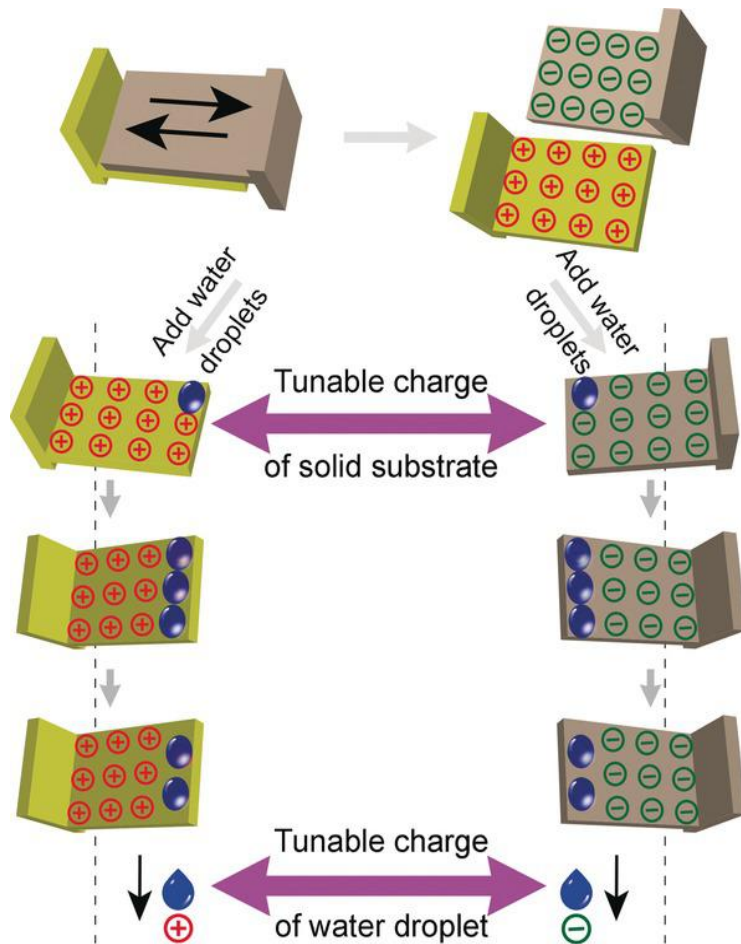
Experimental procedure

1. Nylon(2×2×0.3(cm)) rubbed against PTFE to make it positive charged.
2. Charge on the nylon was measured using a Faraday cup connected to an electrometer.
3. A droplet of de-ionized water (0.1 ml) was slide down the surface under gravity.
4. The droplet was then dropped into the Faraday Cup for measuring its charge.
5. Similarly, Nylon charged negatively by rubbing it against a piece of PEGDA charges droplet negatively.
6. Pressure between 2kPa to 20kPa was applied to vary the surface charge.

PTFE Poly tetra fluoro ethylene

PEGDA Poly ethylene glycol diacrylate

Results



Analyzing the Charged Water

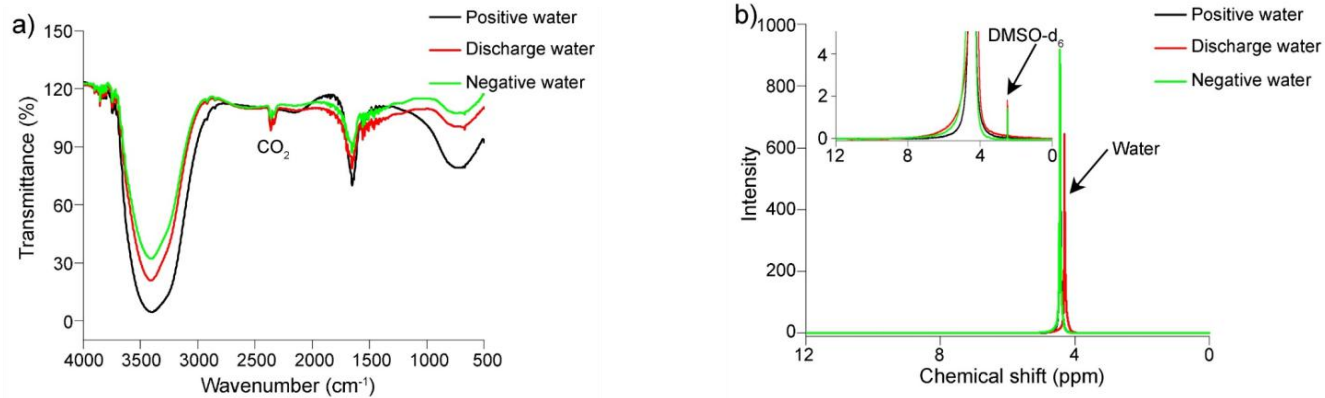
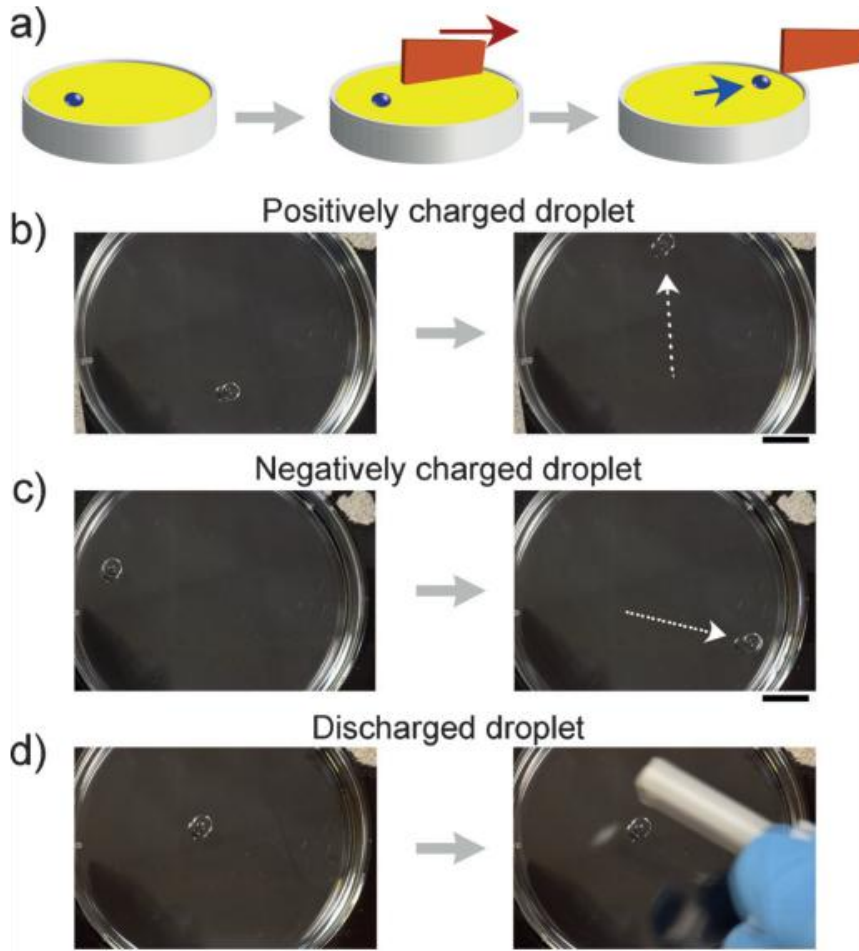


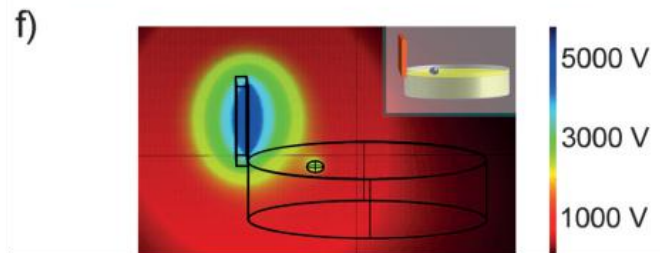
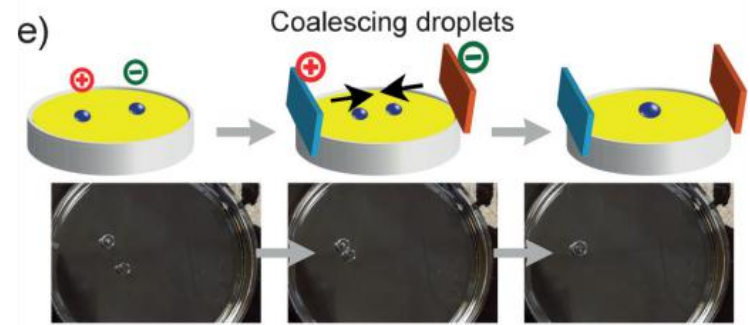
Figure S2. FTIR and ¹H NMR spectra of different types of water. (a) FTIR (b) ¹H NMR in DMSO-d₆. The positively and negatively charged water was charged by flowing across a charged nylon surface.

➔ Results show that the three types of water look similar; hence, at least at the limit of detection of these Instrument, the charged water looks pure.

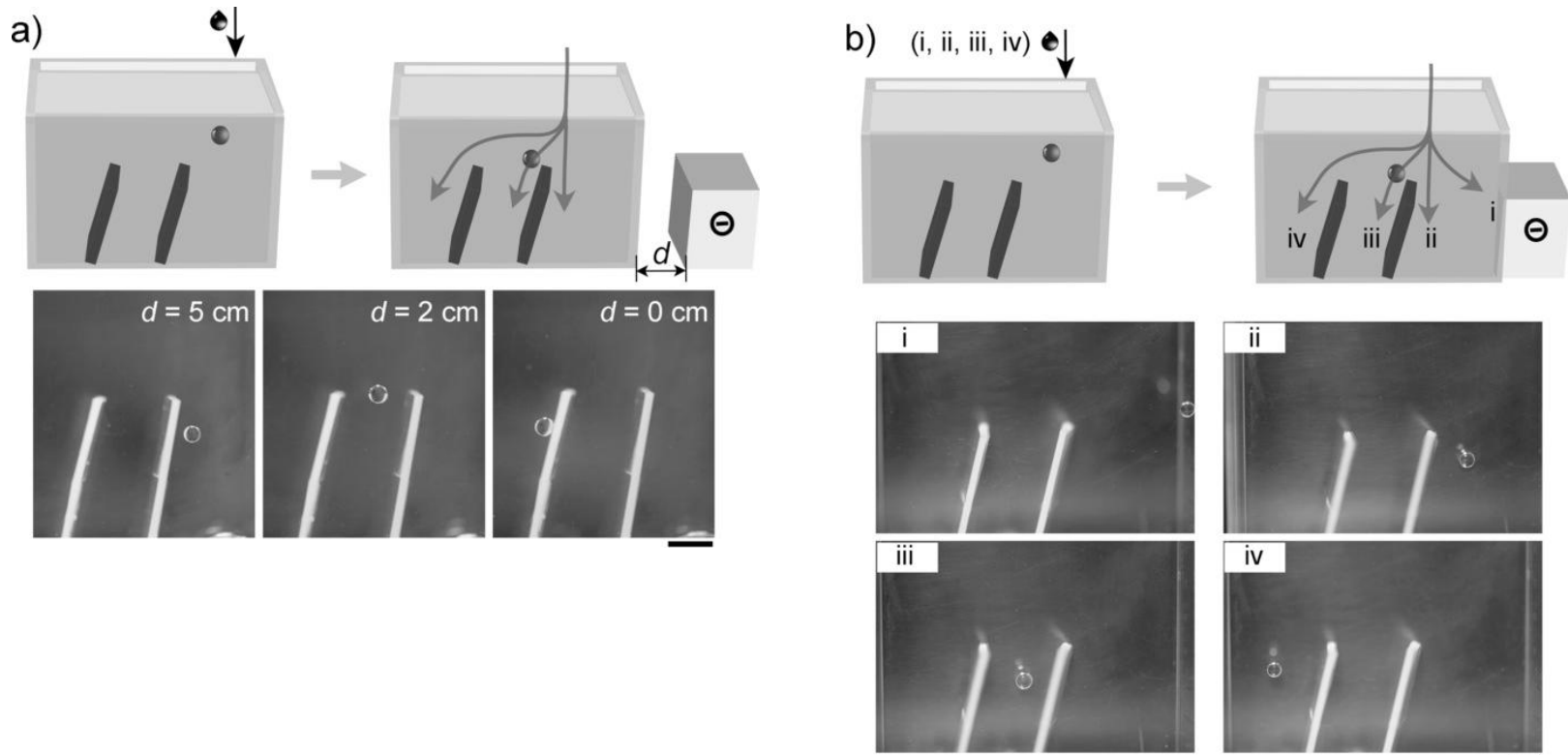
Application: Manipulation



1. A water droplet (0.03 ml) positively by flowing it across an initially uncharged perfluoroalkoxy (PFA) tube.
2. The droplet was placed on the air-liquid interface of a Petri dish filled with silicone oil, and
3. Manipulated with negatively charged PTFE and positively charged nylon rod.

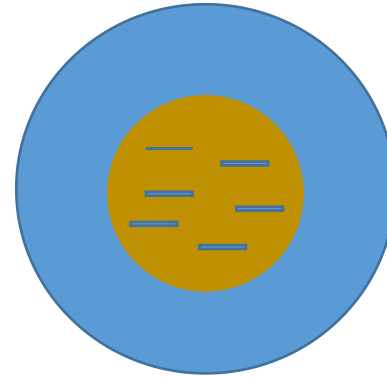
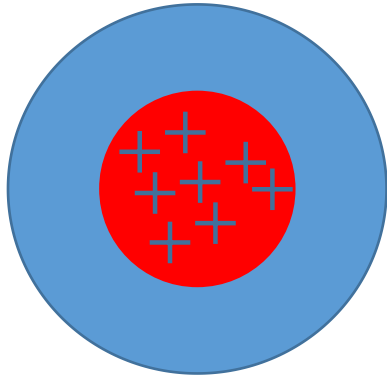
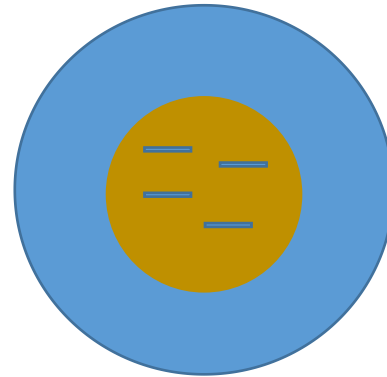
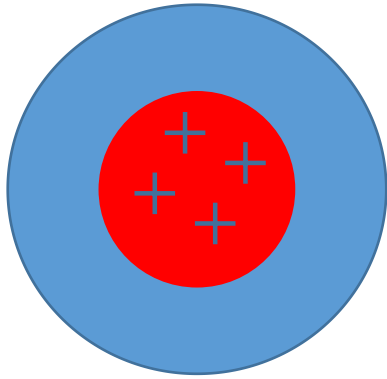


Application: Sorting



Charged water droplet was dropped into a container filled with paraffin oil.

Future plan



Thank you