



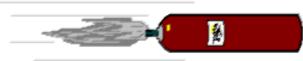
Cylinders: Safety precautions



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Main Causes of Accidents

- > Inadequate training and supervision
- Poor installation
- Poor maintenance
- Faulty equipment and/or design(e.g badly fitting valves or regulators)
- Poor handling
- Poor storage
- >Inadequately ventilated working conditions



A **gas cylinder** is a pressure a vessel for storage and containment of gases at above atmospheric pressure. Inside the cylinder the stored contents may be in a state of compressed gas, vapor over liquid, supercritical fluid, or dissolved in a substrate material, depending on the physical characteristics of the contents.

> Compressed Gas Any material or mixture with an incontainer pressure exceeding 40 psia at 70°F, or a pressure exceeding 104 psia at 130°F, or any liquid flammable material having a vapor pressure exceeding 40 psia at 100°F



Types of Gases

There are three types of gases commonly supplied and used:

- 1. Compressed Gases Nitrogen, Oxygen, Air, Carbon Dioxide, Helium
- 2. Liquefied Gases LPG, Liquefied Nitrous Oxide
- 3. Dissolved Gases Acetylene

Types of Gas Cylinders

In general, there are three types of gas cylinders:

- High Pressure Cylinders High pressure cylinders come in a variety of sizes.
 Ex. Nitrogen, Helium, Hydrogen, Oxygen and Carbon Dioxide.
- 2. Low Pressure Cylinders Low pressure cylinders come in a variety of sizes.
 - **Ex.** LPG and refrigerant gases.

3. Acetylene Cylinders – Acetylene is in a class of its own as the cylinder is filled with an aggregate material and dissolved in a liquid medium (acetone).

- Compressed gases used in a variety of industrial and laboratory situations
- Compressed gases present a unique hazard. Depending on the particular gas, there is a potential for simultaneous exposure to both mechanical and chemical hazards

Gases can be:

- Flammable or combustible
- Explosive
- Corrosive
- Poisonous/toxic
- Inert
- Cryogenic
- Pyrophoric(burns on contact with air)
- or a combination of hazards

- Oxidants support combustion e.g. air & oxygen
- Inerts do not generally react with other materials, asphyxiants (leak displace air) e.g. nitrogen, argon, helium
- Flammables when mixed with oxidant and ignition source will burn e.g. acetylene, hydrogen, propane
- Toxics toxic in small concentrations e.g. ammonia, chlorine, carbon dioxide
- Corrosives react with materials causing reactions e.g.
 chlorine, sulfur dioxide
- Pyrophorics ignite spontaneously in air e.g. silane, phosphine

Classes of Gases

Dangerous Goods Diamond	Class	Examples
FLAMMABLE GAS 2	Class 2.1 Flammable gas	LPG, hydrogen, acetylene
KOHTLAMMABLE NON-TOXIC ELS 2	Class 2.2 Non-flammable, non-toxic gases	Compressed air, nitrogen, argon, carbon dioxide, helium.
OXIDIZING GAS 2	Class 2.2, Sub-risk 5.1 Oxidizing gas	Oxygen, nitrous oxide, Entonox (50% oxygen, 50% nitrous oxide).
Toxic GAS 2	Class 2.3 Toxic Gas	Methyl bromide, anhydrous ammonia, chlorine.
NO	N-FLAMMABLE GAS 2	PAMMABLE GAS



Cylinder shoulder colours				
By hazard property				
Flammable	Red			
Toxic/corrosive	Yellow			
Inert	Bright green			
Oxidising	Pale blue			
Note: More than one hazard property may be shown on the cylinder shoulder e.g. red and yellow				

By specific gas

Argon	Dark green	
Carbon dioxide	Dusty grey	
Helium	Brown	
Nitrogen	Black	
Nitrous oxide	Dark blue	
Oxygen	White	

Red	=
Yellow	=
Light blue	=
Bright green	=
Red and yellow	=
Yellow and light blue	=

- = Flammable
- = Toxic or corrosive
- = Oxidising
- = Inert
- = Flammable and toxic
- = Toxic and oxidising



Flammable





Toxic or corrosive

Oxidising





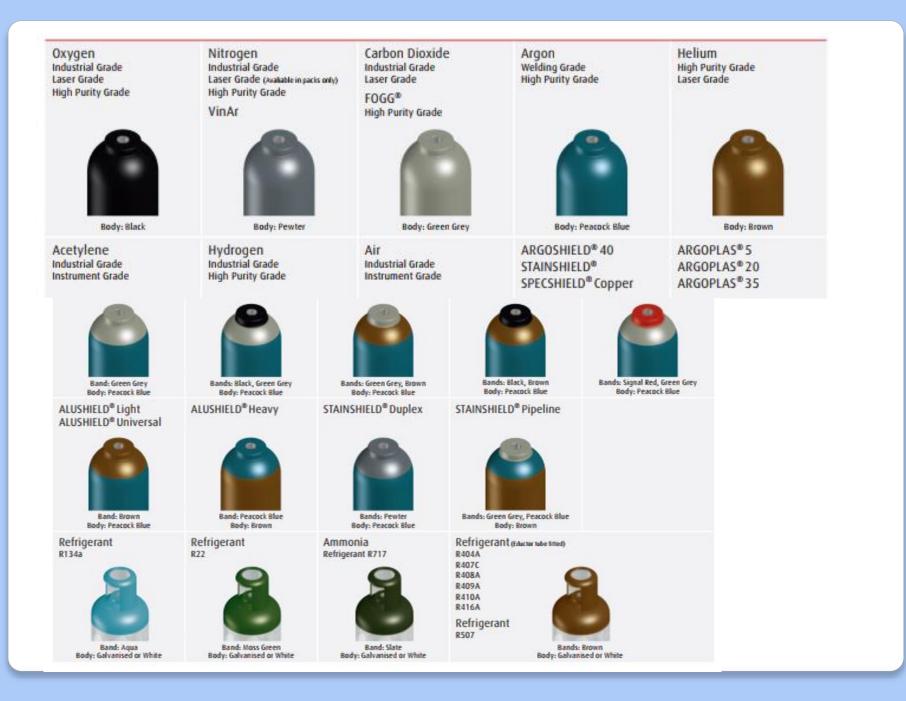


Inert

Flammable and toxic Toxic and oxidising







Flammable vapour air mixtures

A leakage of a flammable gas would form a flammable vapour air mixture.

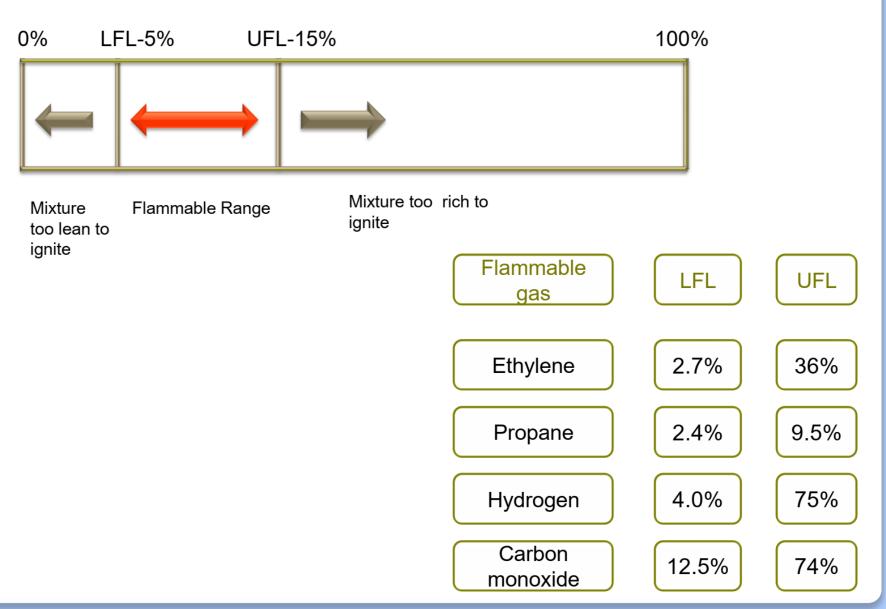
If the vapour air mixture is within the flammable range, it can explode in the presence of an ignition source.

Vapour air mixtures will only ignite within a well-specified range of composition.

The lower flammable limit (LFL) is the leanest mixture that can ignite, i.e., the mixture with the smallest fraction of combustible gas.

The upper flammable limit (UFL) is the richest fraction of combustible gas mixture that can ignite.

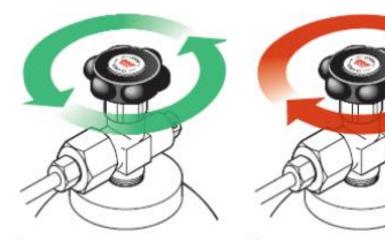
Flammable limits for methane gas.



Safe Working with Gas Cylinders Daily Inspection

 Cylinders should be inspected daily and prior to each use for corrosion, leaks, cracks, etc.
 Inspection should include the cylinder, piping, safety relief devices, valves, protection caps and stems.

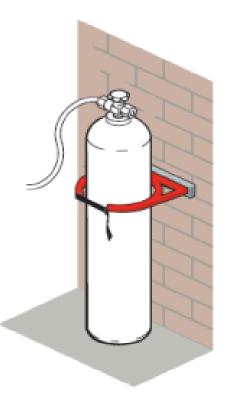
Leaking regulators, cylinder valves or other equipment should be taken out of service.



Close







Do not expose gas cylinders to temperatures above 45° C. Overheating of cylinders can result in build up of pressure and explosion.

Do not store gas cylinders with other combustible materials.

Flammable substances, such as oil and other solvents, must not be stored in the same area. The floor used for storing gas cylinders must be dry, to prevent the corrosion of gas cylinders.

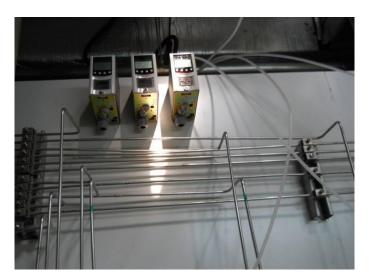
Never use oxygen regulator for flammable gases. An adaptor must not be used for connecting regulators. **Cross contamination of internal parts can result in rapid oxidation and fire.** Same is applicable for other oxidising gases.

The empty and full cylinders must be stored separately and clearly labelled/marked.

Do not use cylinders without a regulator.

Close the valve after usage or when the equipment is not in use.

Install gas detectors for detecting leaks.



Place the spindle key near the gas cylinder, so that the valve can be quickly closed in case of an emergency.

Cylinders must always be kept chained or supported in a manner to prevent fall.

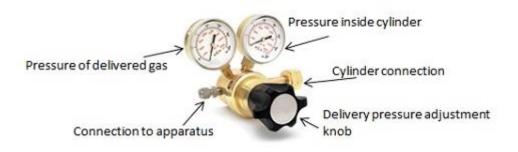
Refer the Material Safety Data Sheet **(MSDS)** for the gas before usage, to know about the hazards and precautions to be taken.

A cylinder must never be emptied to a pressure lower than 25 psi as the residual contents may become contaminated if the valve is left open.

Close cylinder valve whenever: work is finished cylinder is empty



A regulator is a device that receives gas at a high pressure and reduces it to a much lower working pressure



Acetylene gas under excess pressure becomes unstable

After the regulator is attached, the cylinder valve should be opened just enough to indicate pressure on the regulator gauge (no more than one full turn) and all the connections checked with a soap solution for leaks.

Never use oil or grease on the regulator of a cylinder valve.

Before a regulator is removed from a cylinder, the cylinder valve shall

be closed and the regulator relieved of gas pressure.

Regulators, gauges, hoses and other apparatus shall not be used on gas cylinders having different chemical properties

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First aid measures

In case of contact of corrosive/toxic gas with eyes-flush with water for at least 15 min.

In case of skin contact, remove contaminated clothing and flush the affected part with water. An emergency shower can be used for the same.

In case of inhalation of gas, remove the person to fresh air. Get immediate medical attention. Do not give anything to drink if the person is not conscious.

Primary Governing Bodies / Safety Codes

Compressed Gas Association (CGA) Semiconductor Equipment & Materials International (SEMI) US Environmental Protection Agency (EPA) Uniform Fire Code / Local City Regulatory Committee (UFC) Uniform Building Code (UBC) / BOCA National Fire Prevention Code (NFPC) International Conference of Building Officials (IBOC) Toxic Gas Ordinance (TGO)

Always Treat Gas Cylinders with Respect

WE RISE AND FALL AS ONE PEOPLE, ONE NATION.

- BARACK OBAMA

THANK YOU